

Atomic and Molecular Physics with XFELs I

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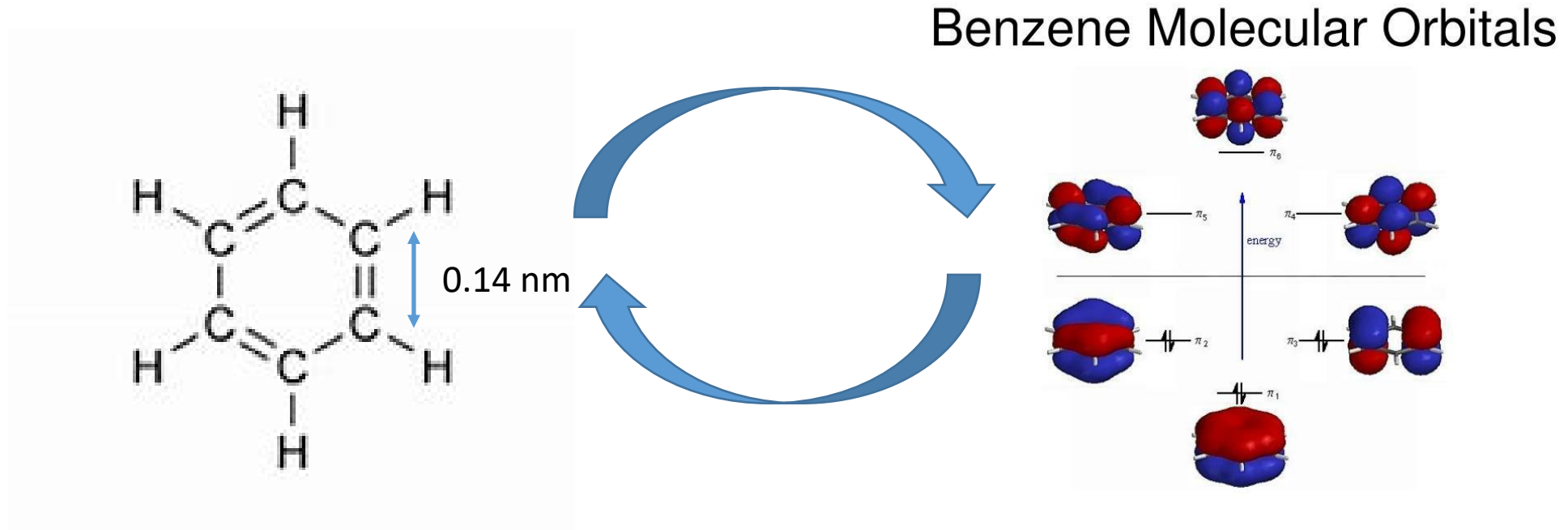
Contents

1. Why do atomic and molecular physics with XFELs?
2. Time-resolved pump-probe methods with XFELs
3. Ultrafast electronic and electron-nuclear coupling dynamics
4. Attosecond XFEL science

1. Why do atomic and molecular physics with XFELs

- Atoms and molecules are the fundamental building blocks of matter
- X-rays allow structural and electronic state resolved measurements and permit excited state dynamics to be followed
- Gas phase systems (i.e. atoms, molecules & clusters) can open to study channels obscured in condensed phase and allow individual molecules to be studied (e.g. through coincidence methods)
- Fundamentals of electronic dynamics (e.g. charge migration, Auger decay, ICD, exciton dynamics) and electron-nuclear coupling (e.g. charge transfer, vibronic coupling, quantum dynamics at conical intersections, coupling to environment) can be studied in these systems with full experimental resolution and tractable theory
- Clusters & droplets give systematic access to the development of intramolecular interactions (e.g. solvation effects etc.)

1. In molecules and condensed matter the spacings between atomic nuclei are $\sim 0.1-0.2$ nm

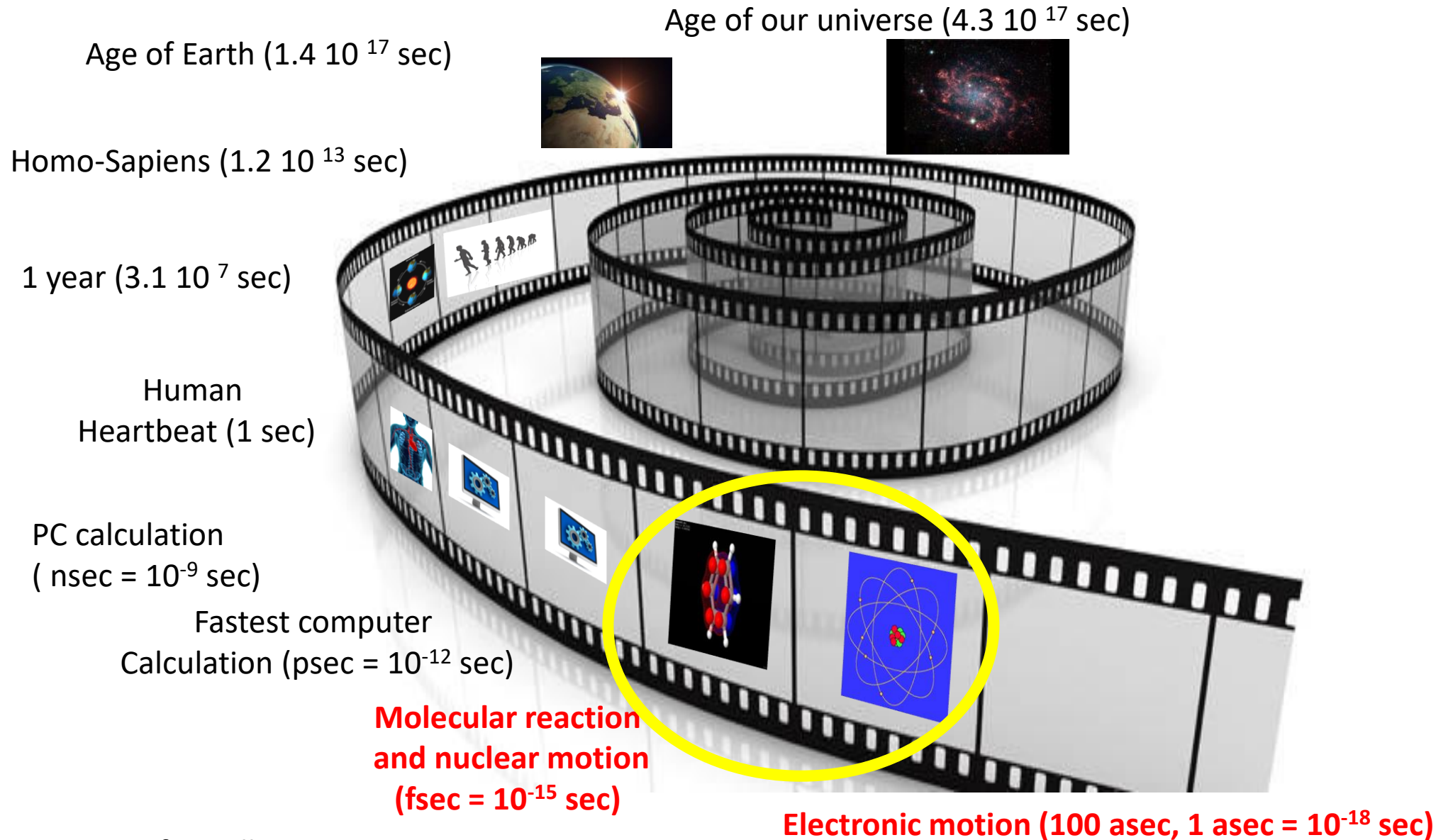


Arrangement of atomic nuclei in a benzene molecule

Electrons arranged in this structure according to the rules of quantum mechanics which ultimately determines the arrangement of atomic nuclei

In nature these structures are NOT static - there is constant internal motion and the possibility of fast rearrangement of atomic nuclei and electron e.g. in a chemical reaction, either thermally driven or triggered by photon or electronic interactions

1. Time scales

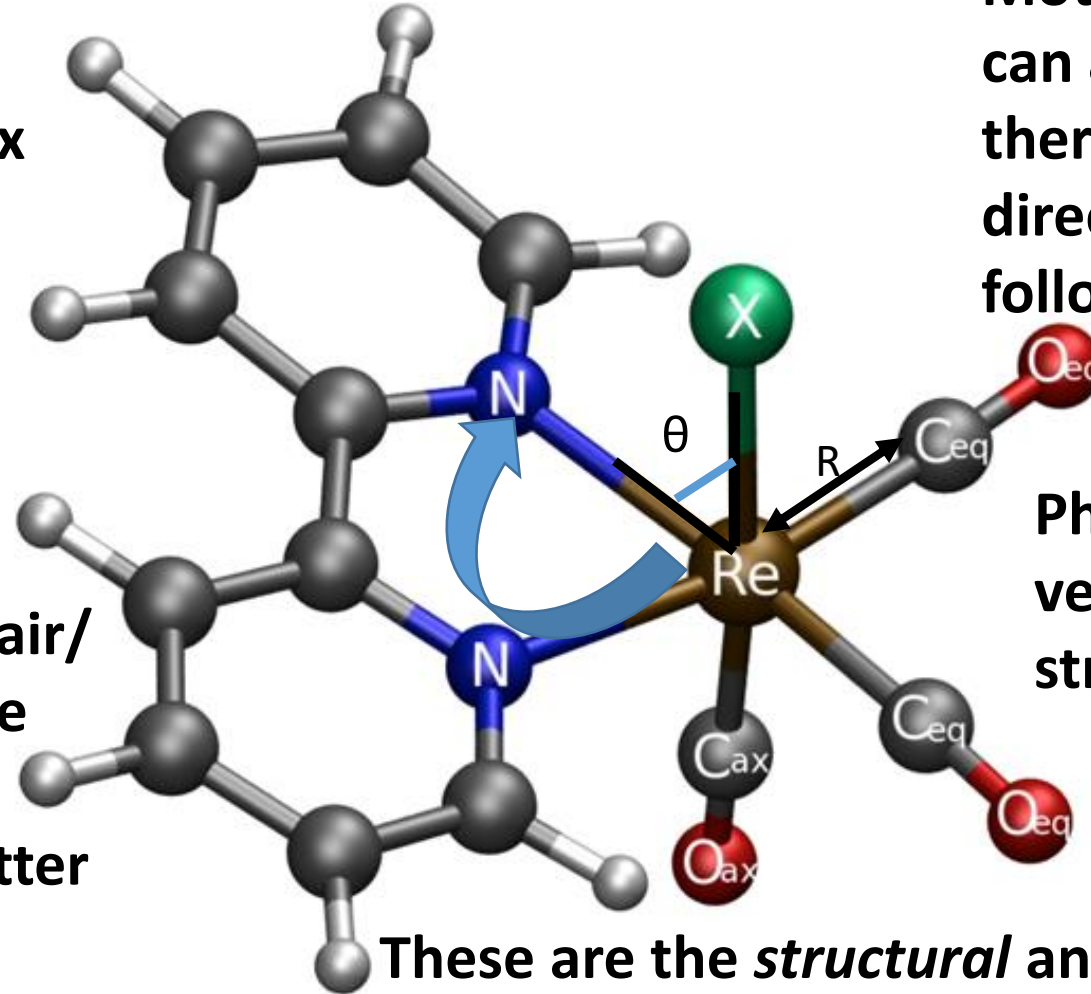


1. Structure and dynamics in matter

Consider a material system
e.g. this metal-ligand complex
- a largish molecule

The atoms are arranged in
a specific geometry
with a bond angle and length
that can be defined for each pair/
group of atoms in the structure

This is the *structure* of the matter



Motion of the structure
can also occur: through
thermal fluctuations or
directed changes e.g.
following photo-excitation

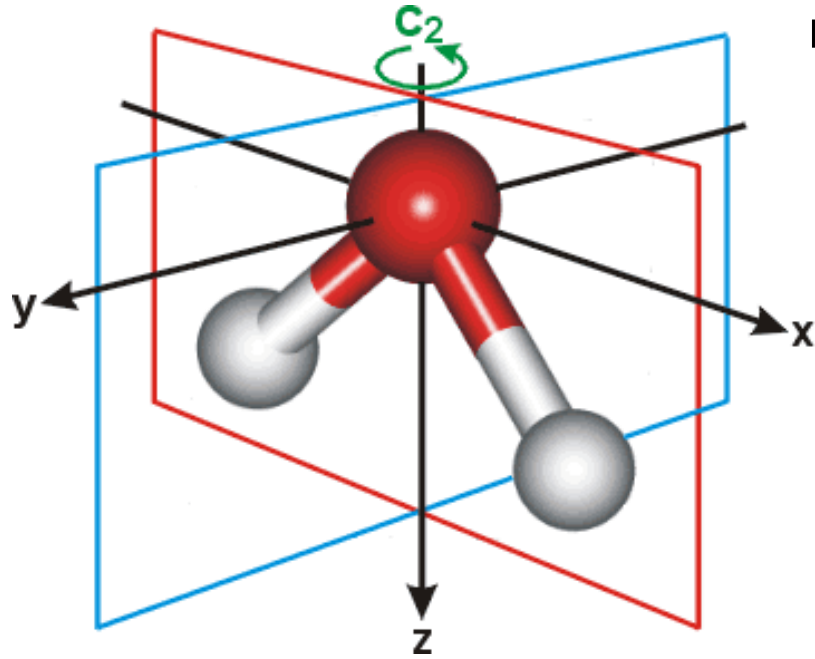
Photo-excitation can drive
very rapid electron and
structural motion

These are the *structural* and *electronic dynamics*
of the system that underpin chemical and physical
changes

1. Rotational motion in a molecule

$$E_R = \frac{L^2}{2I}$$

For a rigid rotor rotational energy E_R depends upon the rotational constant B_0 and the angular momentum J of the rotational state:



$$E_R = B_0 J(J + 1)$$

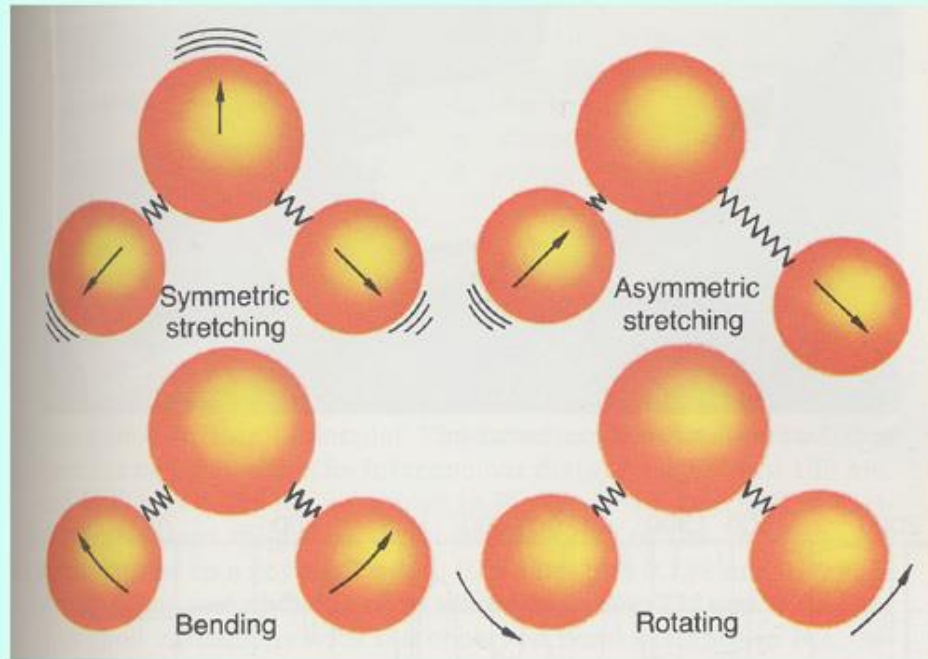
$$B_0 = \frac{\hbar^2}{2\mu R_0^2}$$

$$T_{rot} = \frac{h}{2B_0}$$

The rotational period varies from ~100fs for a light molecule to many picoseconds for large molecules.

1. Vibrational and nuclear motion in a molecule

Modes of Vibration of Molecules

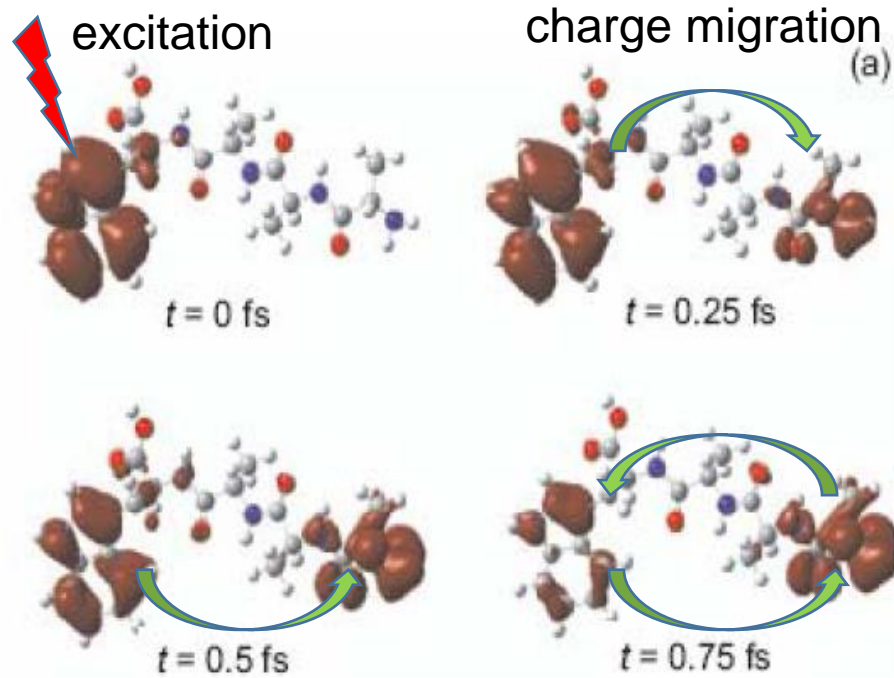


$$E_n = \left(n + \frac{1}{2}\right)\hbar\omega$$

The vibrational states and energies are approximately those of a quantum harmonic oscillator...

Each mode oscillates (vibrates) with a characteristic frequency, corresponding to vibrational periods of 8 fs (H_2) to picoseconds for very weakly bound molecules.

1. Electronic motion in a molecule



Calculated electron dynamics in a dipeptide



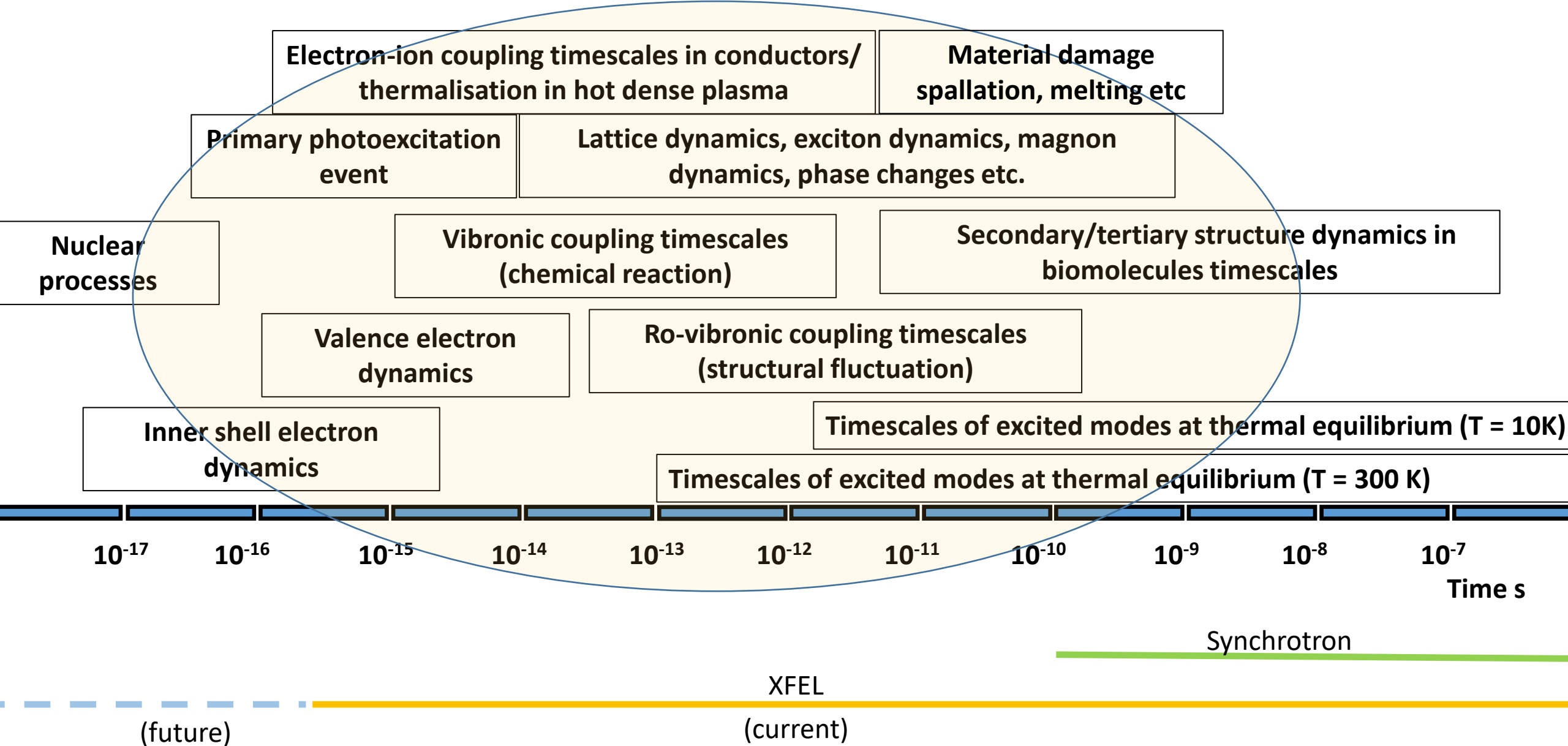
Electron Orbit in Bohr Model

$T_{\text{orbit}} \approx 150 \text{ as}$ for H ground state

Motion on timescales of 0.01 fs- 10 fs

In a chemical reaction or physical change all three types of motion: electronic, vibrational & rotational will occur in a highly correlated fashion

1. Science driver for ultrafast X-rays is real-time access to the processes and fluctuations in matter down to the quantum scale



1. Why are the ultrafast dynamics of electronically excited systems so important to us ?



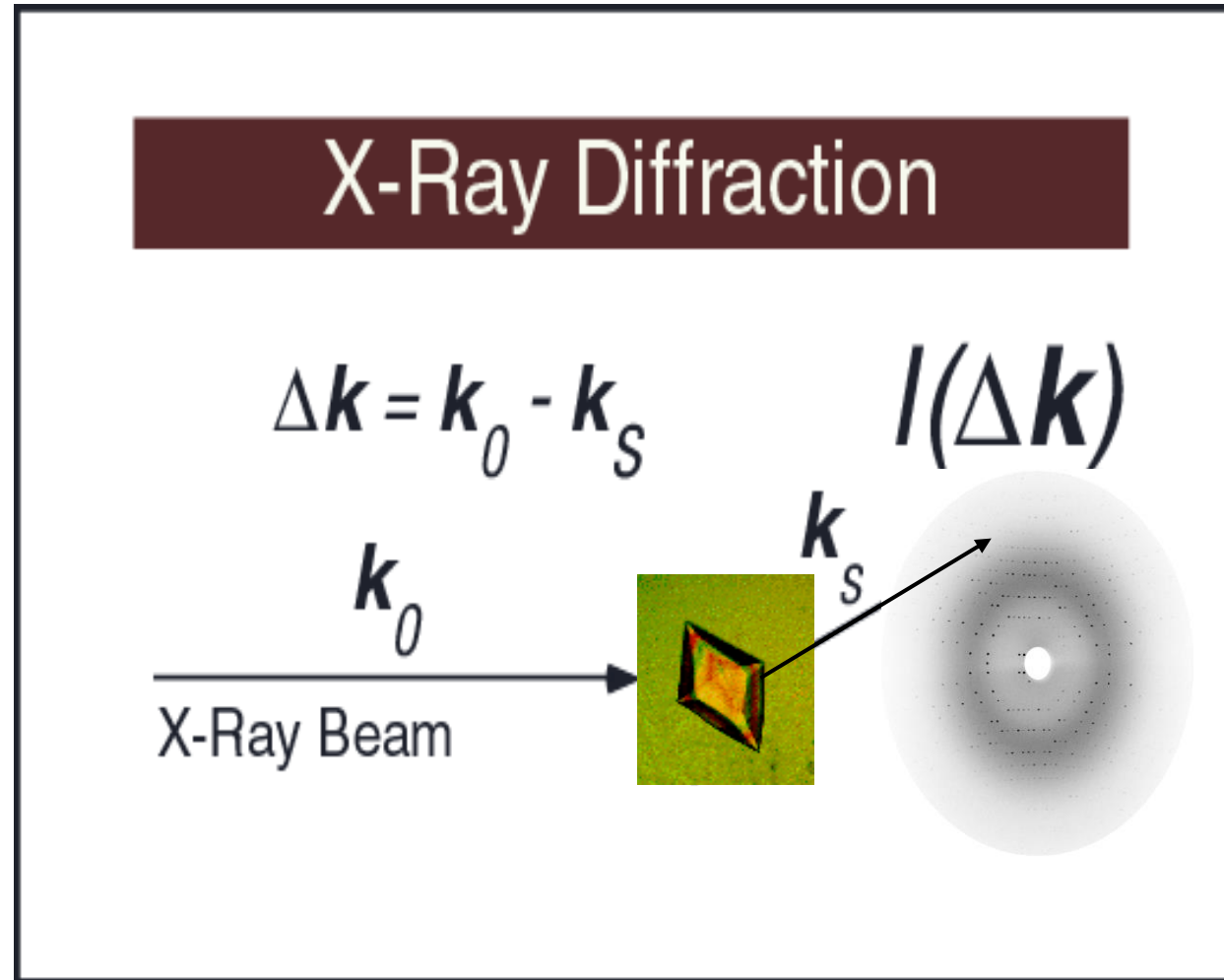
Life-on earth, and increasingly our sustainable technologies, are powered by solar driven electronic excitation.

This provides a timely impetus for the science, along with the needs of chemical, materials, communications, detector and computing technologies

Ultrafast X-rays give us access to the $< 10^{-10}$ m *spatial* and $< 10^{-15}$ s *temporal* scales needed to fully track and understand the electronic excitation driven dynamics in matter.

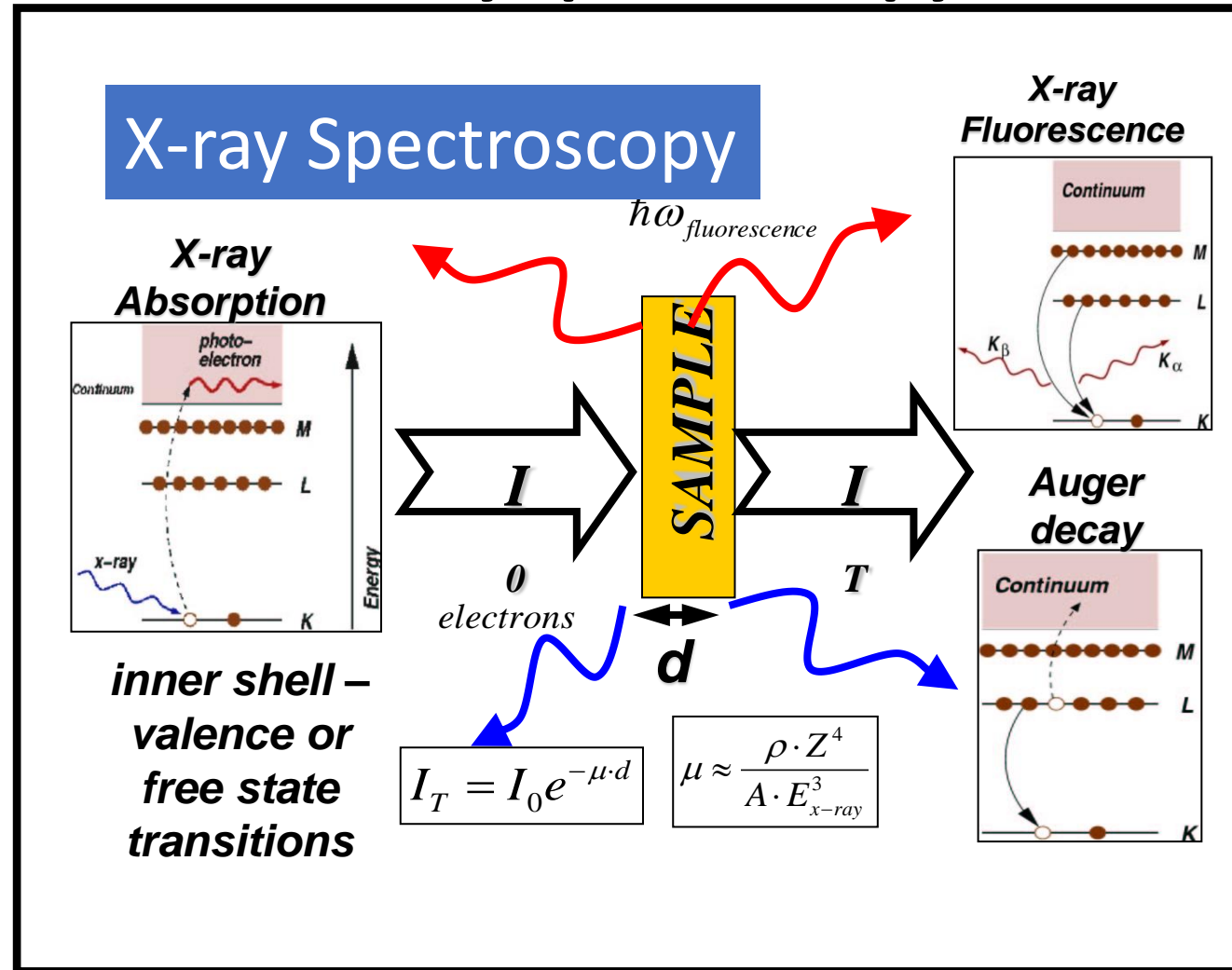
They give access to fundamental decoherence times in molecules and materials that will advance our understanding in key areas of science

2. X-ray probing of the nanoscale *structure* of matter using scattering



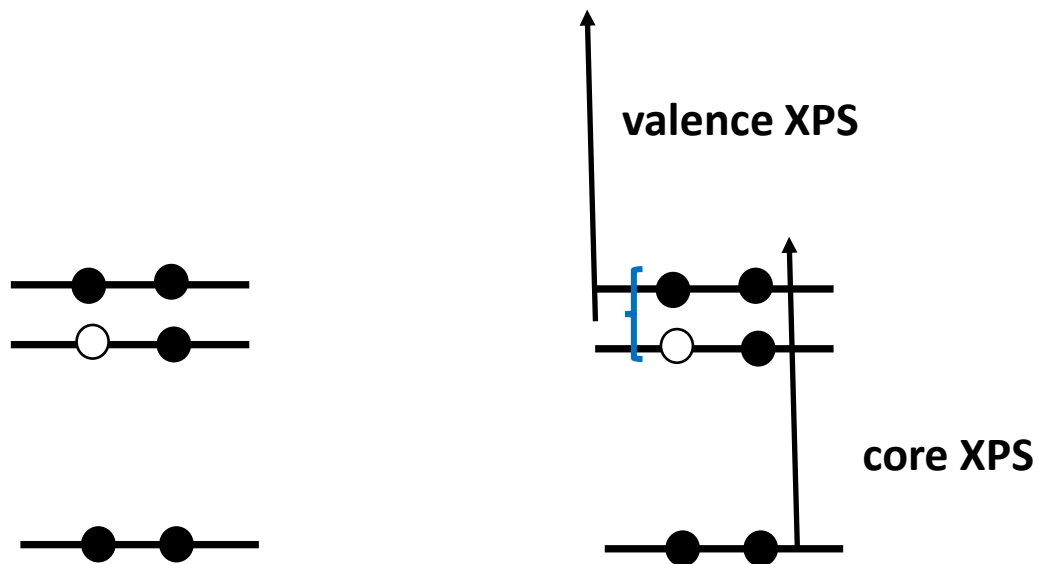
Structure from Thomson/Compton scattering of x-rays by electrons located at atomic sites. In crystal coherent amplification for scattering satisfying the Bragg condition, even non-crystalline systems will yield a characteristic scattering pattern that permits structural retrieval

2. X-ray probing of the nanoscale *structure* of matter using X-ray spectroscopy



Characteristic core/ core-valence electronic transitions provide chemical/site specific information of electronic structure and occupations

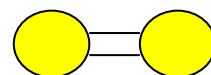
2. X-ray probing of the dynamics of matter using charged particle fragments: photoelectron, Auger electrons & ions



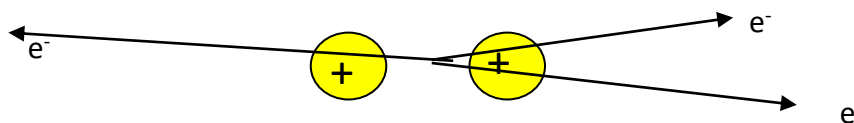
Characteristic core/ core-valence photo-electronic spectrum provides site specific information of electronic state. Additional information from electron angular distribution and Auger electrons

Coulomb explosion provides information on instantaneous structure.

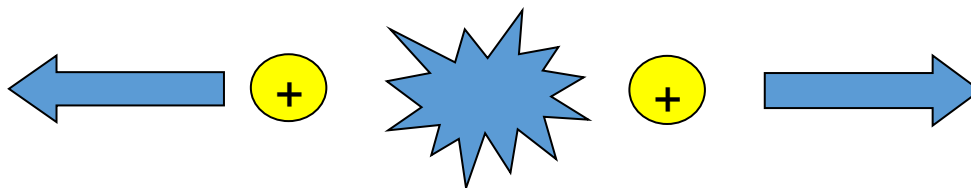
Electron- ion positions in coincidence permit full reconstruction of event



Bonding and other electrons stripped by x-ray ionisation, leading to multi-electron dissociative ionisation (MEDI)



Coulomb potential energy released as kinetic energy



Direction of release gives alignment state of molecule (at least for simple molecules)

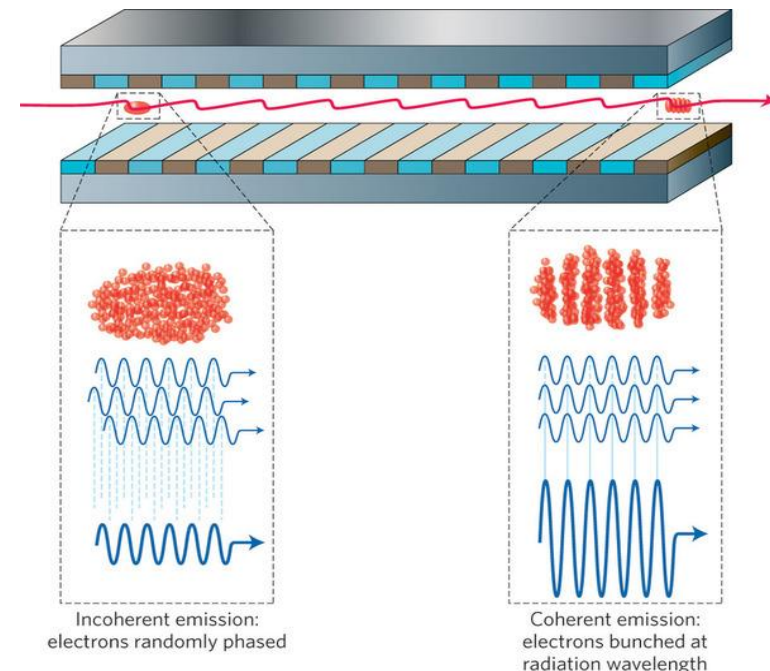
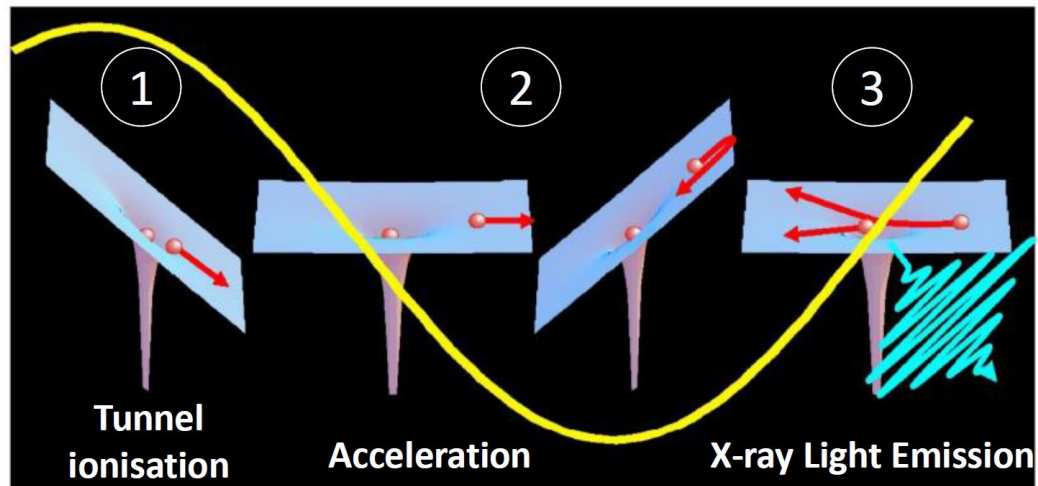
2. X-ray photon sources for probing femtosecond to attosecond dynamics using time-resolved spectroscopy

High harmonic generation - HHG

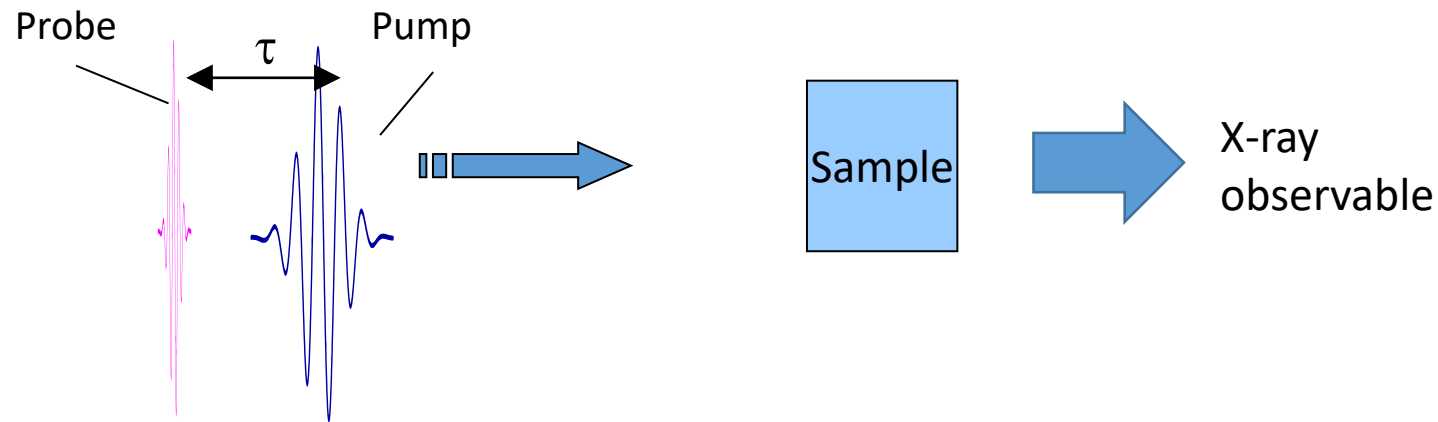
- Isolated attosecond pulse (IAP) duration < 200 as
- Photon energy to 100 eV with 800 nm drives and to > 500 eV with longer wavelength
- > 10^8 (25 eV) - < 10^3 (500 eV) photons/pulse
- Excellent coherence and synchronisation

X-ray free electron laser - XFEL

- Pulses few fs to < 0.5 fs
- Photon energy to > 10 keV
- > 10^{11} photons/pulse
- Synchronisation to external lasers > 10 fs
- Two colour



2. Time-resolved X-ray measurements by the pump-probe methodology



In a pump-probe measurement the sample is excited/activated by a pump pulse and probed after a controllable delay τ through observation of a suitable observable. As τ is varied the dynamics can be mapped

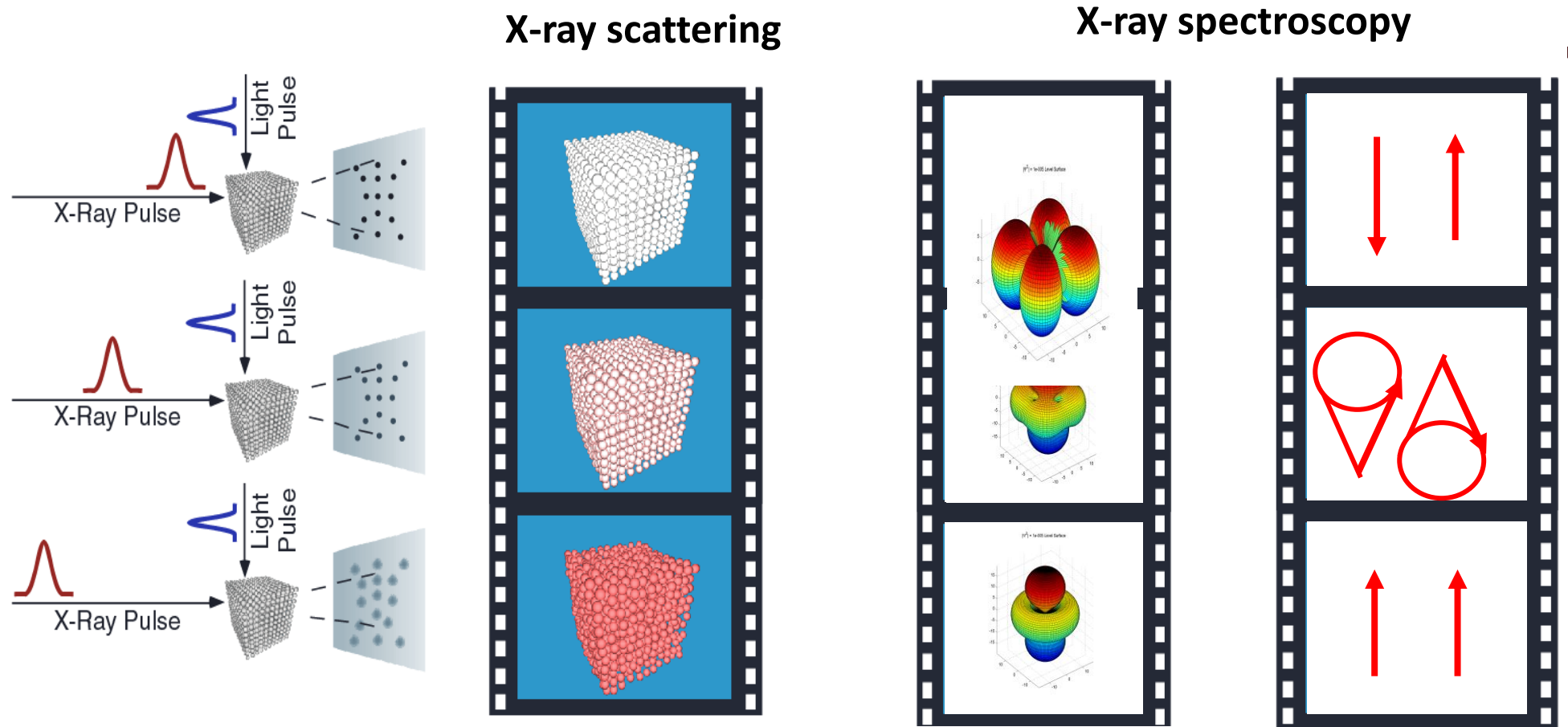
Pump (X-ray to THz) – activates the sample:

e.g. Sudden heating, injection of hot electrons, electronic state photoexcitation/photoionisation, excitation of vibrational/phonon or rotational modes etc.

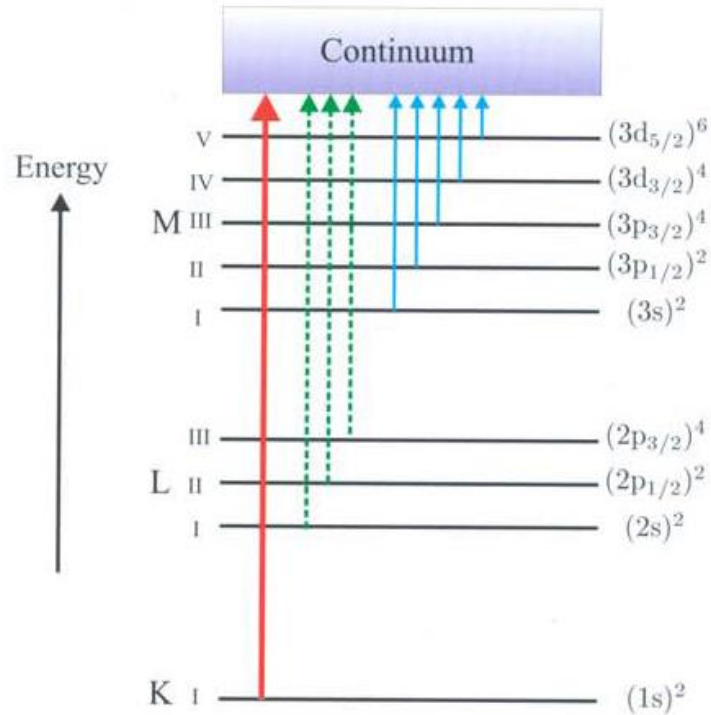
Probe (X-ray) – interrogates the sample after delay time τ by generating a suitable observable:

e.g. X-ray scattering/diffraction, resonant inelastic X-ray scattering (RIXS), X-ray spectroscopy (absorption (XAS) and emission (XES), X-ray photoelectron spectroscopy (XPS) etc.)

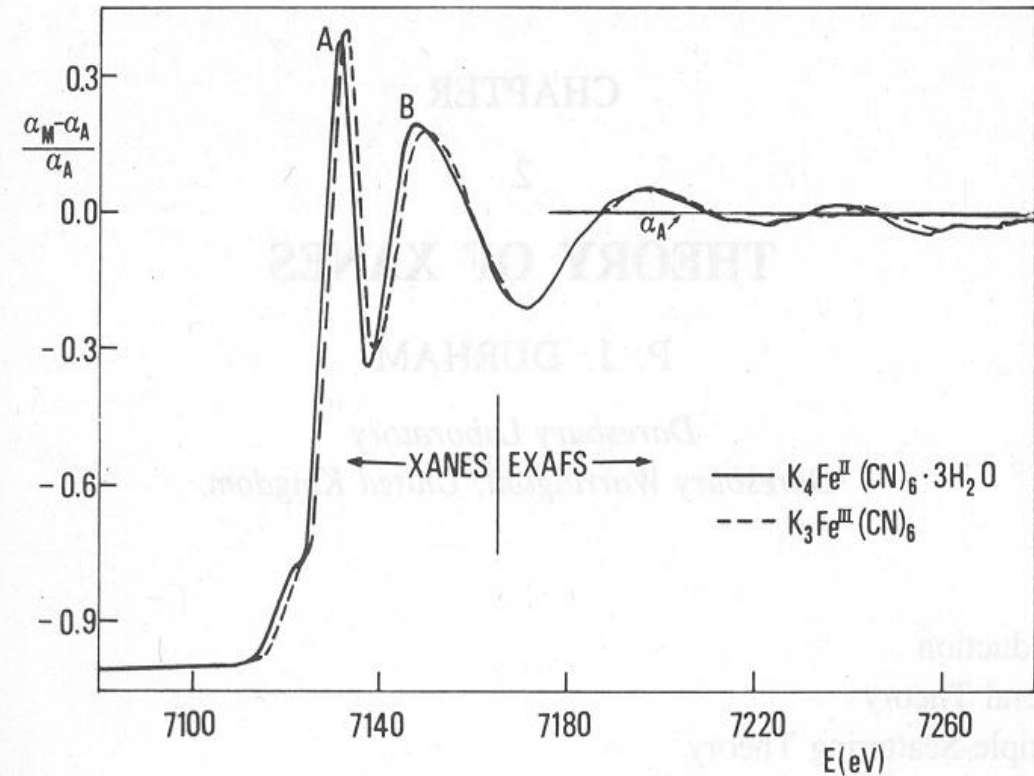
2. Pump – probe measurements with incisive X-ray probes



2. X-ray spectroscopies and allied x-ray photoelectron spectroscopies are examples of incisive X-ray probes



Summary of the nomenclature of the absorption edges of the elements. The electronic shells are labelled $(n/l)^{2j+1}$ where n is the principal quantum number, l is the orbital momentum and j is the total angular momentum. $(2j+1)$ is the multiplicity.



XANES: X-ray Absorption Near Edge Structure

EXAFS: Extended X-ray Absorption Fine Structure

2. Access to x-ray K, L and M edges vital to spectroscopic measurements with element/site specificity

Key:
 atomic number - element name
 element symbol
 K-edge [keV]
 L-edge [keV]

0.1 - 1 keV
 1 - 2 keV
 2 - 7.5 keV

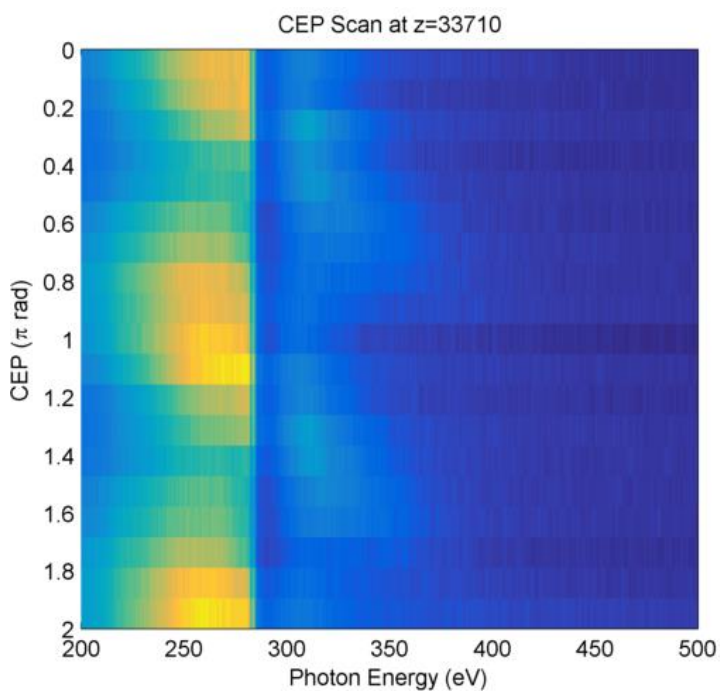
1 - hydrogen H 0.0136 --																	2 - helium He 0.0246 --				
3 - lithium Li 0.0547 --	4 - beryllium Be 0.1115															5 - boron B 0.188 --	6 - carbon C 0.2842 --	7 - nitrogen N 0.4099 L1: 0.0373	8 - oxygen O 0.5431 L1: 0.0416	9 - fluorine F 0.6967 --	10 - neon Ne 0.8702 0.0216
11 - sodium Na 1.0708 0.03081	12 - magnesium Mg 1.3030 0.0495															13 - aluminium Al 1.5596 0.07255	14 - silicon Si 1.839 0.09942	15 - phosphorus P 2.1455 0.135	16 - sulfur S 2.472 0.1625	17 - chlorine Cl 2.8224 0.200	18 - argon Ar 3.2059 0.2484
19 - potassium K 3.608 0.2946	20 - calcium Ca 4.0385 0.3462	21 - scandium Sc 4.492 0.3987	22 - titanium Ti 4.966 0.4538	23 - vanadium V 5.465 0.5121	24 - chromium Cr 5.989 0.5741	25 - manganese Mn 6.539 0.6387	26 - iron Fe 7.112 0.7068	27 - cobalt Co 7.709 0.7781	28 - nickel Ni 8.333 0.8527	29 - copper Cu 8.979 0.9327	30 - zinc Zn 9.659 1.0218	31 - gallium Ga 10.367 1.1164	32 - germanium Ge 11.103 1.2170	33 - arsenic As 11.867 1.3236	34 - selenium Se 12.658 1.4339	35 - bromine Br 13.474 1.550	36 - krypton Kr 14.326 1.6784				
37 - rubidium Rb 15.200 1.804	38 - strontium Sr 16.105 1.940	39 - yttrium Y 17.038 2.080	40 - zirconium Zr 17.998 2.223	41 - niobium Nb 18.986 2.371	42 - molybdenum Mo 20.000 2.520	43 - technetium Tc 21.044 2.677	44 - ruthenium Ru 22.117 2.838	45 - rhodium Rh 23.220 3.004	46 - palladium Pd 24.350 3.173	47 - silver Ag 25.517 3.351	48 - cadmium Cd 26.711 3.538	49 - indium In 27.940 3.730	50 - tin Sn 29.200 3.929	51 - antimony Sb 30.491 4.132	52 - tellurium Te 31.814 4.341	53 - iodine I 33.169 4.557	54 - xenon Xe 34.561 4.786				
55 - caesium Cs 35.985 5.012	56 - barium Ba 37.441 5.247	57-70 *	71 - lutetium Lu 63.314 9.244	72 - hafnium Hf 65.351 9.561	73 - tantalum Ta 67.416 9.881	74 - tungsten W 69.525 10.207	75 - rhenium Re 71.676 10.535	76 - osmium Os 73.871 10.871	77 - iridium Ir 76.111 11.215	78 - platinum Pt 78.395 11.564	79 - gold Au 80.725 11.919	80 - mercury Hg 83.102 12.284	81 - thallium Tl 85.530 12.658	82 - lead Pb 88.005 13.035	83 - bismuth Bi 90.524 13.419	84 - polonium Po 93.105 13.814	85 - astatine At 95.730 14.214	86 - radon Rn 98.404 14.619			
87 - francium Fr 101.137 15.031	88 - radium Ra 103.922 15.444	89-102 **	103 - lawrencium Lr 153.181 24.xxx	104 - rutherfordium Rf [261.11] [261.11]	105 - dubnium Db [262.11] [262.11]	106 - seaborgium Sg [263.12] [263.12]	107 - bohrium Bh [264.12] [264.12]	108 - hassium Hs [265.13] [265.13]	109 - meitnerium Mt [268] [268]												

*lanthanoids

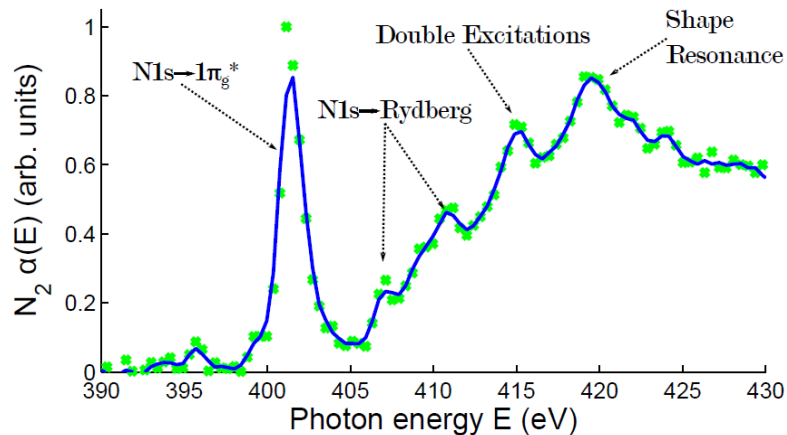
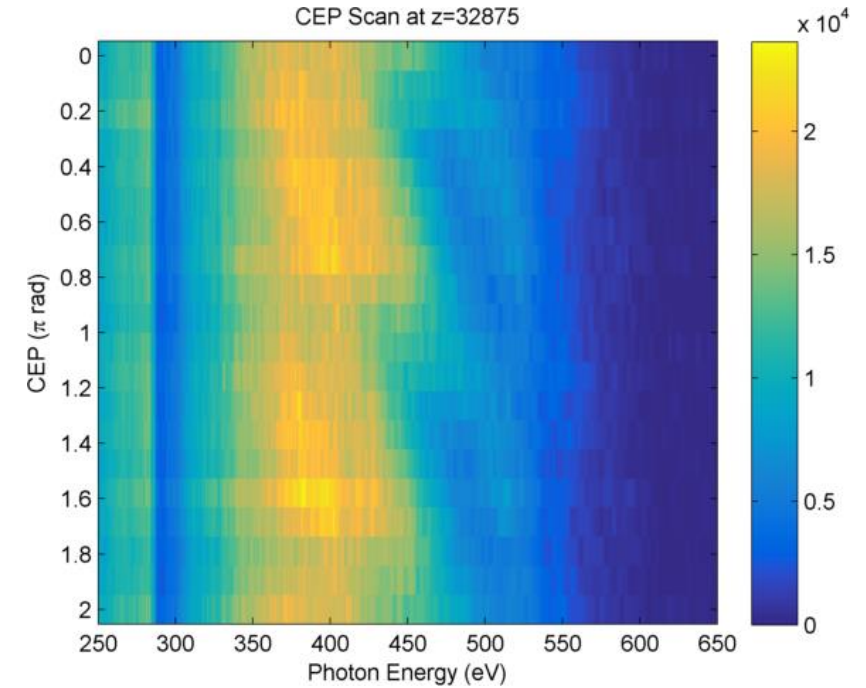
**actinoids

57 - lanthanum La 38.925 5.483	58 - cerium Ce 40.443 5.723	59 - praseodymium Pr 41.991 5.964	60 - neodymium Nd 43.569 6.208	61 - promethium Pm 45.184 6.459	62 - samarium Sm 46.834 6.716	63 - europium Eu 48.519 6.977	64 - gadolinium Gd 50.239 7.243	65 - terbium Tb 51.996 7.514	66 - dysprosium Dy 53.789 7.790	67 - holmium Ho 55.618 8.071	68 - erbium Er 57.486 8.358	69 - thulium Tm 59.390 8.648	70 - ytterbium Yb 61.332 8.944
89 - actinium Ac 106.755 15.871	90 - thorium Th 109.651 16.300	91 - protactinium Pa 112.601 16.733	92 - uranium U 115.606 17.166	93 - neptunium Np 118.673 18.xxx	94 - plutonium Pu 121.805 19.xxx	95 - americium Am 125.004 19.xxx	96 - curium Cm 128.23 20.xxx	97 - berkelium Bk 131.573 21.xxx	98 - californium Cf 134.952 21.xxx	99 - einsteinium Es 138.446 22.xxx	100 - fermium Fm 142.010 23.xxx	101 - mendelevium Md 145.658 23.xxx	102 - nobelium No 149.379 24.xxx

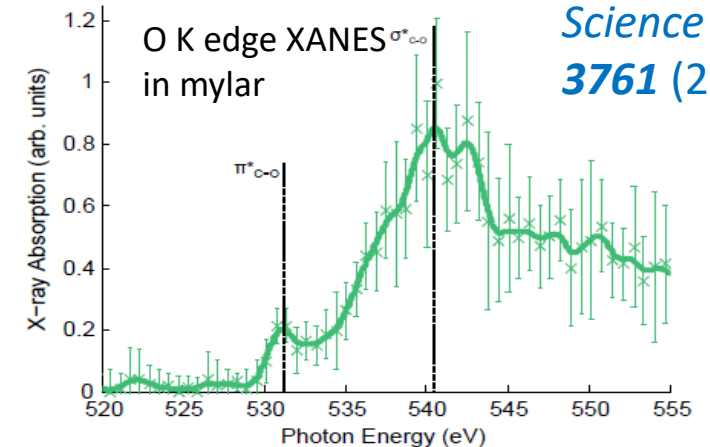
2. Attosecond supercontinuum pulses from HHG covering 150 – 600 eV



Using a few-cycle 1.8 μm pulses CEP stable pulses & by changing gas species, pressure, and focus position, an isolated sub-femtosecond pulse can be generated centred at photon energies from 150 – 600 eV



Top left: 200-350eV
 Top right: 280-350 eV
 Bottom left: 350-450 eV
 Bottom right: 500-600eV



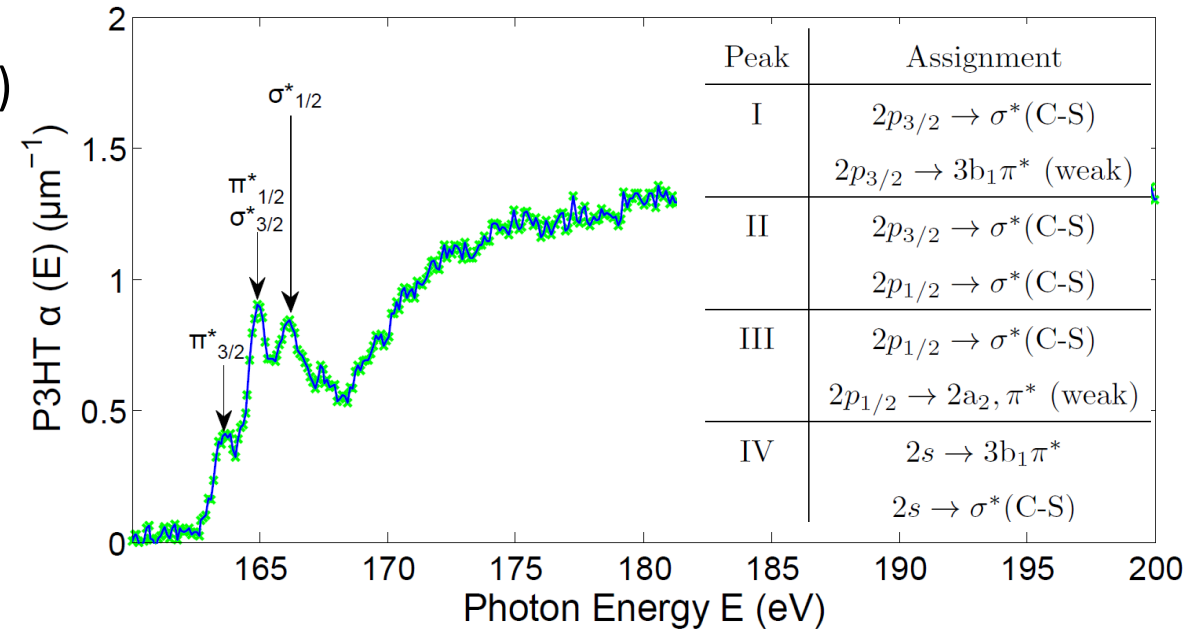
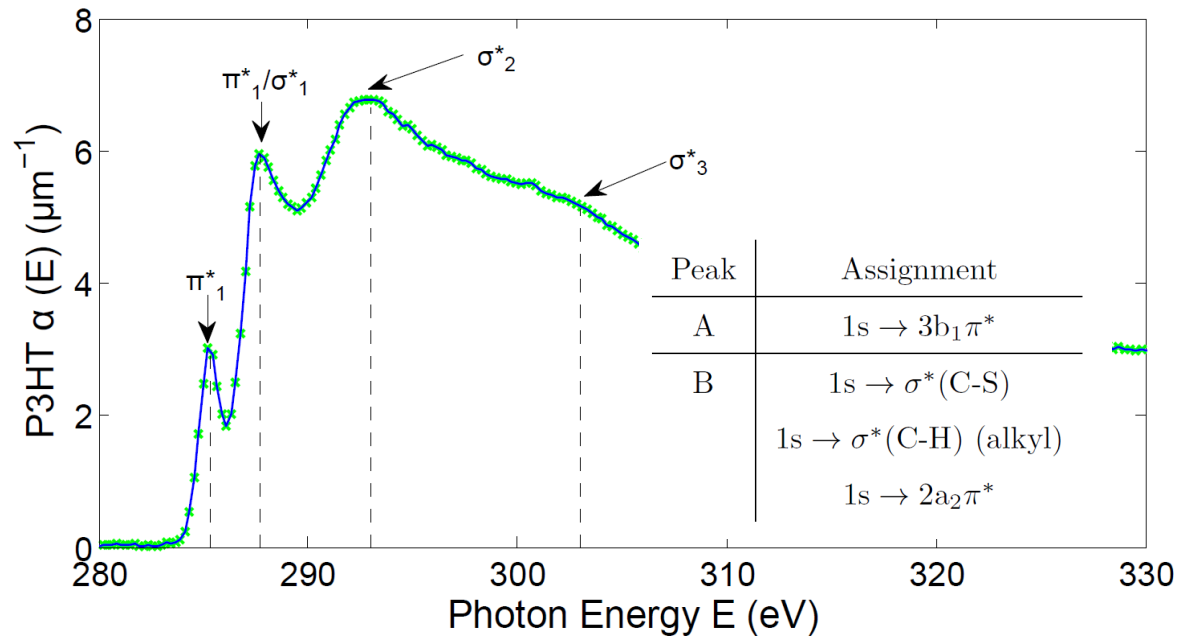
A. Johnson et al
Science Advances 4,
3761 (2018)

2. XANES with SXR HHG super-continuum in organic semiconductor P3HT



For time-resolved SXR absorption we fabricate free standing P3HT* (polythiophene) polymer films of ~ 100 nm thickness using spin coating methods.

* *Base material for important class of polymer optoelectronic materials*

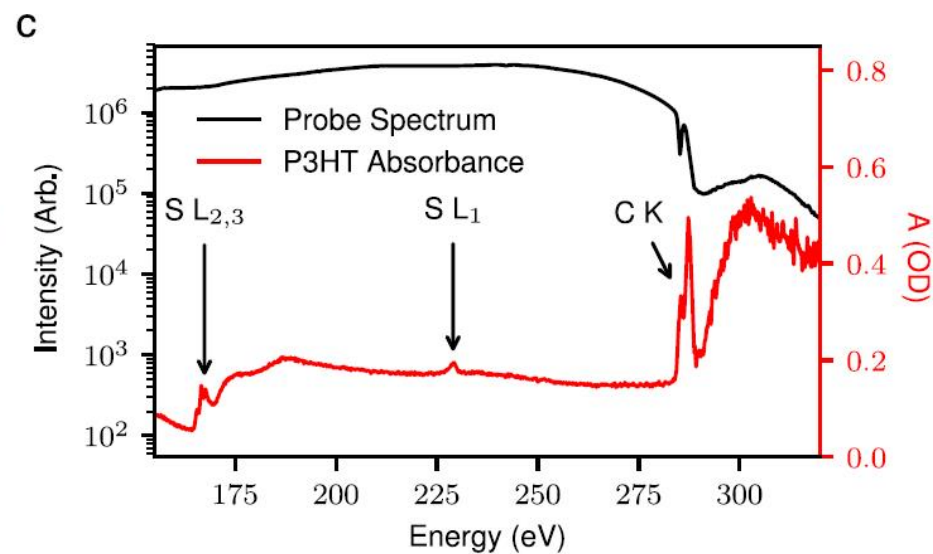
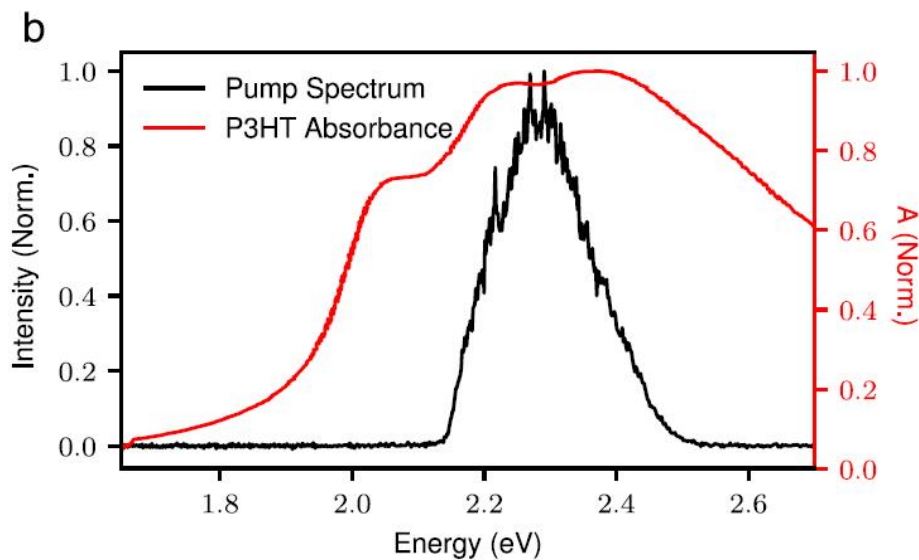
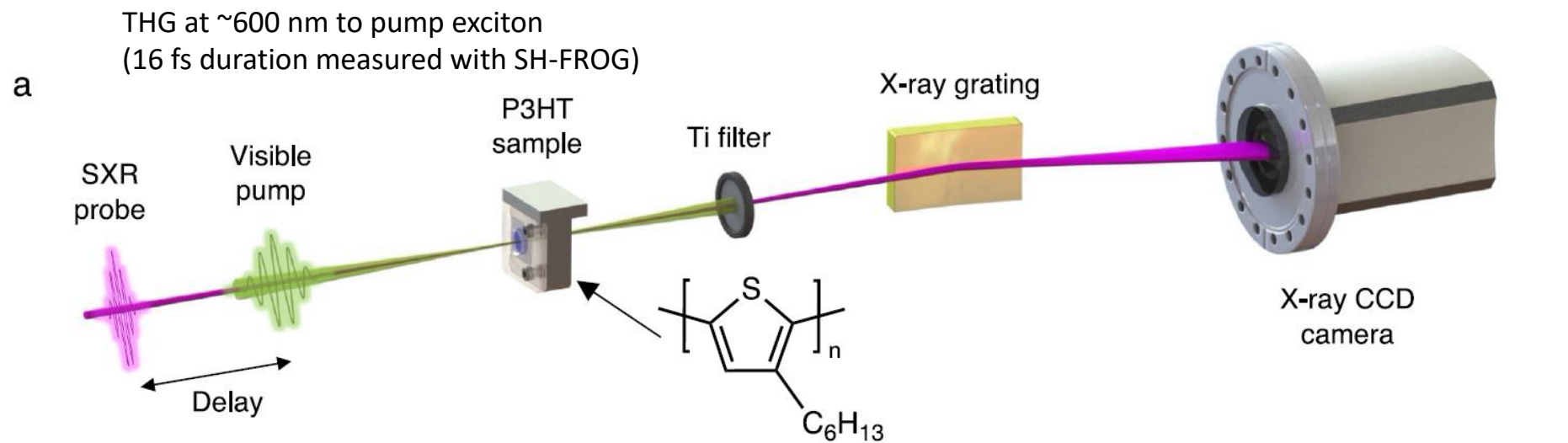


A. Johnson et al, *Structural Dynamics*, 3, 062603 (2016)

XANES at S L edge and C K edge measured in P3HT films with HHG from our source (likely close to an IAP).

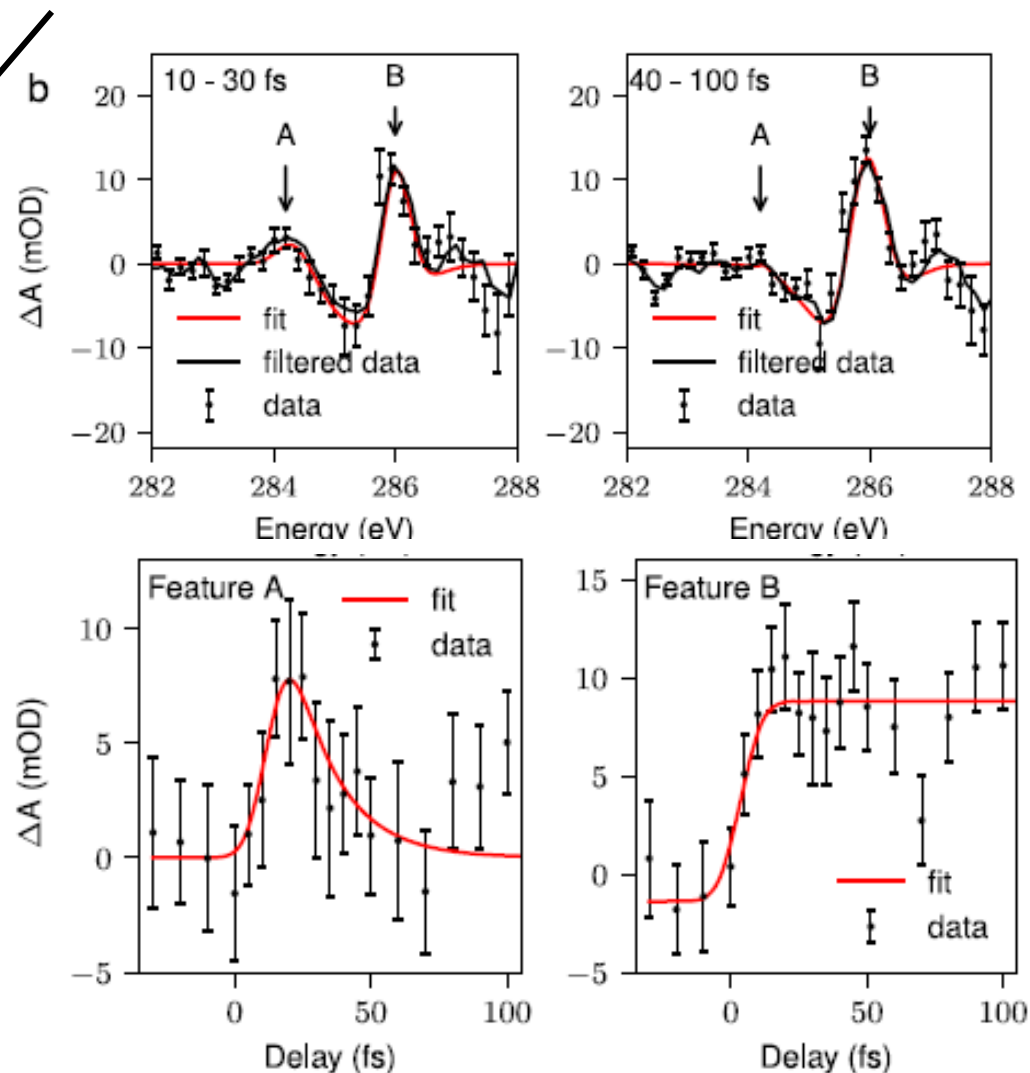
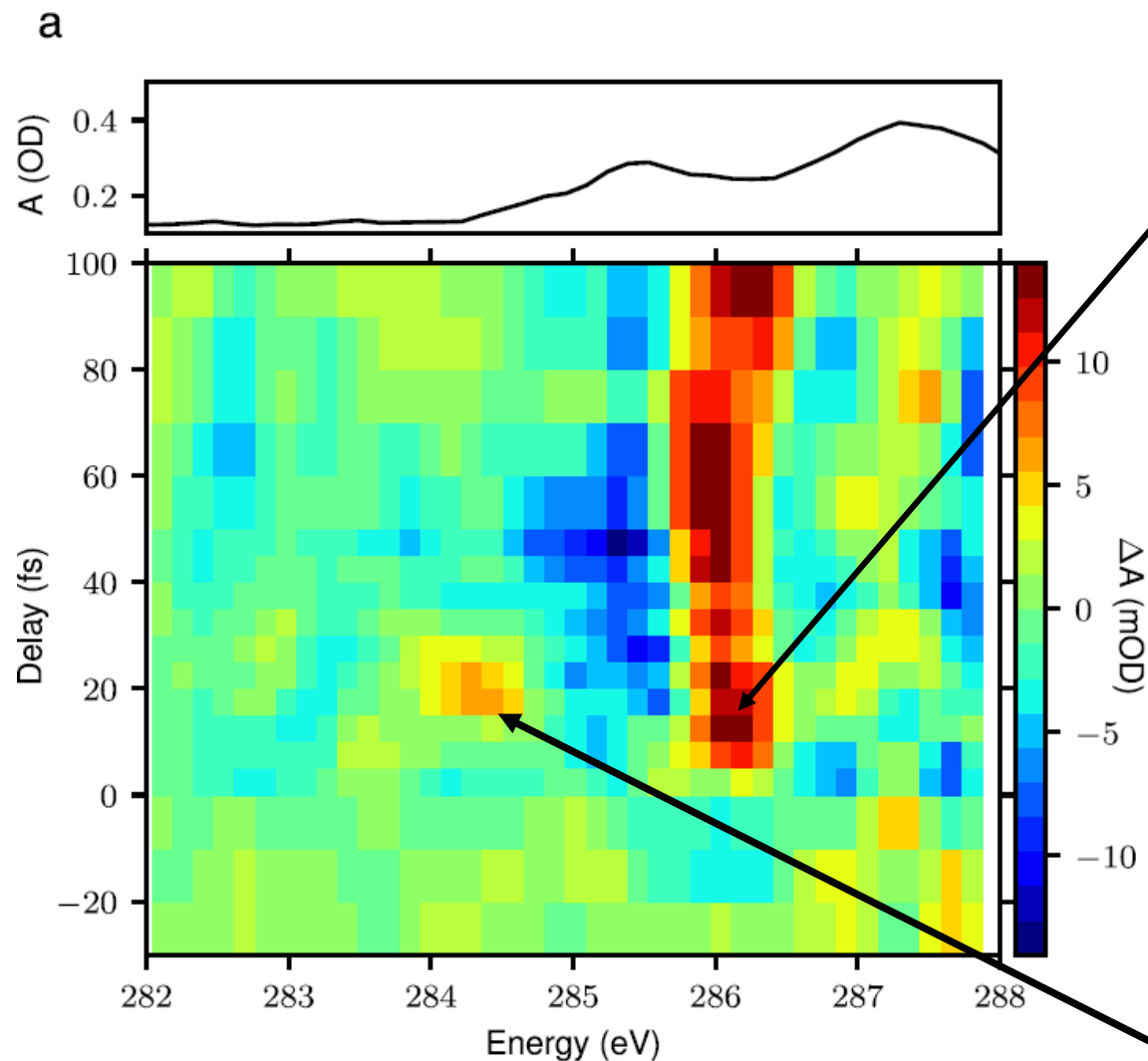
2. Time-resolved HHG based XAS in organic semiconductor P3HT

Dougie Garratt, David Wood, Lukas Miseikis, Esben Larsen, Mary Matthews, Artem Bakulin, Tom Penfold, JPM



2. Time-dependence of XAS in P3HT at C K edge spectrum

B. Edge shift following exciton creation

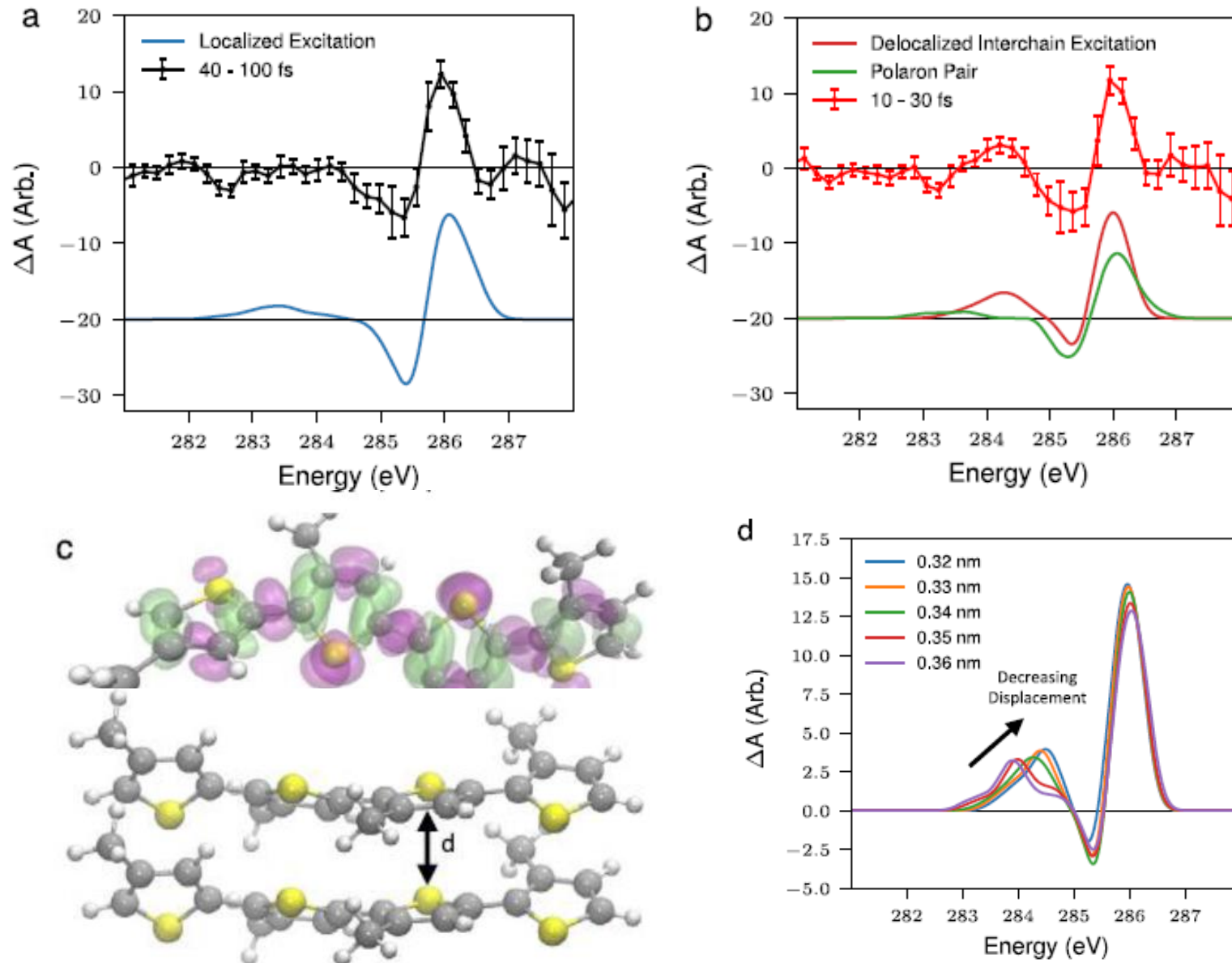


A. Transient feature – pre-edge absorption associated with exciton delocalised across neighbouring chains

2. X-ray transient absorption signal provides a direct probe of the initial delocalisation and cooling of the hot exciton formed by the pump

XAS spectra simulated with TDDFT by Tom Penfold (Newcastle)

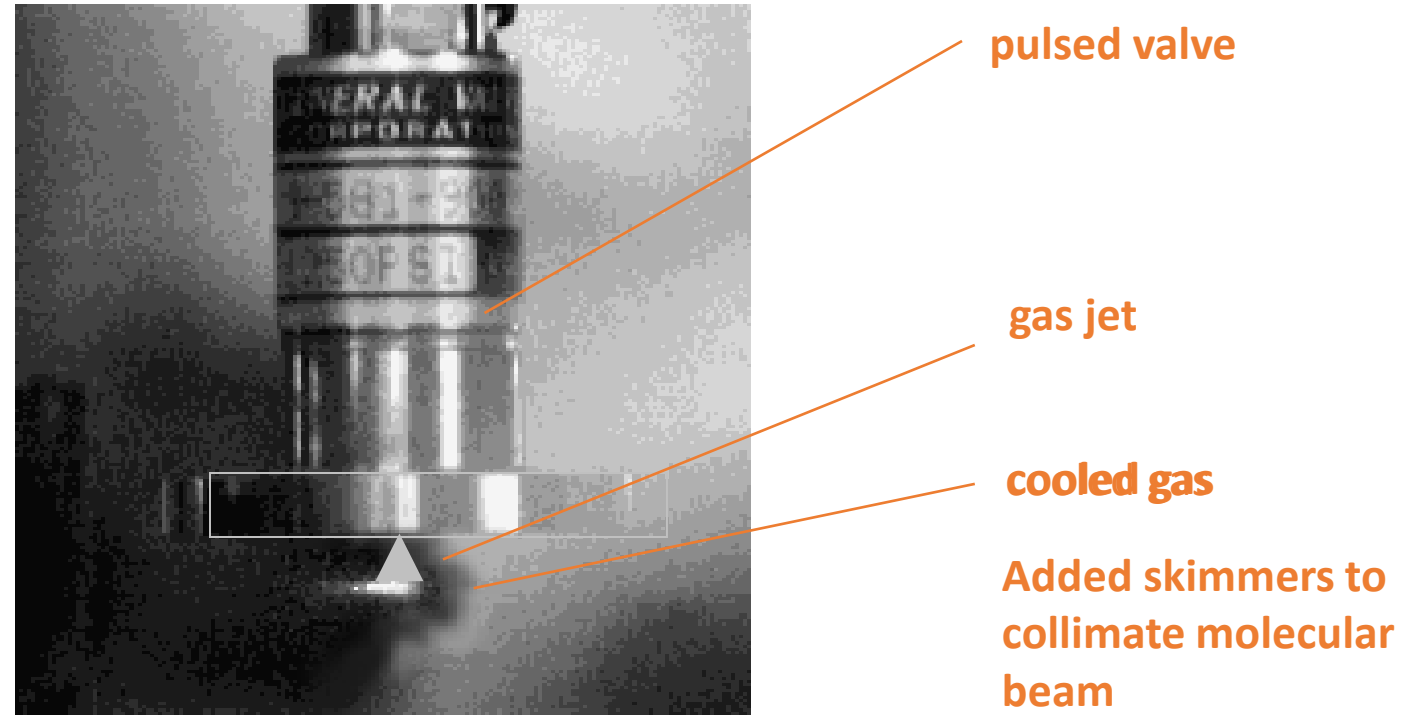
Garratt, D., Misiekis, L., Wood, D. *et al.* Direct observation of ultrafast exciton localization in an organic semiconductor with soft X-ray transient absorption spectroscopy. *Nat Commun* **13**, 3414 (2022). <https://doi.org/10.1038/s41467-022-31008-w>



2. Charge fragment methods: Photoelectron spectroscopy, Auger spectroscopy and fragmentation patterns

- Cold targets and helium droplets and cluster sources (from supersonic nozzles)
- Warm targets from ovens (effusive beams) or techniques for larger molecules (e.g. electrospray)
- Access to all particle fragments in gas phase systems, often with high efficiency enabling covariance or coincidence reconstruction of fragment correlations (ion-ion, ion-electron or electron-electron)
- Time-of-flight (TOF) ion (charge/mass fragment analysis & kinetic energy release) and electron spectrometers (for photoelectron and Auger spectrum)
- Angular distribution of ions and electrons – from an array of TOFs or VMI
- Full momentum reconstruction and coincidence of all fragments (COLTRIMs)

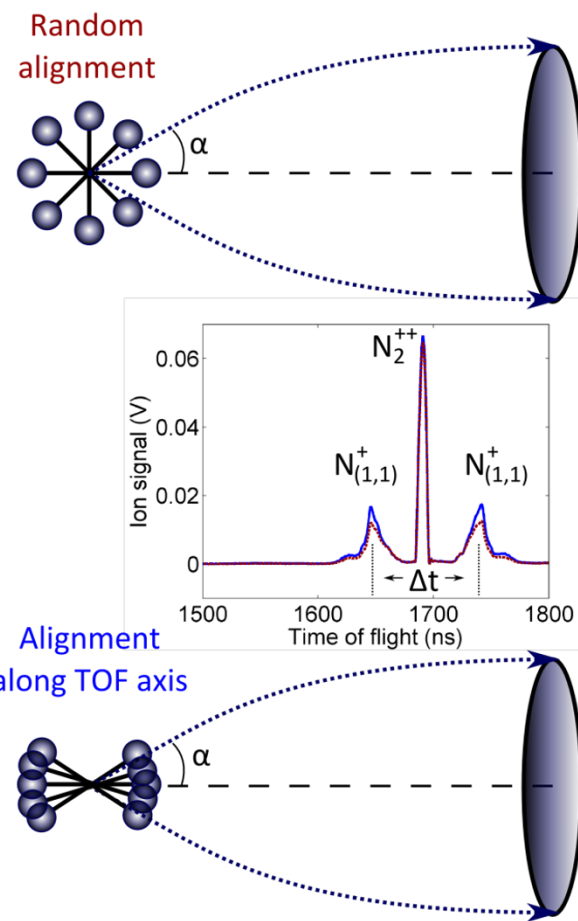
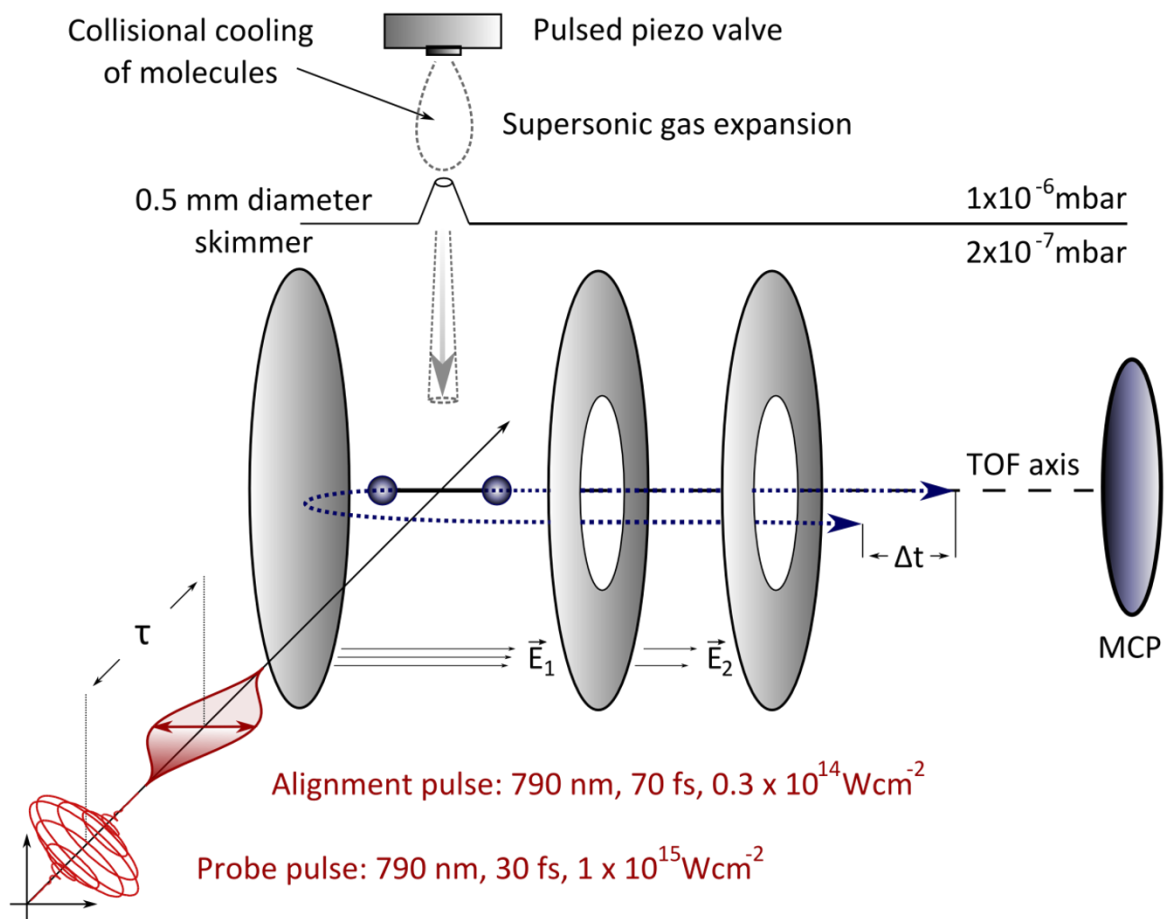
2. Cooling of rotational states of molecules is achieved by supersonic expansion of gas through a nozzle into vacuum



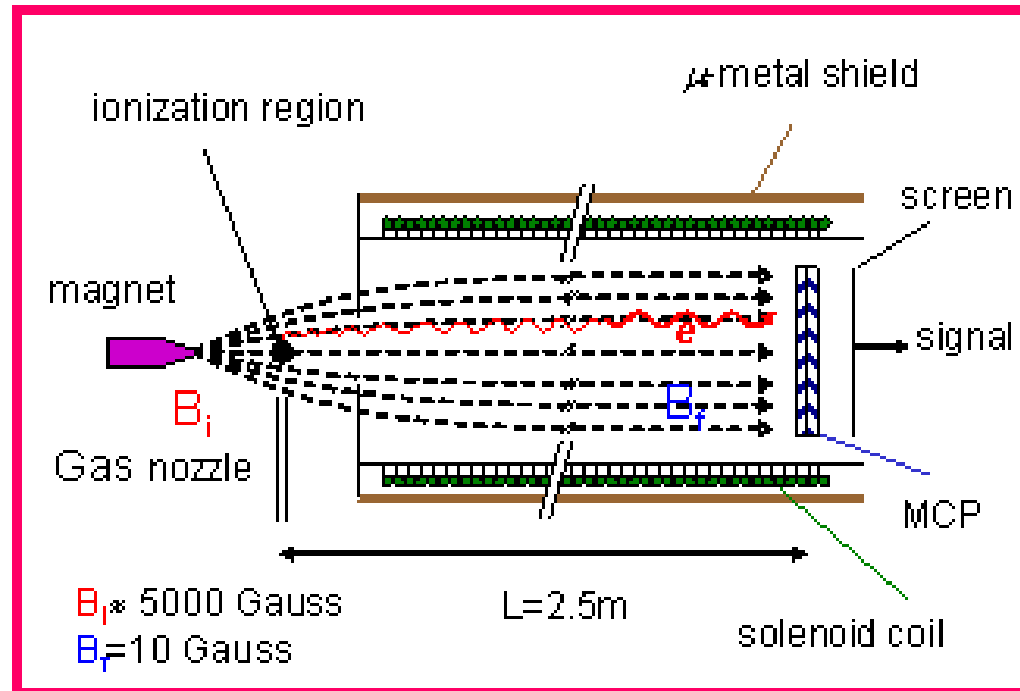
When the gas expands supersonically into a vacuum it cools rapidly with collisions causing the translational cooling to couple to rotational cooling.

Nozzle diameter d is typically 100 – 500 microns and experiments are performed at a scaled distance x/d : $x/d = 5 - 100$

2. Rotationally cooled molecules can be aligned in the laboratory frame using pulsed laser fields



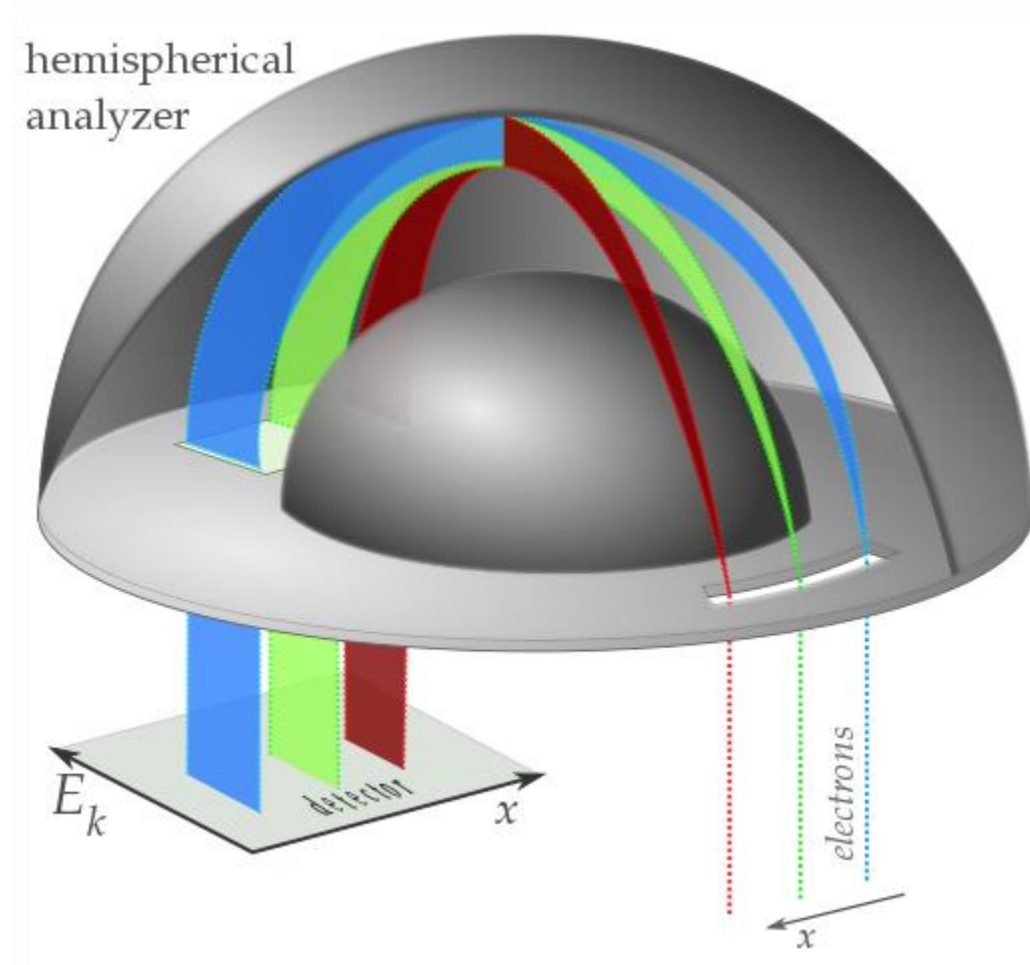
2. Electron/Ion TOFs are widely used, high collection efficiency with high energy resolution achieved using the magnetic bottle configuration



P. Kruit and F. H. Read, J. Phys. E 16, 313 (1983)

J. H. D. Eland et al, Phys. Rev. Lett. 90, 053003 (2003)

2. Other widely used electron spectrometer options are the hemispherical analyzers that can preserve some angular resolution



Velocity map imaging (VMI) is widely used to retain angular distribution information from ions and electrons by projection on a 2D plane

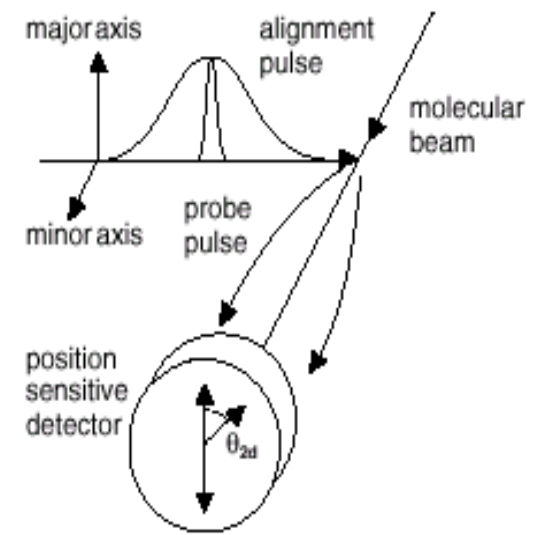
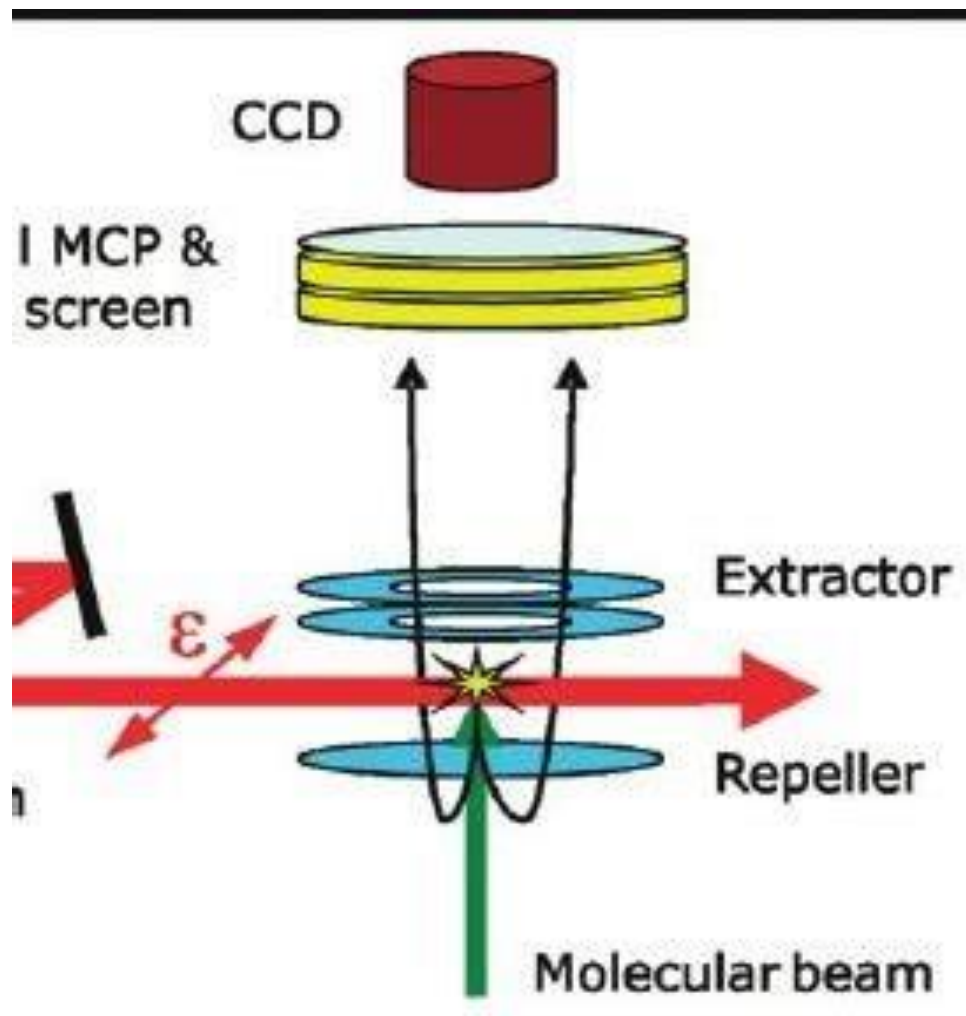
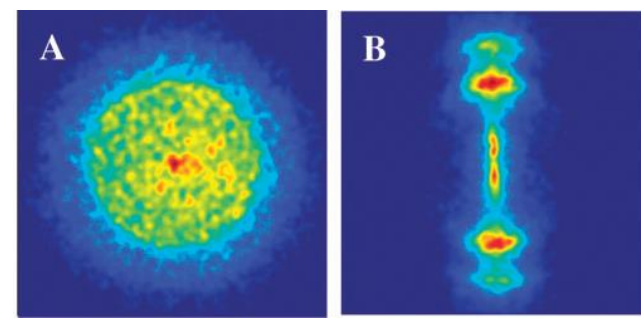
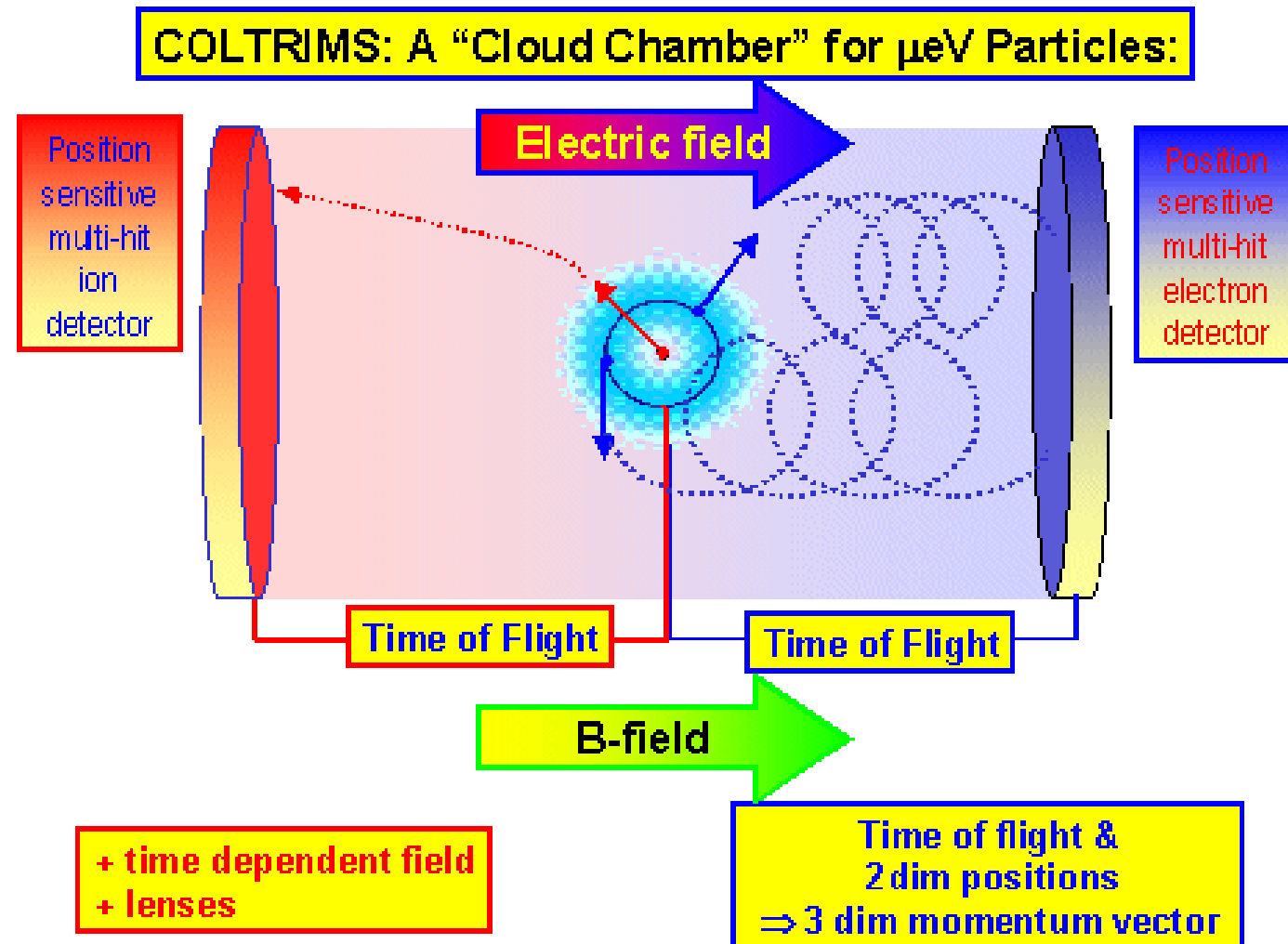


FIG. 3. Schematic overview of the apparatus. The major axis of the YAG polarization is parallel to the detector surface and the minor axis is parallel to the molecular beam.



2. Full 3D fragment distribution of both electrons and ions can be preserved in a COLTRIMS (COLd Target Recoil Ion Momentum system)

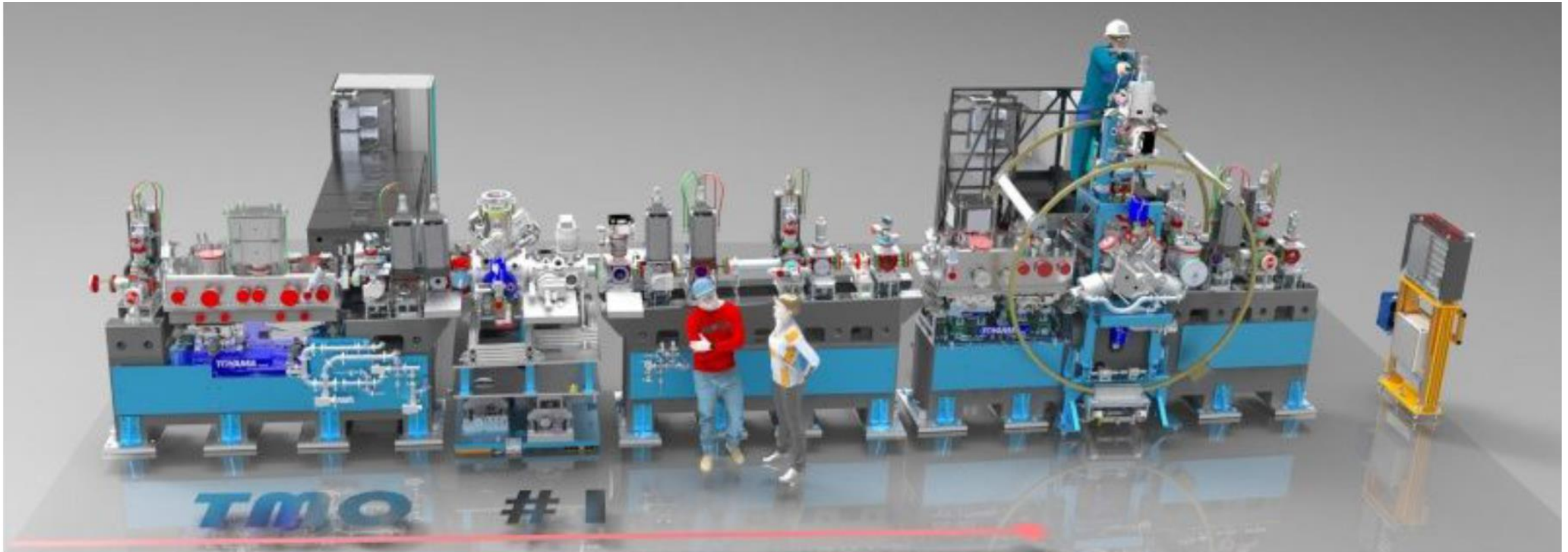


2. Upgraded end-stations are being built for high rep-rate attosecond capability at LCLS – operational from 2023

Variable Kirkpatrick-Baez focus mirrors to achieve high intensity

High resolution electron spectrometers, ion velocity mapping VMI, and coaxial electron VMI + single shot spectral monitor

COLTRIMS – Momentum microscope for coincidence detection



End of part I

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