

Chemical Dynamics



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CHEMISCHES
INSTITUT



IWR
Interdisciplinary Center
for Scientific Computing



Ultrafast X-ray Summer School 2019

QUESTION

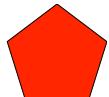


I am a ...

 ... a physicist.

 ... a chemist.

 ... a biologist / life scientist.

 ... I still need to find out.

QUESTION



I mostly work as ...

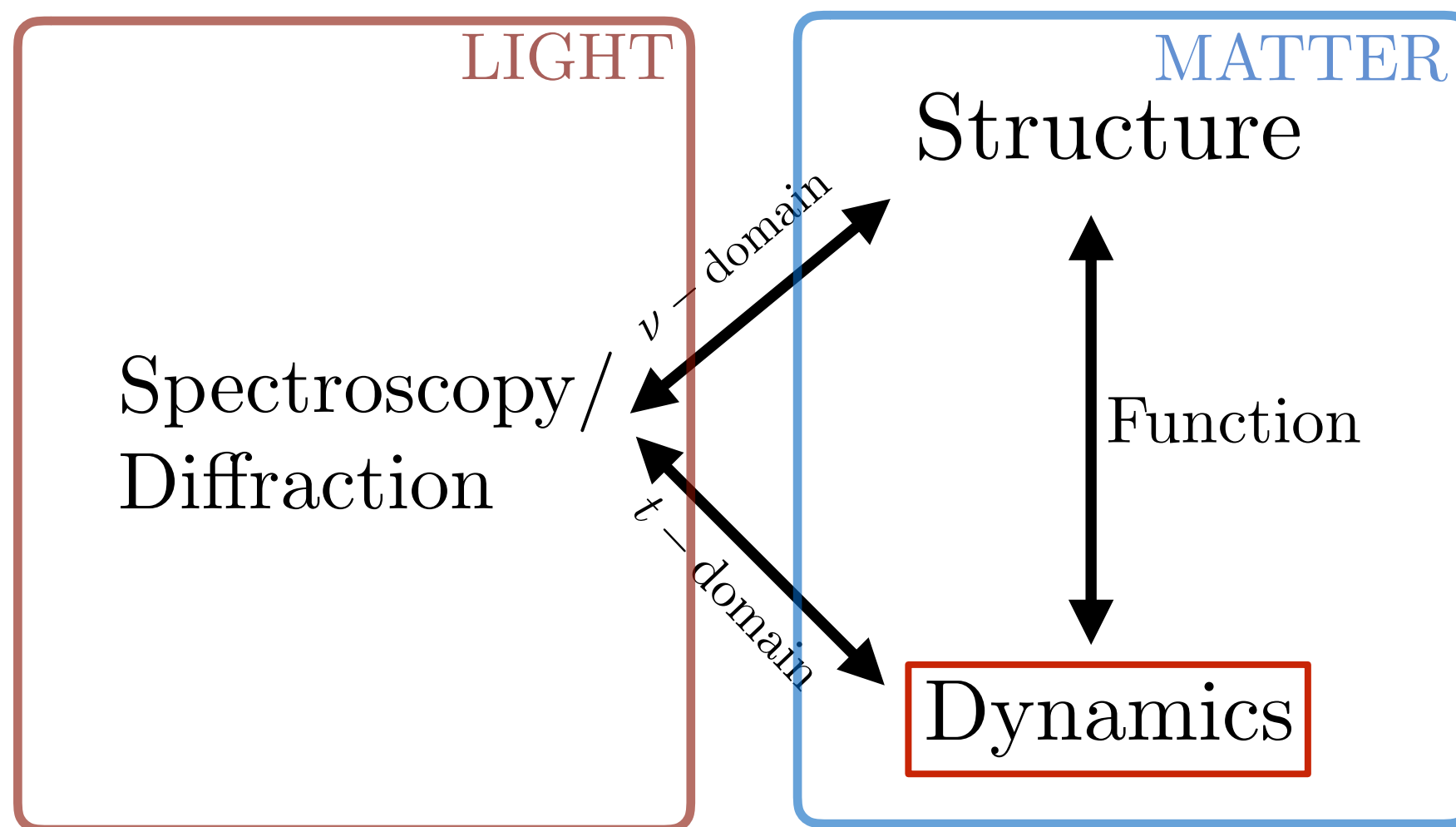
 ... an experimentalist.

 ... a theoretician.

 ... both an experimentalist and a theoretician.

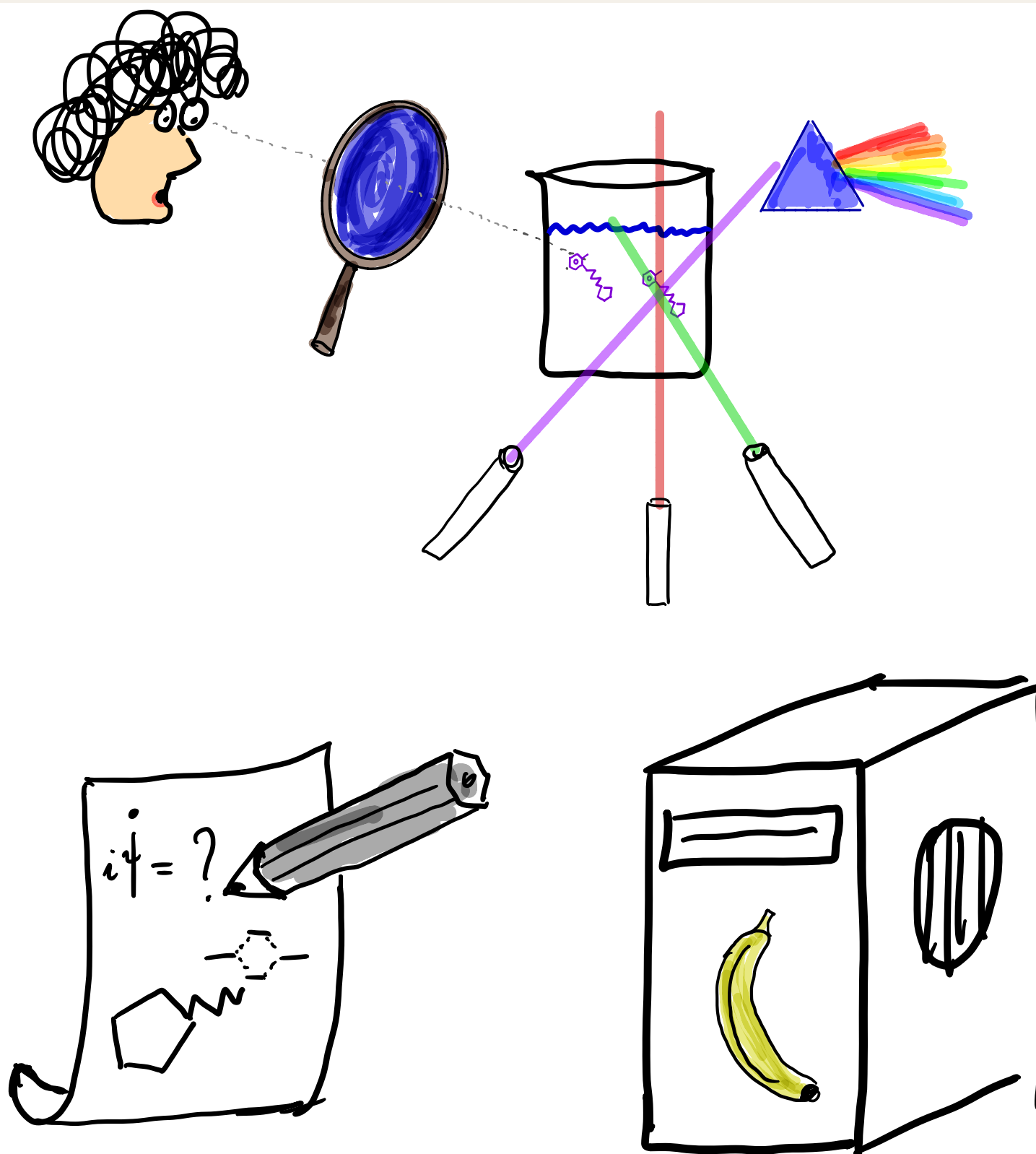


- Introduction
- From kinetics to dynamics of chemical processes
- Theoretical description of the structure and dynamics of molecules
- Time-dependent description of molecular spectroscopy
- Tracing chemical dynamics with (ultrafast) x-rays.
Examples and applications.

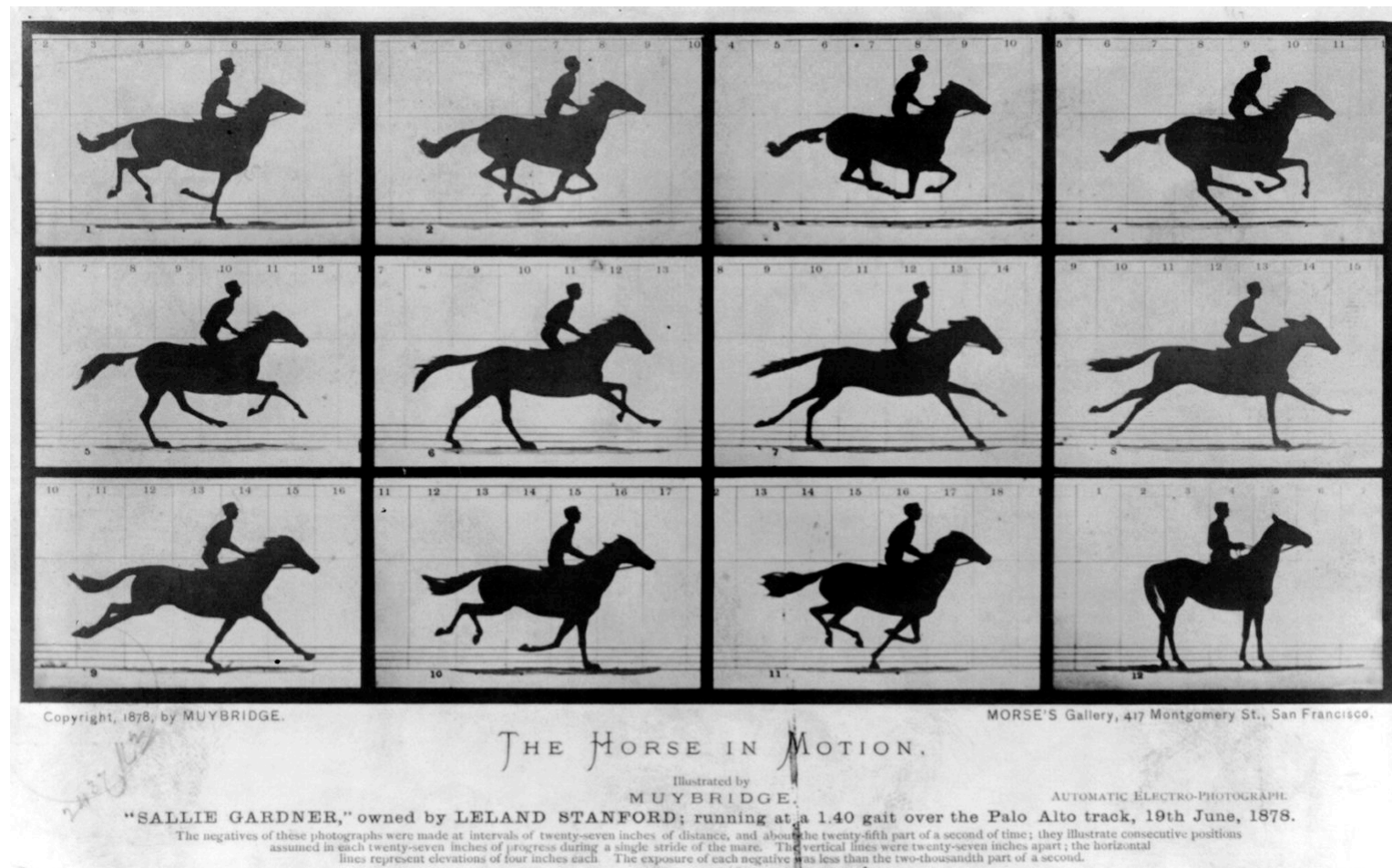


- Molecular level reaction mechanisms
- Elementary charge and energy transfer processes
- Means of probing them
- Means of controlling them

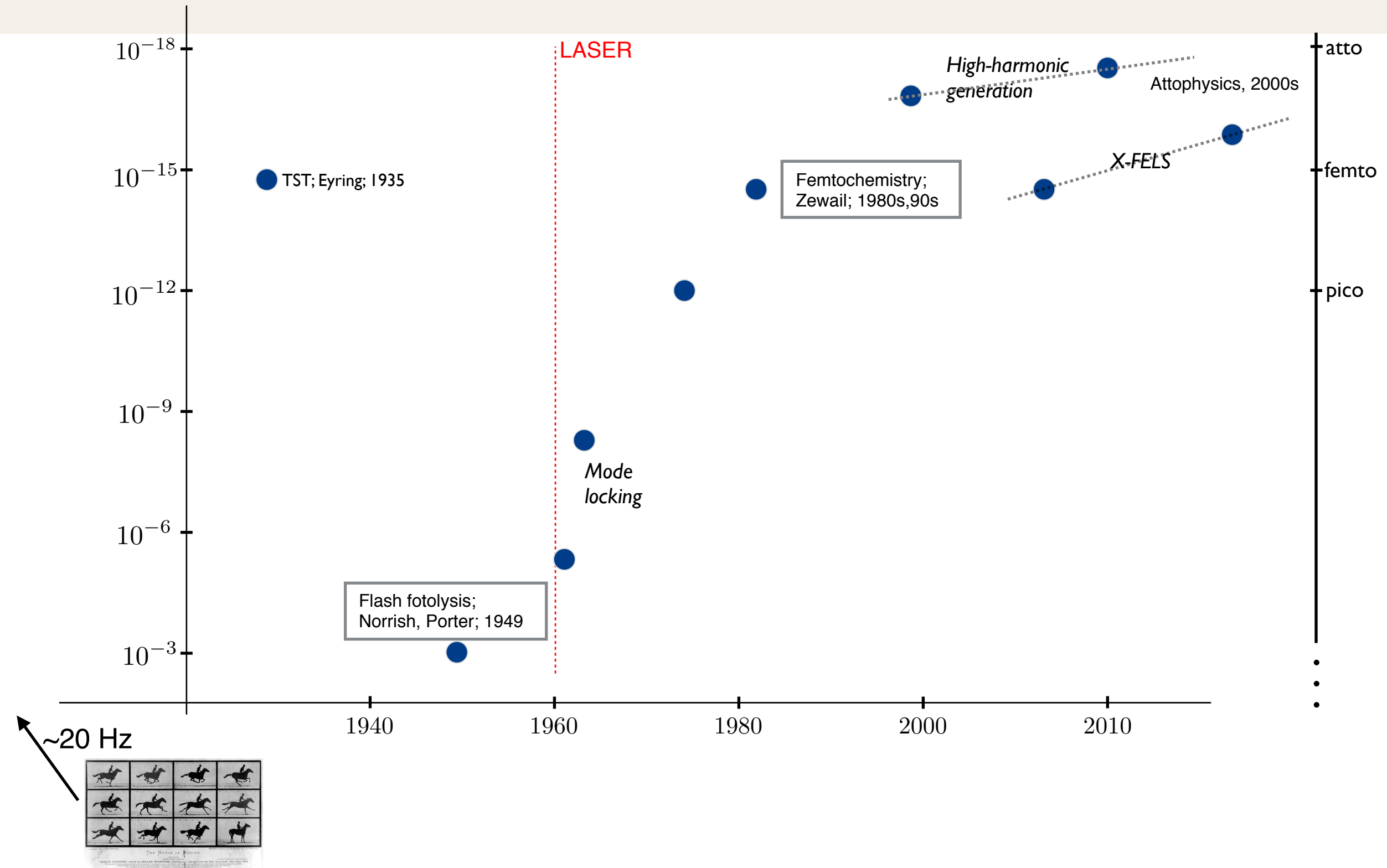
Chemical (Molecular Reaction) Dynamics



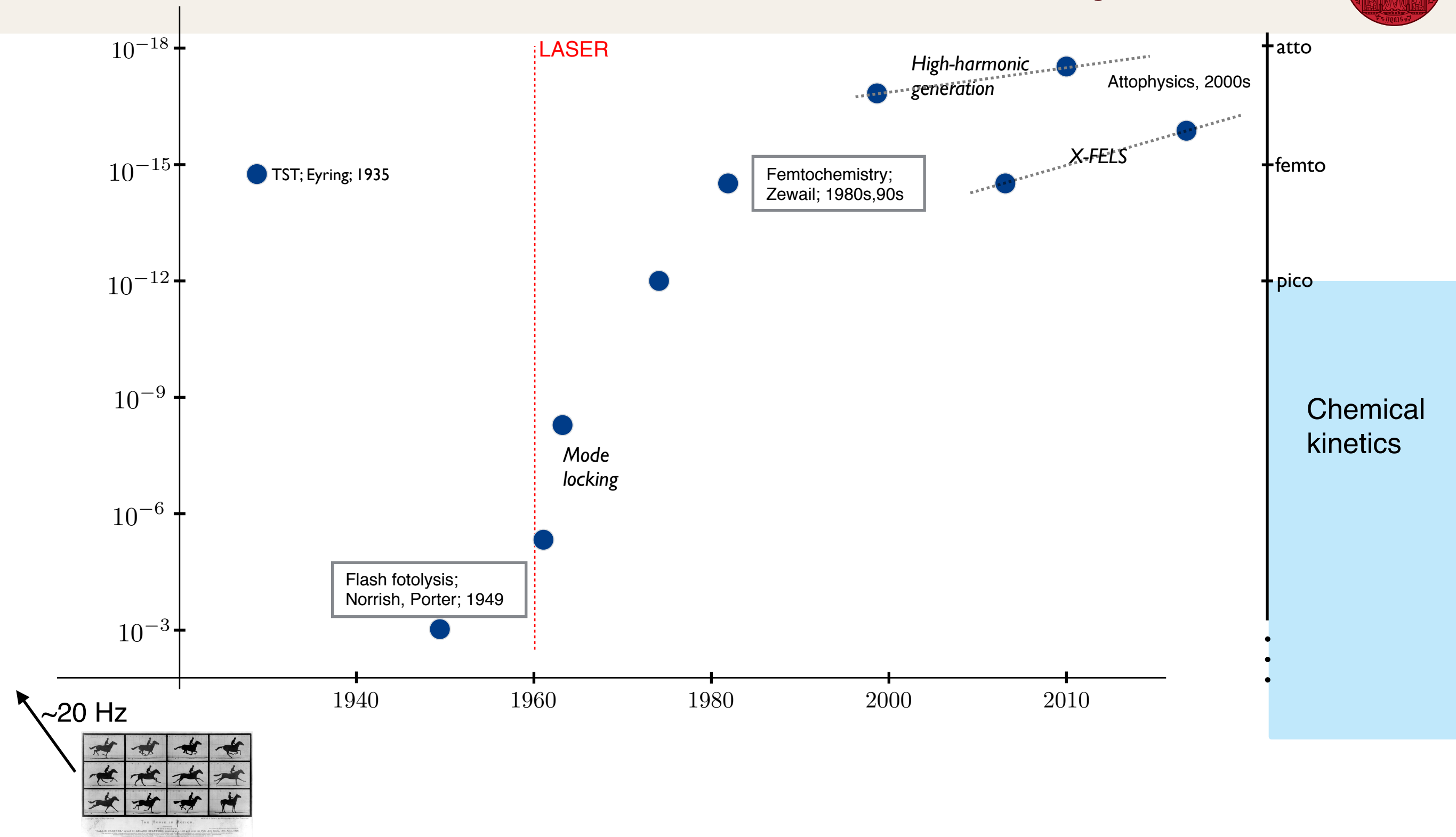
Time resolved measurements of kinetics and dynamics



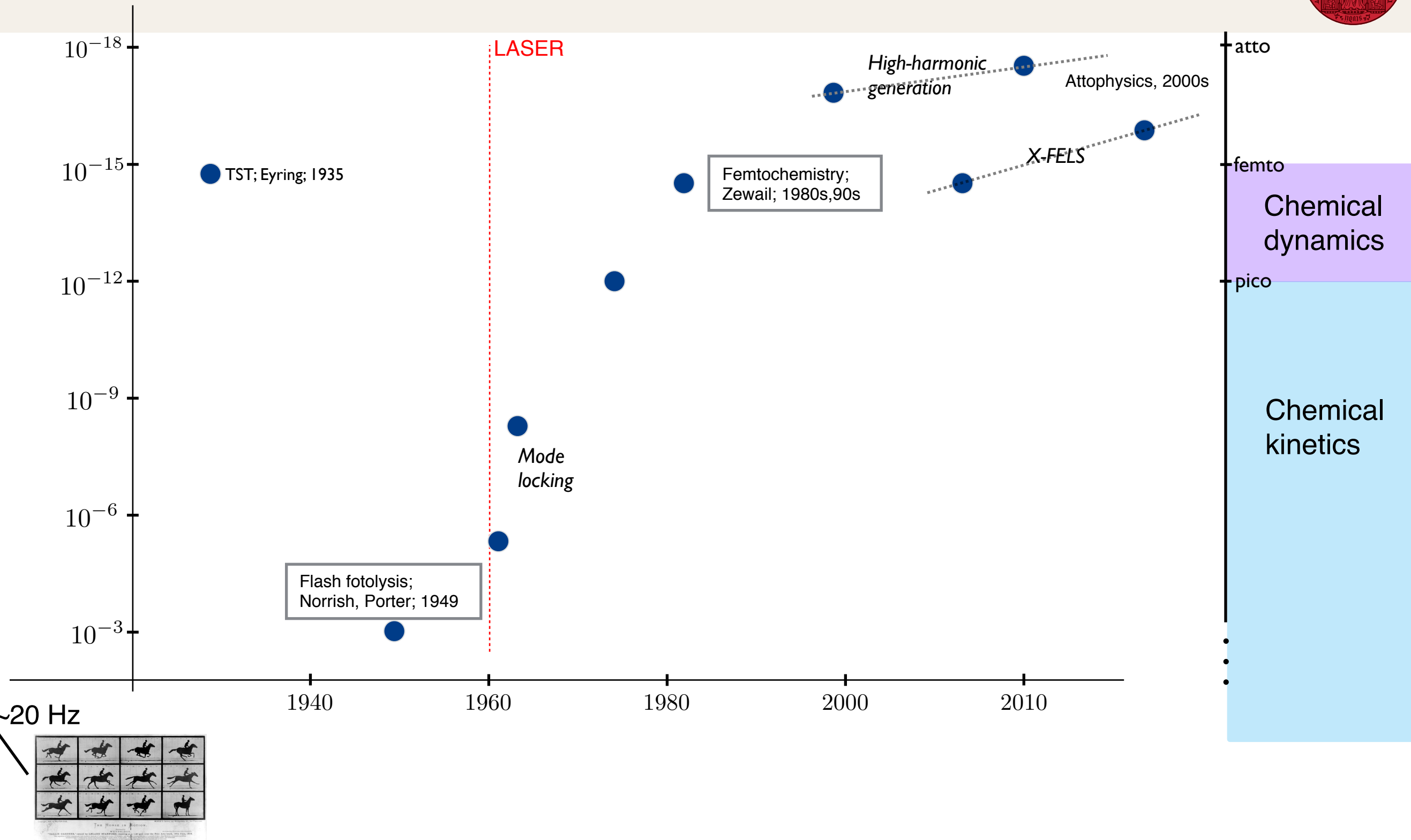
Time resolved measurements of kinetics and dynamics



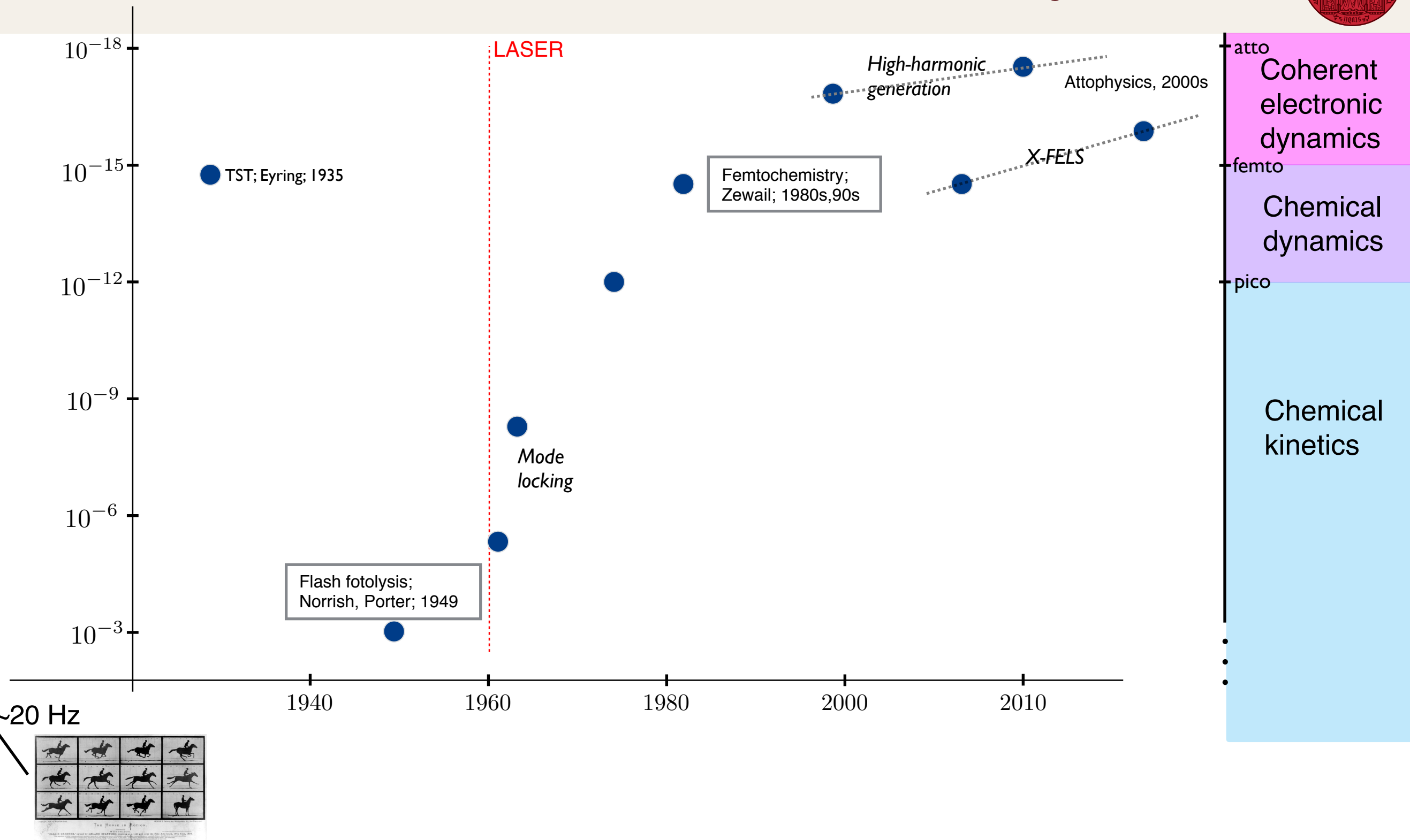
Time resolved measurements of kinetics and dynamics



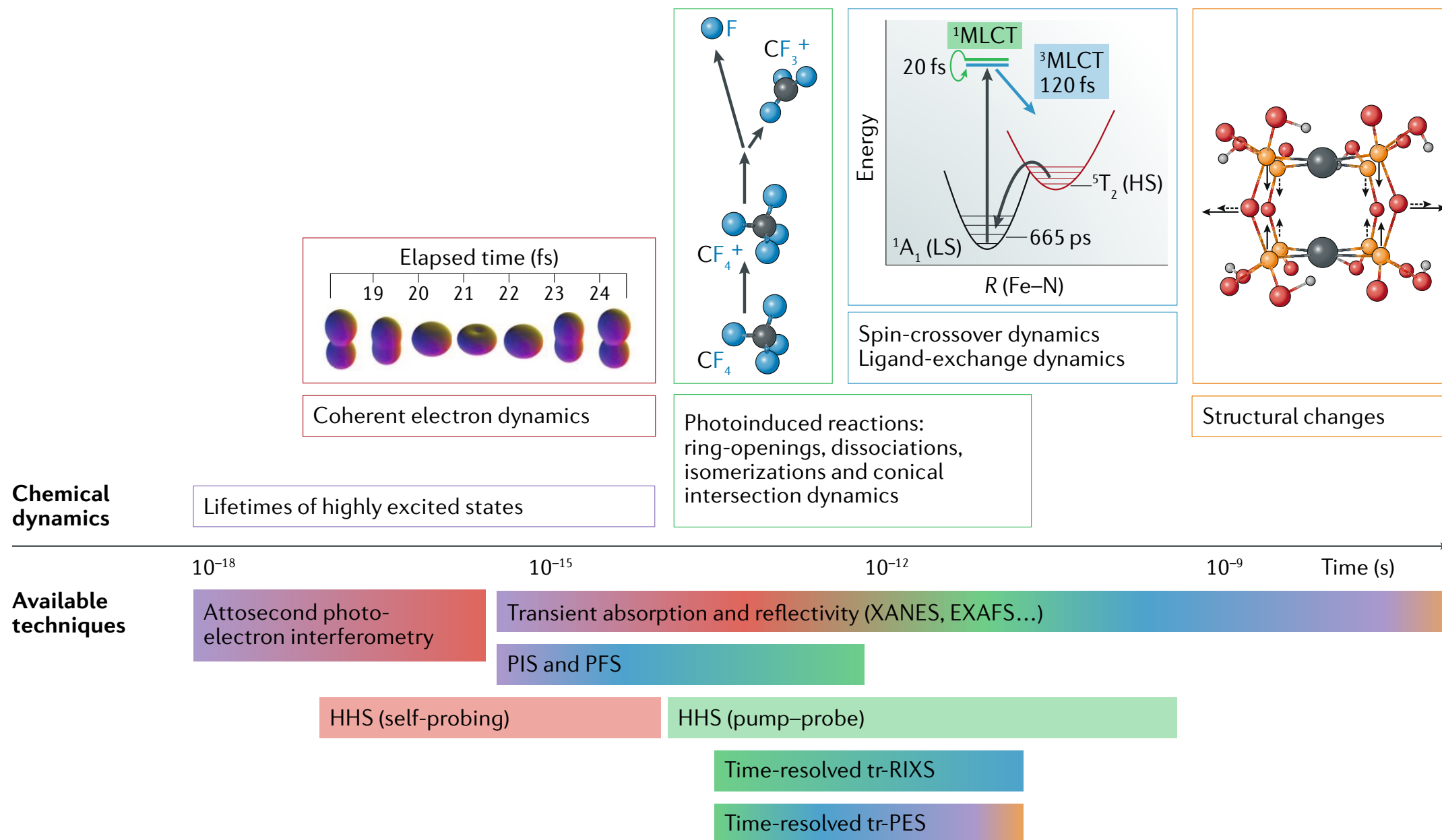
Time resolved measurements of kinetics and dynamics



Time resolved measurements of kinetics and dynamics



Typical time-scales in Chemical Dynamics



The ultrafast X-ray spectroscopic revolution in chemical dynamics
 Kraus, Zürich, Cushing, Neumark, Leone, Nat. Rev. Chem., **2**, 82 (2018)



From kinetics to dynamics of chemical processes

QUESTION

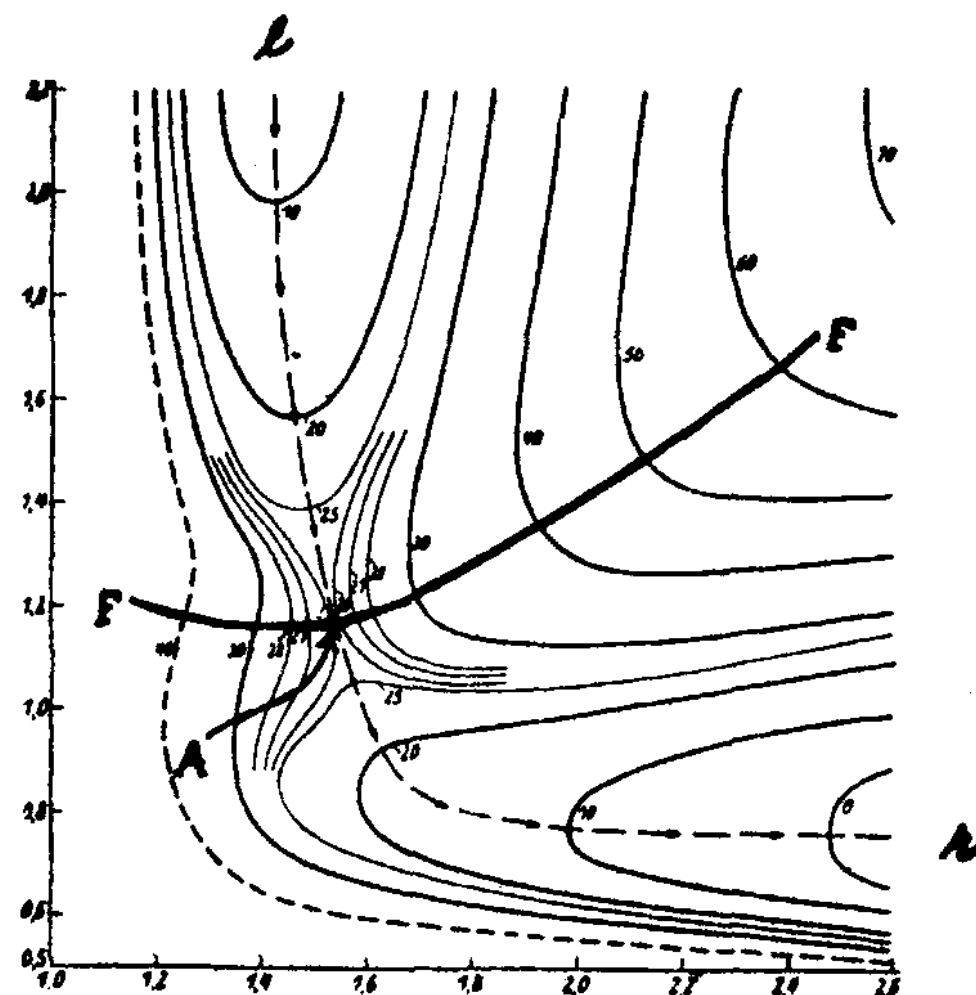
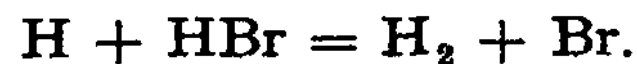


What is a transition state?

-  An unstable configuration of a molecule
-  The maximum energy configuration along the minimum energy pathway connecting stable structures.
-  A dividing surface in phase space along which recrossing is minimized.
-  All of the above.

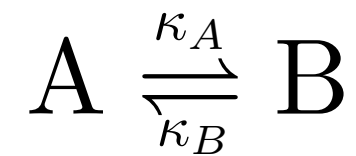
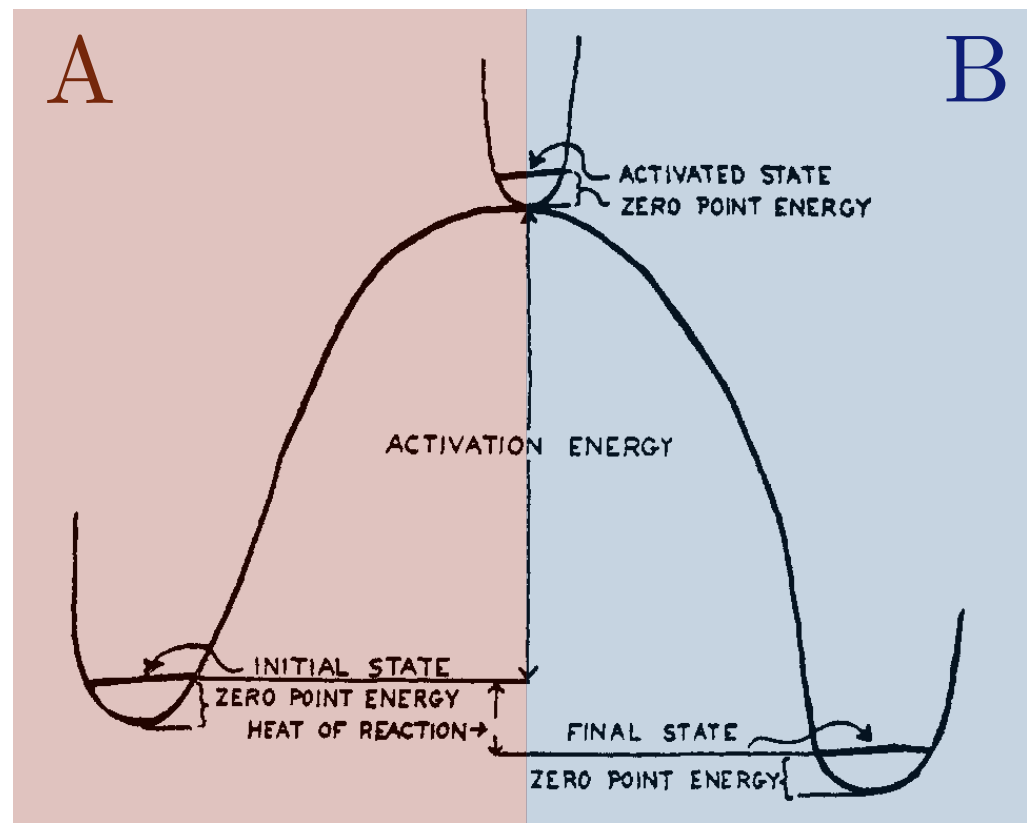


Fig. 1.—The regions (l) and (r), the activation point A and the surface F, in the section of the configuration space, corresponding to a linear configuration of the atoms H, H, Br. The ordinate is the distance of the two H atoms in A, the abscissa the distance of the middle H from the Br. The numbered lines are contour lines of the energy surface, constructed by Eyring and Polanyi for the reaction



Wigner 1938

From kinetics to dynamics



$\langle \Theta_A \rangle$: fraction of A

$$\frac{d}{dt} \langle \Theta_B \rangle = \kappa_A \langle \Theta_A \rangle - \kappa_B \langle \Theta_B \rangle$$

$$\frac{d}{dt} \langle \Theta_A \rangle = \frac{d}{dt} \langle \Theta_B \rangle = 0$$

$$\frac{\langle \Theta_B \rangle}{\langle \Theta_A \rangle} = \frac{\kappa_A}{\kappa_B} = K_{\text{eq}}$$

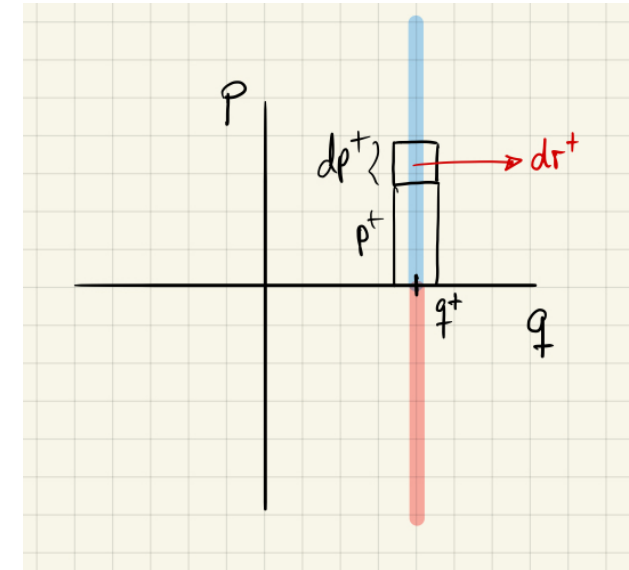
From kinetics to dynamics



Micro-canonical reaction rate in the energy shell $E \pm dE$

$$\kappa(E) = \frac{N^\ddagger(E - \epsilon_0)}{h\rho(E)}$$

ϵ_0 : classical barrier



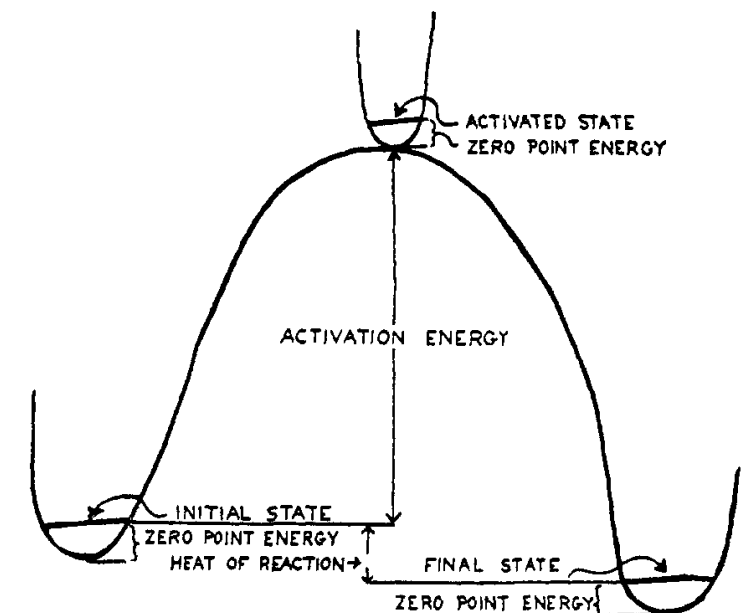
$$\Gamma^\ddagger = \int_0^{E-\epsilon_0} dr^\ddagger = \frac{1}{h} N^\ddagger(E-\epsilon_0) := k(E) \rho(E)$$

$$\kappa(T) = \frac{1}{Q(T)} \int_0^\infty \kappa(E) e^{-E/KT} dE$$

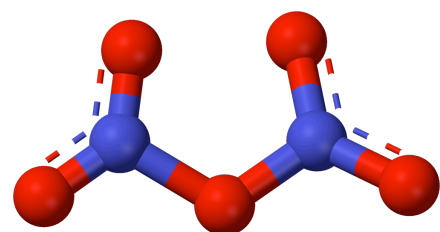
$$\kappa(T) = \frac{KT}{h} \frac{Q^\ddagger}{Q} e^{-\epsilon_0/KT}$$

$$T = 300 \text{ K} \implies \frac{KT}{h} = 6.25 \times 10^{12} \text{ s}^{-1}$$

$\approx 160 \text{ fs per attempt}$



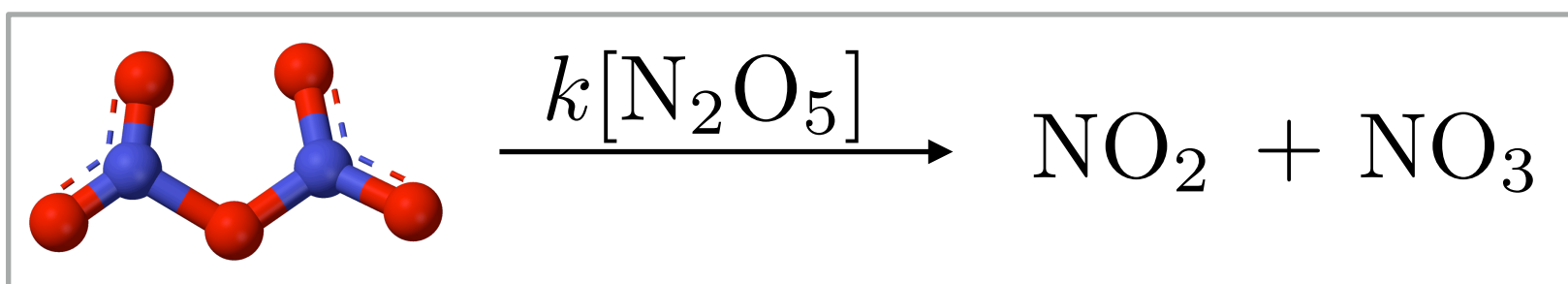
From kinetics to dynamics



$$-\frac{d[\text{R}]}{dt} = k[\text{R}]$$

Viggiano et al., *J. Chem. Phys.* **74** 6113 (1981)
Rate constants for the collisional dissociation of N₂O₅ by N₂.
KJ Leidler, *J. Chem. Ed.* **61** 494 (1984)
The development of the Arrhenius Equation.

From kinetics to dynamics



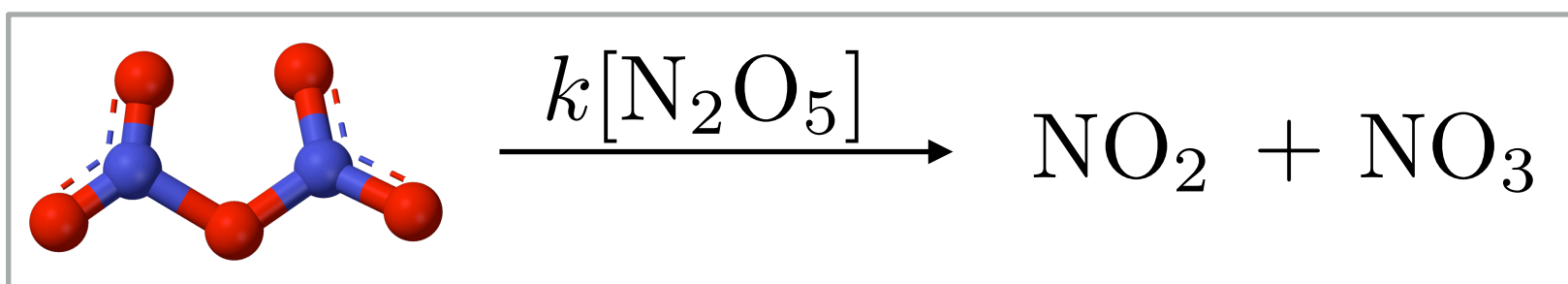
$$-\frac{d[\text{R}]}{dt} = k[\text{R}]$$

$$k(T)[s^{-1}] = A e^{-B/T} \quad \text{S Arrhenius, Z phys. Chem. 4, 226 (1889)}$$

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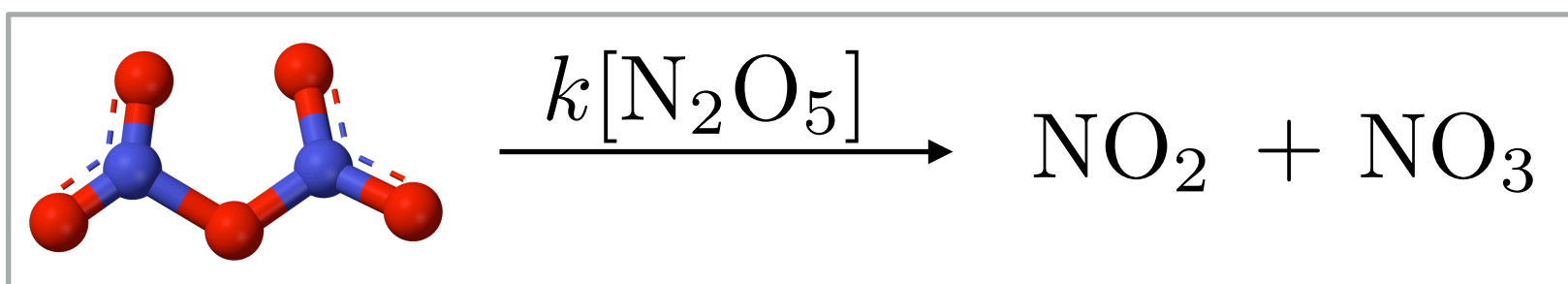
$$A = 5.08 \times 10^{13} [s^{-1}]$$

$$\frac{1}{A} = \tau = 1.97 \times 10^{-14} [\text{s}] \Rightarrow 19.7 [\text{fs}]$$

Viggiano et al., *J. Chem. Phys.* **74** 6113 (1981)
Rate constants for the collisional dissociation of N₂O₅ by N₂.

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From kinetics to dynamics

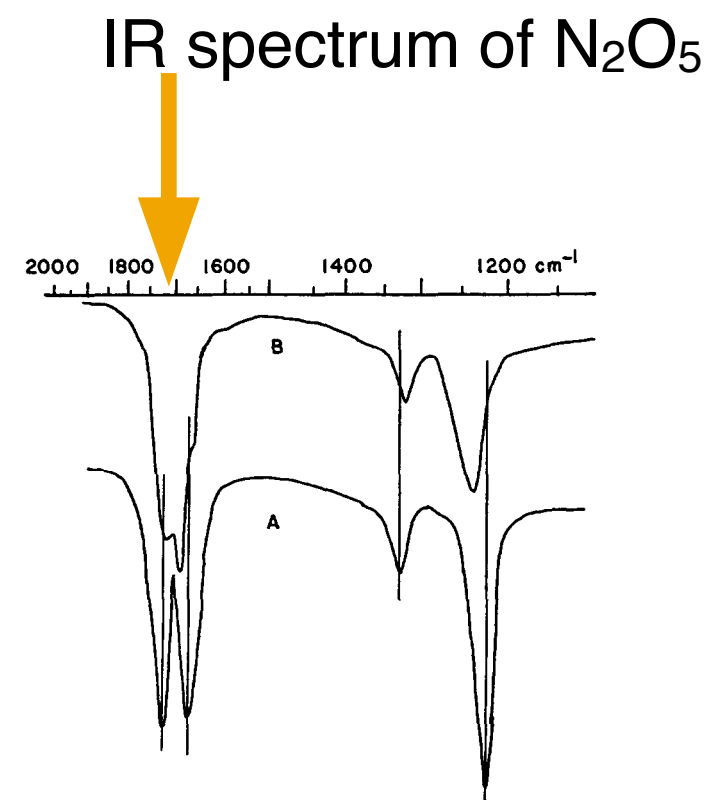


$$-\frac{d[\text{R}]}{dt} = k[\text{R}]$$

$$k(T)[s^{-1}] = A e^{-B/T} \quad \text{S Arrhenius, Z phys. Chem. 4, 226 (1889)}$$

$$A = 5.08 \times 10^{13} [s^{-1}]$$

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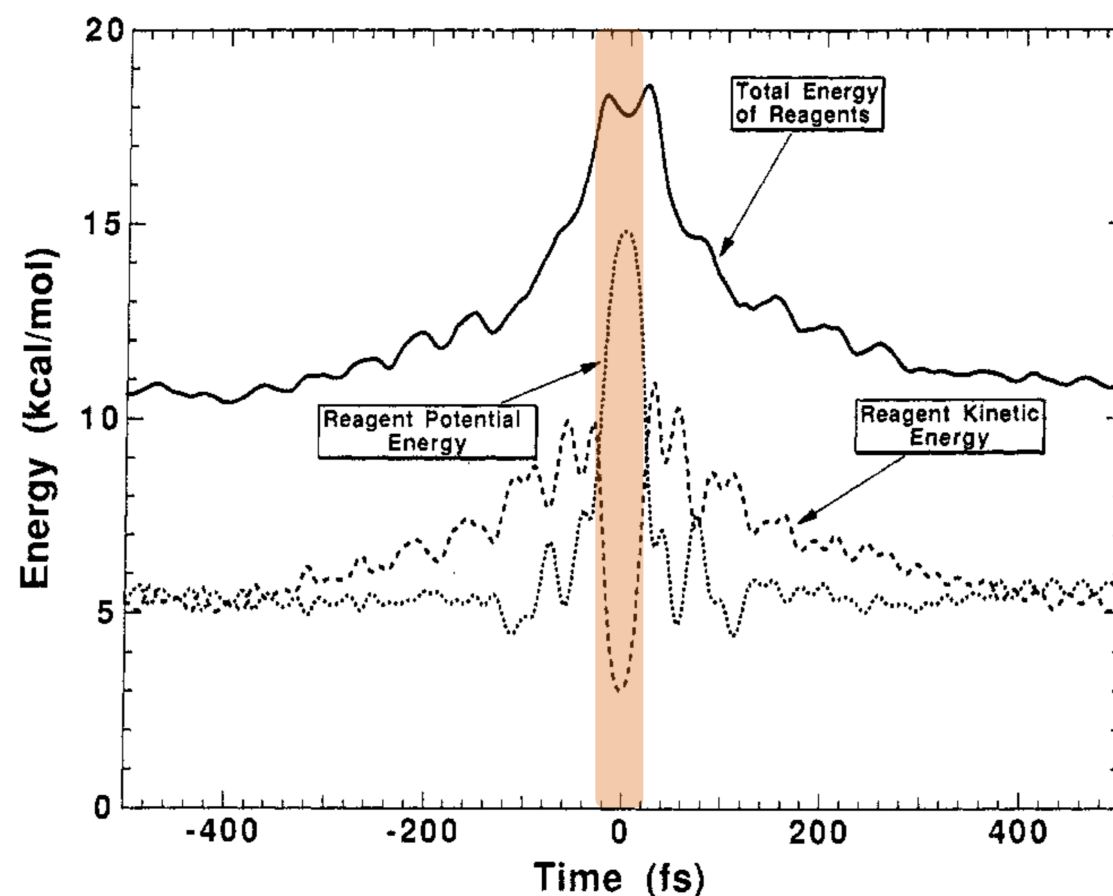
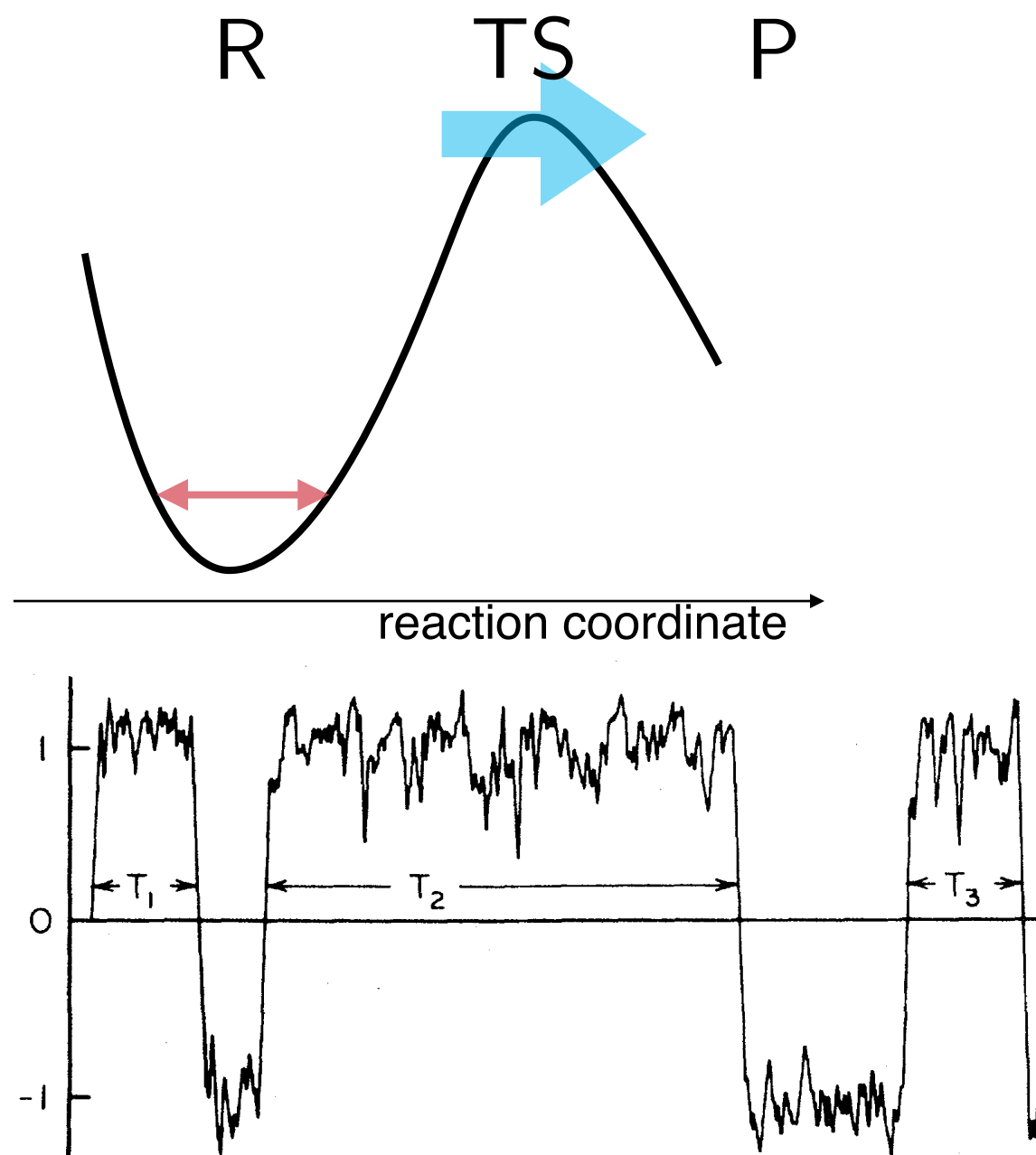
Viggiano et al., *J. Chem. Phys.* **74** 6113 (1981)
Rate constants for the collisional dissociation of N_2O_5 by N_2 .

KJ Leidler, *J. Chem. Ed.* **61** 494 (1984)
The development of the Arrhenius Equation.

From kinetics to dynamics



- (Thermal) Chemical reactions are “rare events”.
- Once a reactive event starts, it proceeds very rapidly.



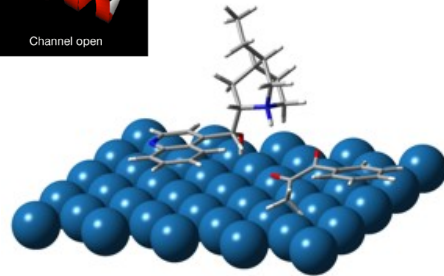
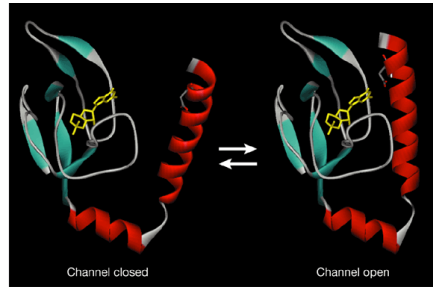
J. Am. Chem. Soc. **1991**, *113*, 74–87

Activation to the Transition State: Reactant and Solvent Energy Flow for a Model S_N2 Reaction in Water

Bradley J. Gertner,[†] Robert M. Whitnell,[†] Kent R. Wilson,^{*,†} and James T. Hynes^{*,‡}

Contribution from the Department of Chemistry, University of California, San Diego, La Jolla, California 92093-0339, and Department of Chemistry and Biochemistry, University of Colorado, Boulder, Colorado 80309-0215. Received April 20, 1990

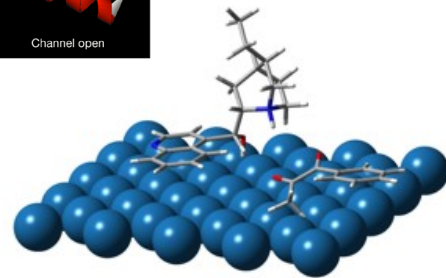
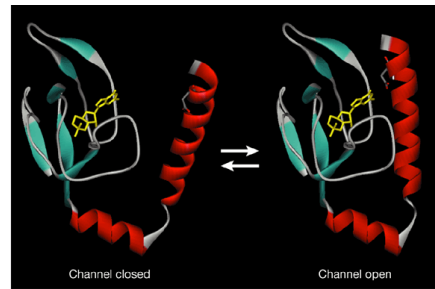
Fundamental reactive steps



Fundamental
reactive steps

- Thermally activated processes
 - A large part of chemistry and biology is here
 - The “trigger” problem

Fundamental reactive steps



Fundamental
reactive steps

- Thermally activated processes
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Ultrafast energy transfer to liquid water by THz pulses

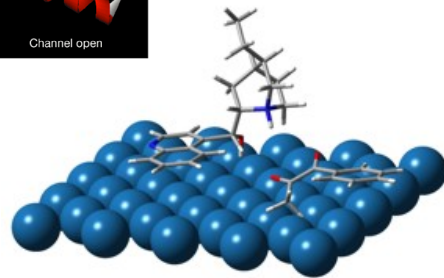
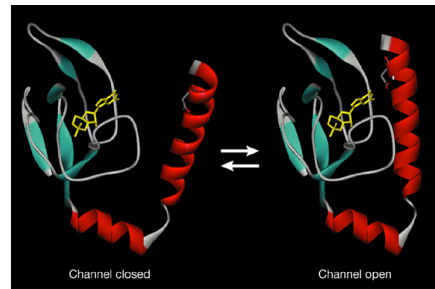
PK Mishra, OV, R Santra

Angew. Chem. Int. Ed. **52**, 13685 (2013)

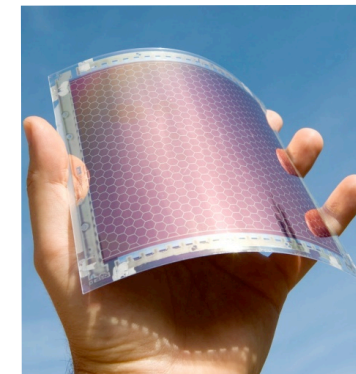
J. Phys. Chem. B **119**, 8080 (2015)

Phys. Rev. E **93**, 032124 (2016)

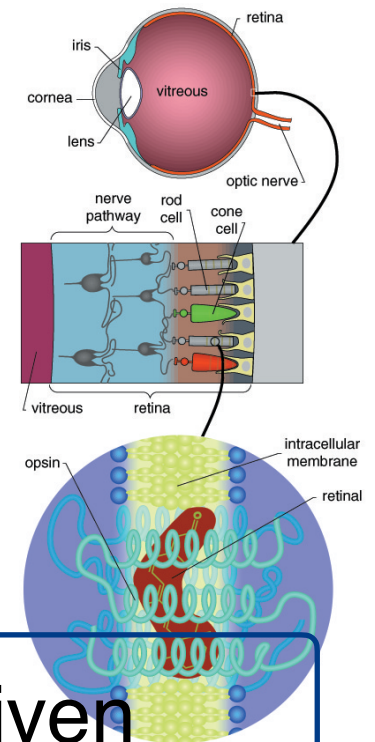
Fundamental reactive steps



Fundamental reactive steps



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■ Thermally activated processes

- A large part of chemistry and biology is here
- The “trigger” problem

Ultrafast energy transfer to liquid water by THz pulses

PK Mishra, OV, R Santra

Angew. Chem. Int. Ed. **52**, 13685 (2013)

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Phys. Rev. E **93**, 032124 (2016)

■ Light activated/driven processes

- Photophysics/photochemistry
- Light to energy conversion & charge separation
- Pump-probe ultrafast spectroscopies

QUESTION



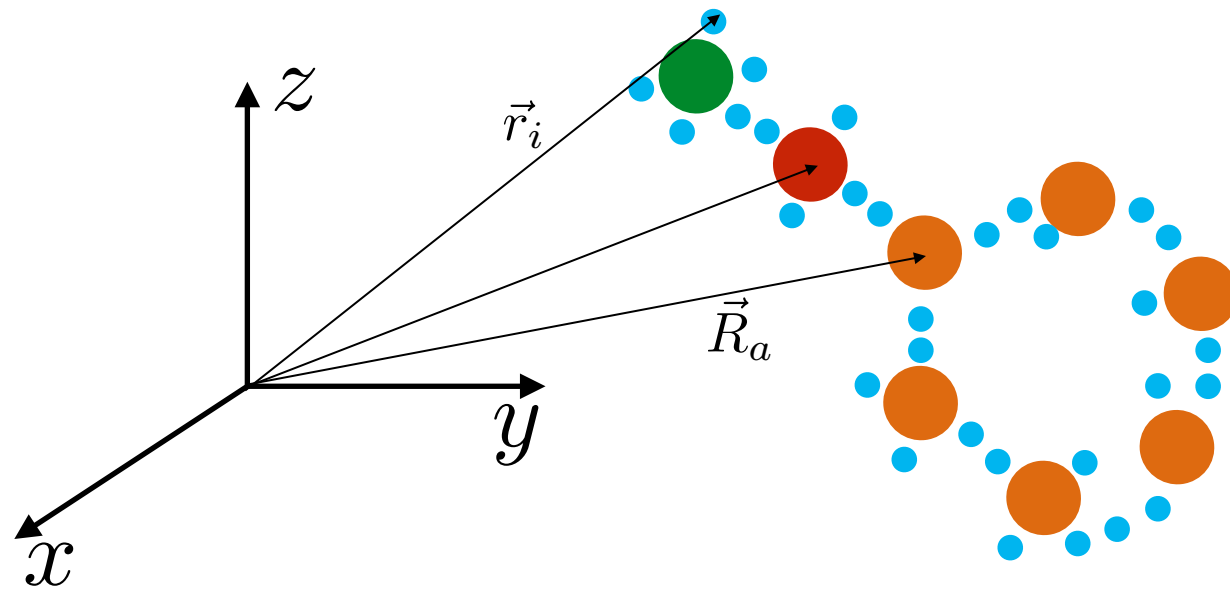
It is possible to define rates for photo-triggered processes:

-  Always.
-  Never.
-  It depends.



Theoretical Description of the Structure and Dynamics of Molecules

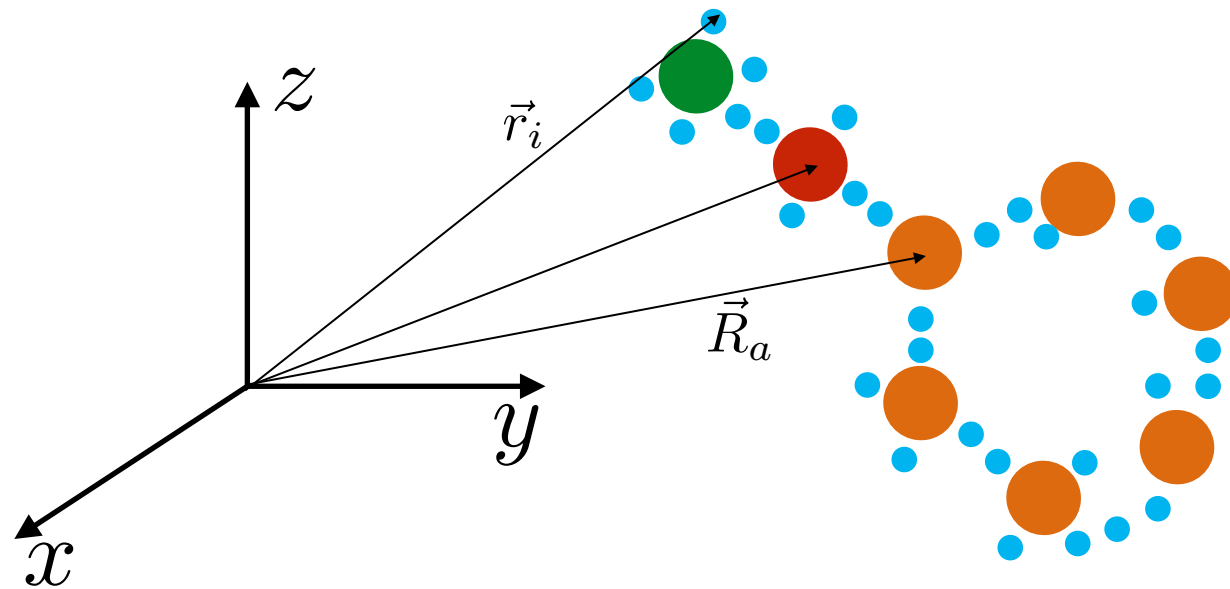
The molecular Hamiltonian



$$\hat{H}(\mathbf{r}, \mathbf{R}, t) = \sum_a \left[-\frac{\hbar^2 \vec{\nabla}_a^2}{2M_a} + \sum_{a>b} \frac{e^2 Z_a Z_b}{|\vec{R}_a - \vec{R}_b|} \right] + \sum_i \left[-\frac{\hbar^2 \vec{\nabla}_i^2}{2m} + \sum_{j>i} \frac{e^2}{|\vec{r}_i - \vec{r}_j|} \right] - \sum_{ai} \frac{e^2 Z_a}{|\vec{R}_a - \vec{r}_i|} - \vec{E}(t) \left(\sum_a e Z_a \vec{R}_a - \sum_i e \vec{r}_i \right)$$

$$i\hbar \frac{\partial \Psi(\mathbf{r}, \mathbf{R}, t)}{\partial t} = \hat{H} \Psi(\mathbf{r}, \mathbf{R}, t)$$

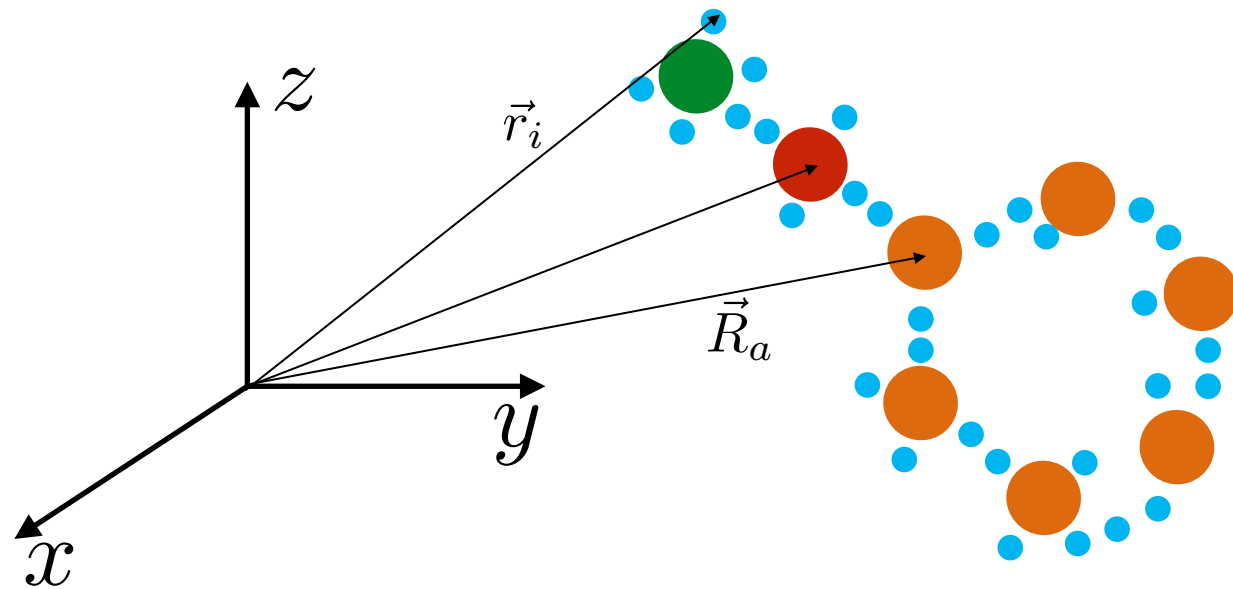
The molecular Hamiltonian



$$\hat{H}(\mathbf{r}, \mathbf{R}, t) = \sum_a -\frac{\hbar^2 \vec{\nabla}_a^2}{2M_a} + \sum_{a>b} \frac{e^2 Z_a Z_b}{|\vec{R}_a - \vec{R}_b|} + \sum_i -\frac{\hbar^2 \vec{\nabla}_i^2}{2m} + \sum_{j>i} \frac{e^2}{|\vec{r}_i - \vec{r}_j|} - \sum_{ai} \frac{e^2 Z_a}{|\vec{R}_a - \vec{r}_i|} - \vec{E}(t) \left(\sum_a e Z_a \vec{R}_a - \sum_i e \vec{r}_i \right)$$

$$i\hbar \frac{\partial \Psi(\mathbf{r}, \mathbf{R}, t)}{\partial t} = \hat{H} \Psi(\mathbf{r}, \mathbf{R}, t)$$

The molecular Hamiltonian



$$\hat{H}(\mathbf{r}, \mathbf{R}, t) = \hat{T}_N + \hat{H}_e - \vec{E}(t) \hat{\mu}$$

$$\hat{H}_e = \sum_i -\frac{\hbar^2 \vec{\nabla}_i^2}{2m} + \sum_{j>i} \frac{e^2}{|\vec{r}_i - \vec{r}_j|} + \sum_{ai} \frac{-e^2 Z_a}{|\vec{R}_a - \vec{r}_i|} + \sum_{a>b} \frac{e^2 Z_a Z_b}{|\vec{R}_a - \vec{R}_b|}$$

$$\hat{H}_e \Phi_j(\mathbf{r}; \mathbf{R}) = V_j(\mathbf{R}) \Phi_j(\mathbf{r}; \mathbf{R})$$

The molecular Schrödinger equation



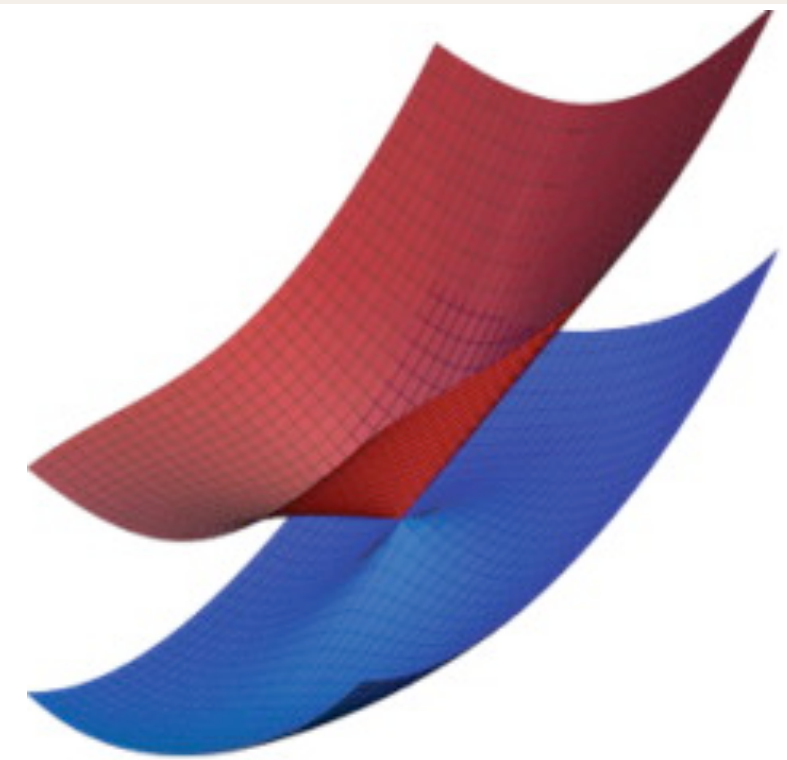
$$\Psi(\mathbf{r}, \mathbf{R}, t) = \sum_i \chi_i(\mathbf{R}, t) \Phi_i(\mathbf{r}; \mathbf{R})$$

Plug into TDSE

$$i\hbar \frac{\partial \chi_j}{\partial t} = (T_N + V_j) \chi_j - \sum_{i \neq j} \Lambda_{ji} \chi_i$$

$$\Lambda_{ji} = \frac{1}{2M} (2\langle \Phi_j | \nabla \Phi_i \rangle \nabla + \langle \Phi_j | \nabla^2 \Phi_i \rangle)$$

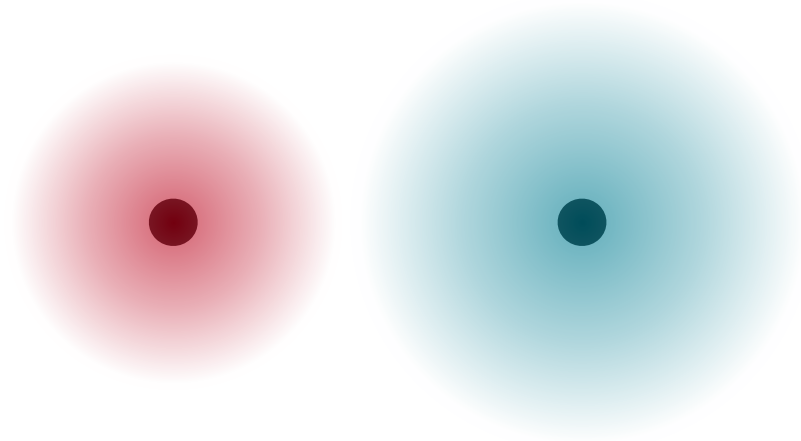
$$\langle \Phi_j | \nabla \Phi_i \rangle = \frac{\langle \Phi_j | (\nabla \hat{H}_{el}) | \Phi_i \rangle}{V_i - V_j}$$



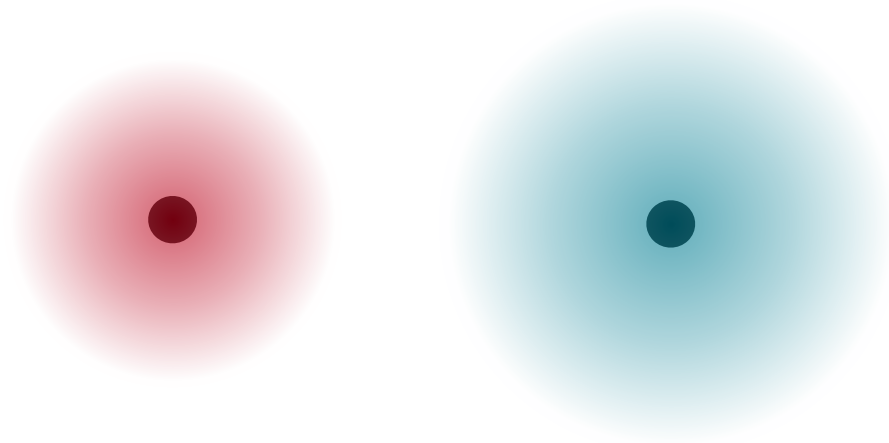
- Photo-acidity
- DNA photo-protection
- ...

In **practical applications** it is often advantageous to use a **diabatic** representation.

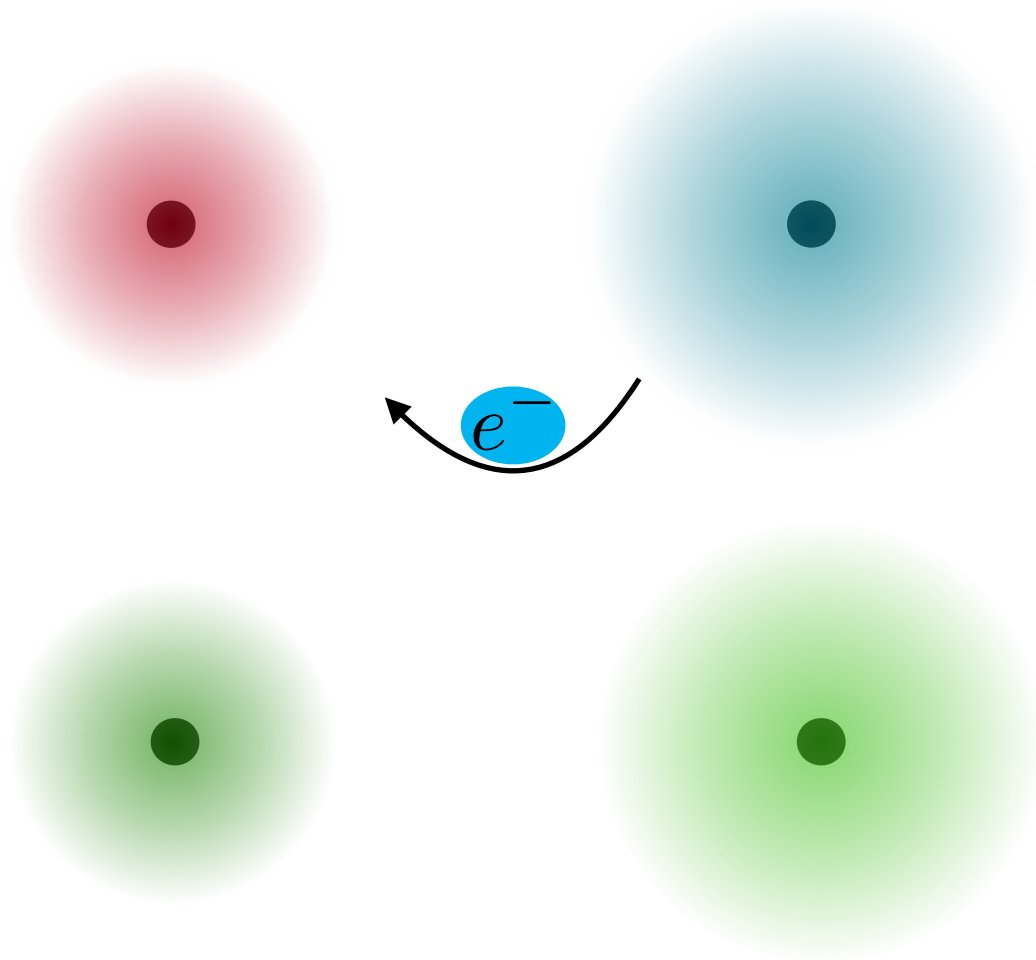
Electron-nuclei coupling and correlation



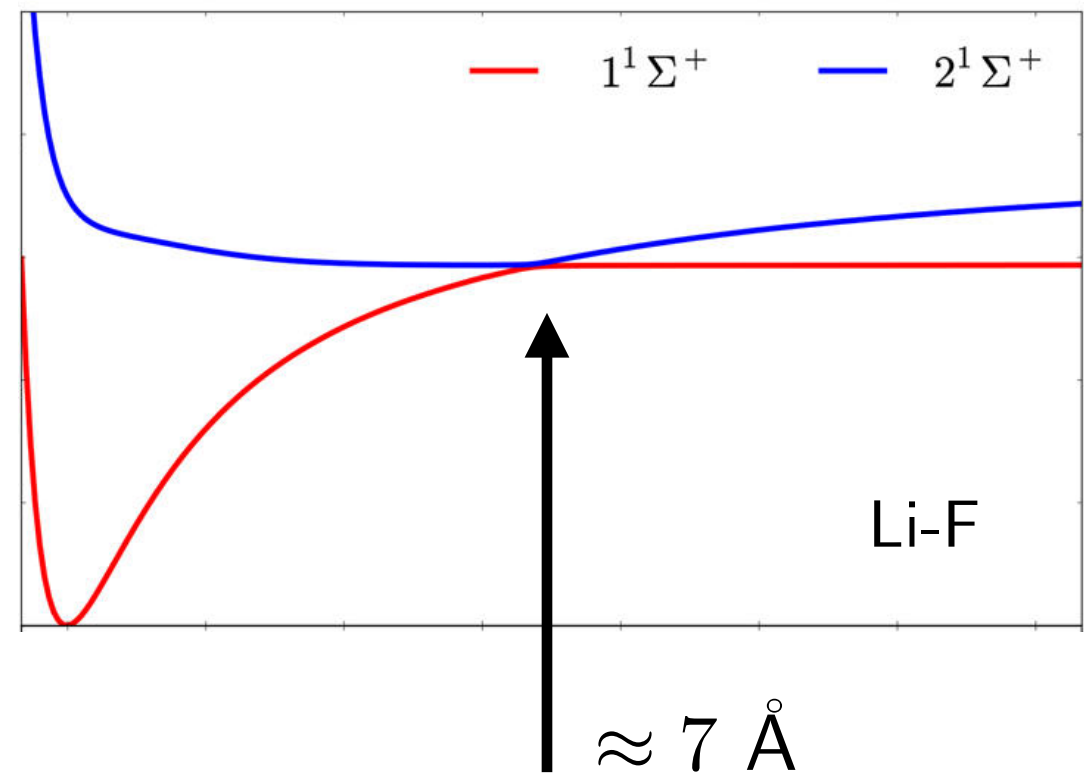
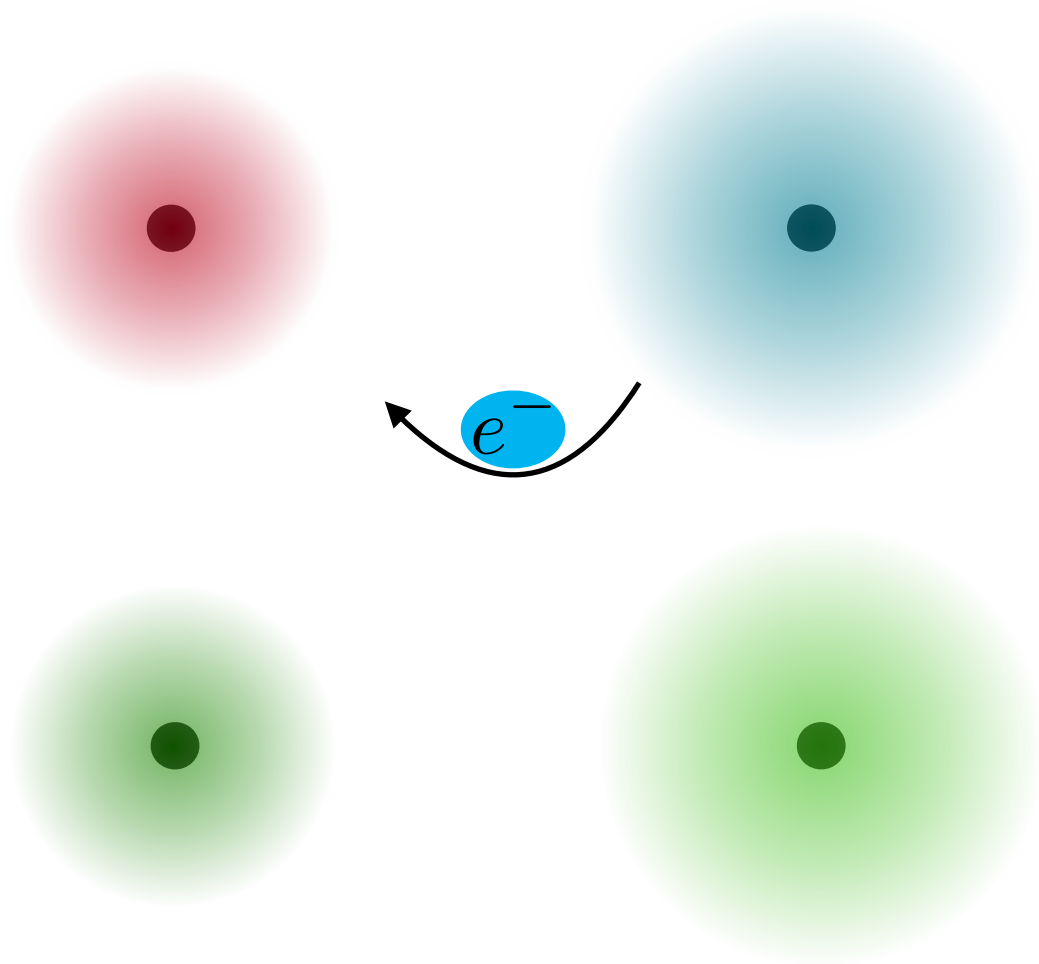
Electron-nuclei coupling and correlation



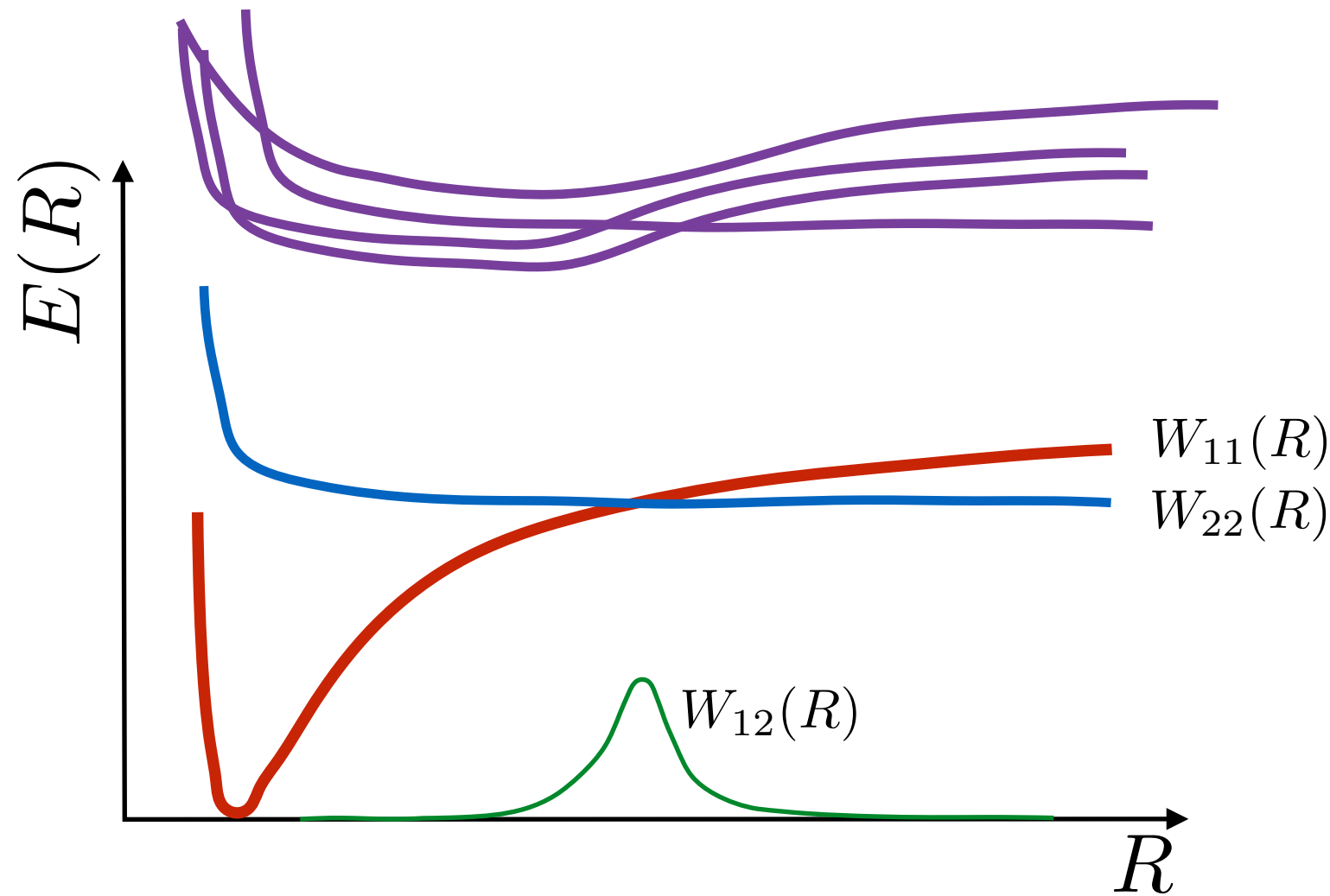
Electron-nuclei coupling and correlation



Electron-nuclei coupling and correlation



The group Born-Oppenheimer approximation



Non-adiabatic transitions to states outside the window of interest are neglected:

$$\Lambda_{1(2),k} \approx 0; \quad k \neq 1, 2$$

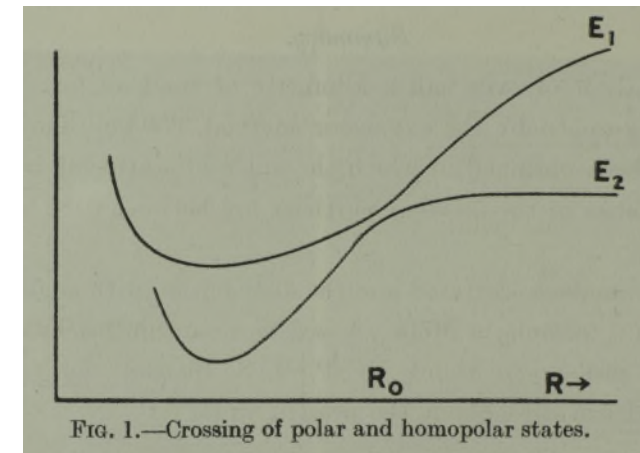
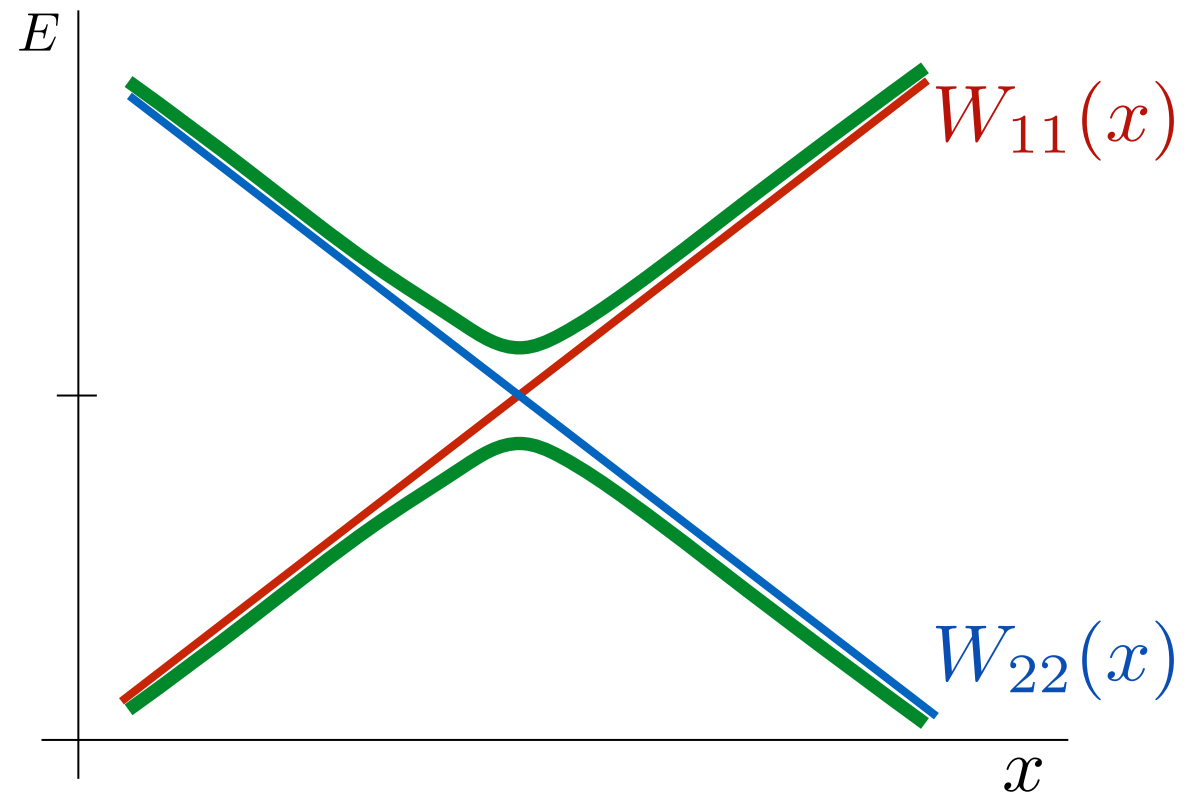
$$i\hbar \frac{\partial}{\partial t} \begin{pmatrix} \tilde{\chi}_1 \\ \tilde{\chi}_2 \end{pmatrix} = \left[\hat{T}_N \otimes \mathbf{1} + \begin{pmatrix} \hat{W}_{11} & \hat{W}_{12} \\ \hat{W}_{12} & \hat{W}_{22} \end{pmatrix} \right] \begin{pmatrix} \tilde{\chi}_1 \\ \tilde{\chi}_2 \end{pmatrix}$$



Why is the BO approximation so useful?

- Because it decouples electronic and nuclear dynamics.
- Because electrons and nuclei are treated on the same footing.
- ▲ One can use it to draw Lewis structures.
- ◆ Because when it breaks one can publish papers on it.

The Landau-Zener model for non-adiabatic transitions



$$\left. \begin{aligned} F_{12} &= \frac{dW_{11}}{dx} - \frac{dW_{22}}{dx} = \text{const.} \\ v &= \dot{x} = \text{const.} \end{aligned} \right\} \begin{aligned} \frac{d}{dt}(W_{11} - W_{22}) &= F_{12} v \\ &= \alpha \end{aligned}$$

The energy gap changes linearly with time.

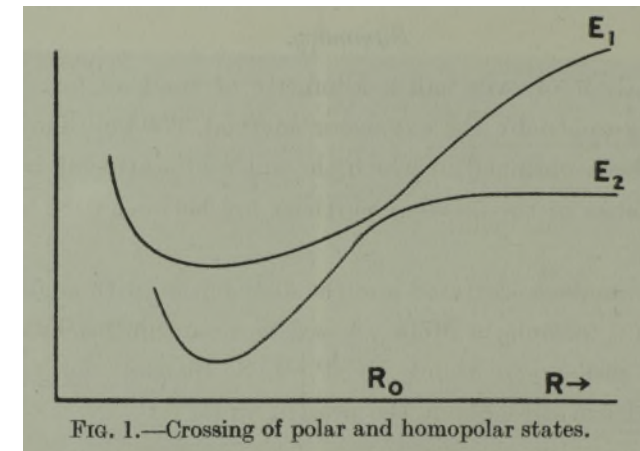
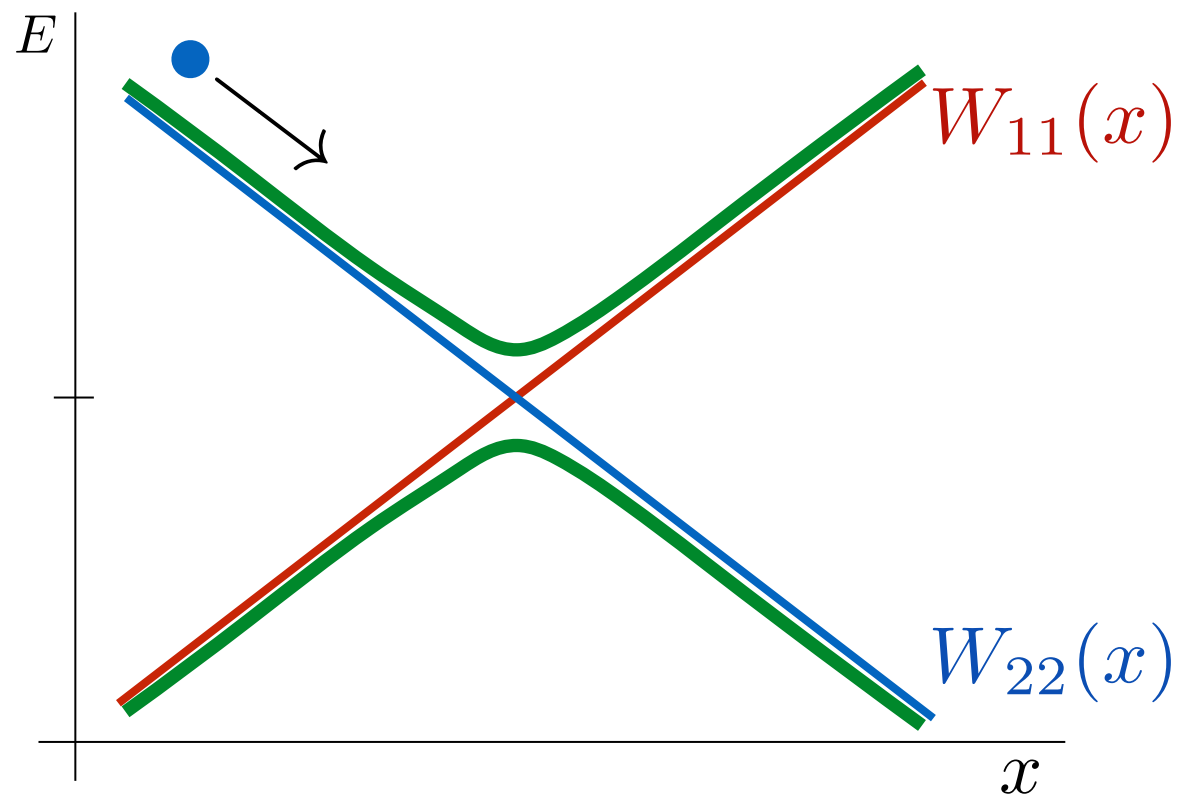
$$W_{12} = \text{const.}$$

Zener, C. *Proc. R. Soc. London A* **1932**, 137, 696.

Landau, L. D. *Phys. Z.* **1932**, 2, 46.

Wittig, C. *J. Phys. Chem. B* **2005**, 109, 8428

The Landau-Zener model for non-adiabatic transitions



$$C_1(-\infty) = 0$$

$$|C_2(-\infty)| = 1$$

$$P_{\text{nat}} = |C_2(\infty)|^2$$

$$P_{\text{nat}} = \exp(-2\pi \omega_{12} \tau_d)$$

$$\omega_{12} = \frac{|W_{12}|}{\hbar} \quad \tau_d = \frac{|W_{12}|}{v|F_{12}|}$$

$$\mathbf{W}(t) = \begin{pmatrix} \alpha t/2 & W_{12} \\ W_{12} & -\alpha t/2 \end{pmatrix}$$

$$\dot{C}_1(t) = W_{12}C_2(t) \exp\left(-i/\hbar \int^t \alpha t' dt'\right)$$

$$\dot{C}_2(t) = W_{12}C_1(t) \exp\left(i/\hbar \int^t \alpha t' dt'\right)$$

Zener, C. *Proc. R. Soc. London A* **1932**, 137, 696.

Landau, L. D. *Phys. Z.* **1932**, 2, 46.

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Fast atomic motion results in...

- ... non-adiabatic transitions.
- ... small probability that the electronic subsystem rearranges.
- ▲ ... break-up of the BO approximation.
- ◆ ... that electrons cannot follow.

Curve crossings: non-crossing rule in diatoms

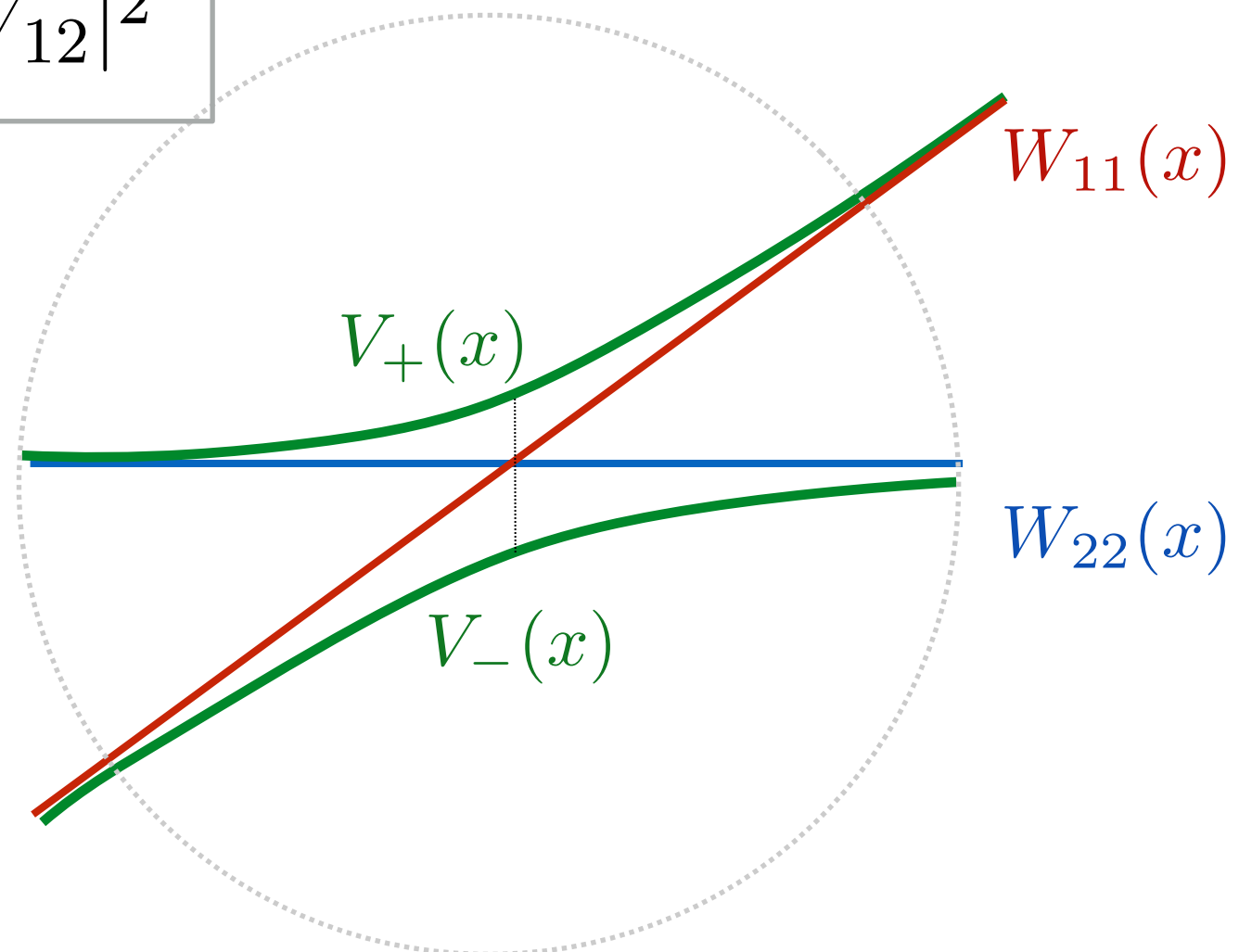
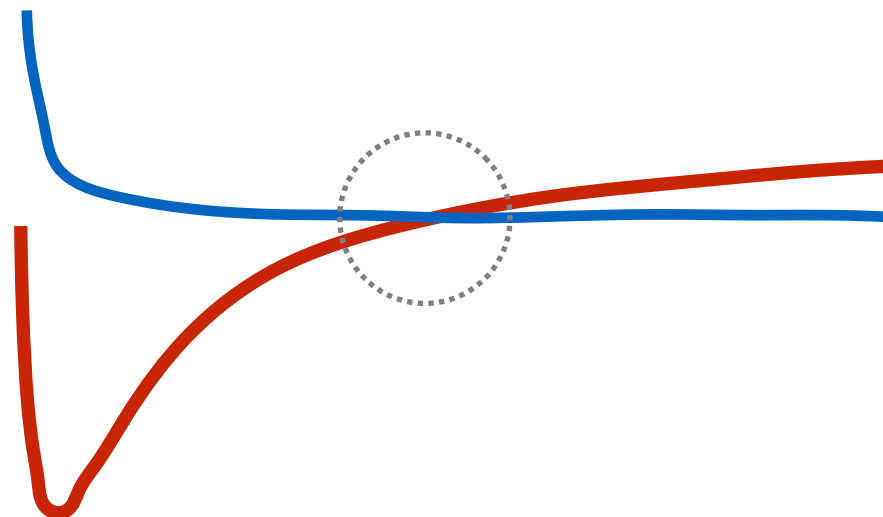


$$\mathbf{W} = \begin{pmatrix} W_{11} & W_{12} \\ W_{21} & W_{22} \end{pmatrix}$$

$$W_{ij} = \langle \Phi_i^{(d)} | H_{el} | \Phi_j^{(d)} \rangle$$

$$V_{\pm} = \frac{W_{11} + W_{22}}{2} \pm \frac{1}{2} \sqrt{(W_{11} - W_{22})^2 + 4|W_{12}|^2}$$

$$\Delta V = \sqrt{(W_{11} - W_{22})^2 + 4|W_{12}|^2}$$

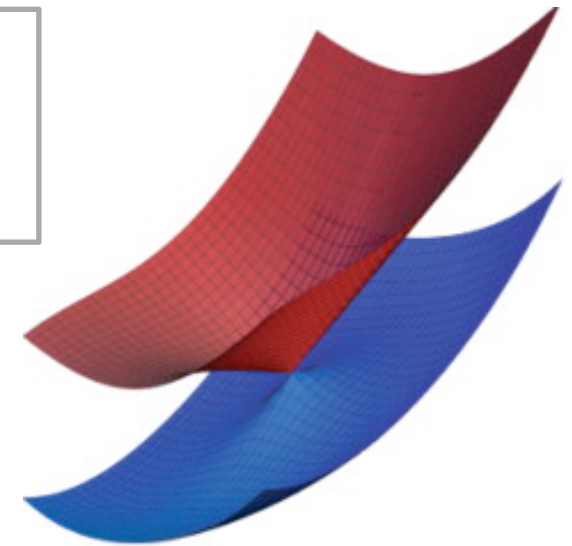


Curve crossings: conical intersections



$$\mathbf{W} = \begin{pmatrix} W_{11} & W_{12} \\ W_{21} & W_{22} \end{pmatrix}$$

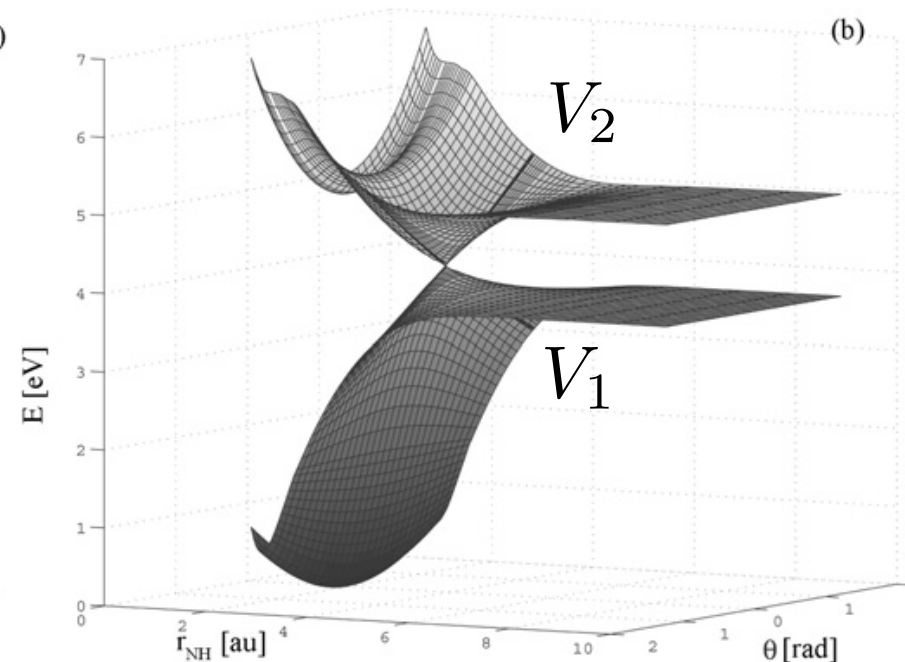
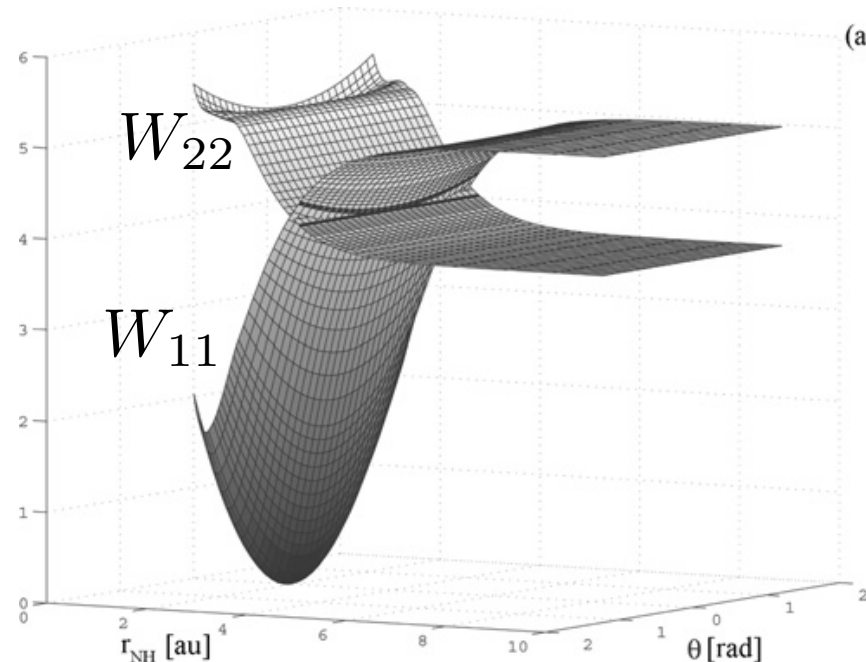
$$W_{ij} = \langle \Phi_i^{(d)} | H_{el} | \Phi_j^{(d)} \rangle$$



$$V_{\pm} = \frac{W_{11} + W_{22}}{2} \pm \frac{1}{2} \sqrt{(W_{11} - W_{22})^2 + 4|W_{12}|^2}$$

$$\Delta V = \sqrt{(W_{11} - W_{22})^2 + 4|W_{12}|^2}$$

Two conditions to satisfy: CI have dimensionality M-2



Adapted from *Faraday Discuss.* 2004, **127**, 283-293

Numerically solving the dynamics problem



$$\begin{aligned}\Psi(\mathbf{R}, t) &= \sum_J A_J(t) \Phi_J(\mathbf{R}) \\ &= \sum_{j_1, \dots, j_f}^{N_1, \dots, N_f} A_{j_1, \dots, j_f} \varphi_{j_1}(R_1) \varphi_{j_2}(R_2) \cdots \varphi_{j_f}(R_f)\end{aligned}$$

N^f configurations

Standard method

$$i\hbar \dot{A}_I = \sum_J \langle \varphi_I(\mathbf{R}) | T + V(\mathbf{R}) | \varphi_J(\mathbf{R}) \rangle A_J$$

Time-dependent configurations:

- MCTDH
- Moving Gaussian basis

R. Kossloff, *J. Phys. Chem.* **92**, 2087 (1988)

C. Leforestier et al., *J. Comp. Phys.* **94**, 59 (1991)

M.H. Beck et al. *Phys. Rep.* **324**, 1, (2000)

M. Ben-Nun, T.J. Martinez, *Adv. Chem. Phys.* **121**, 439 (2002)

Worth et al. *J. Phys. Chem A* **107**, 621 (2003);

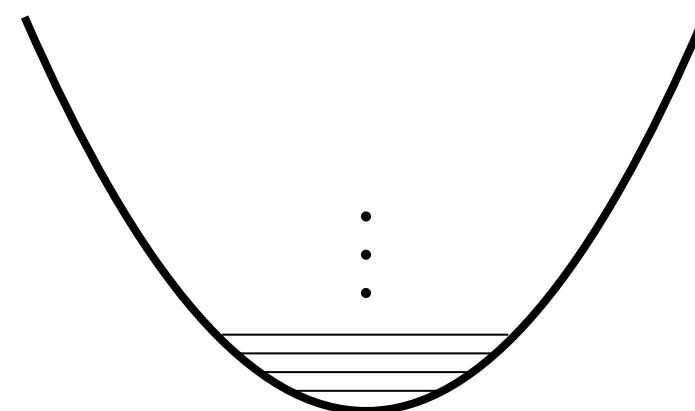
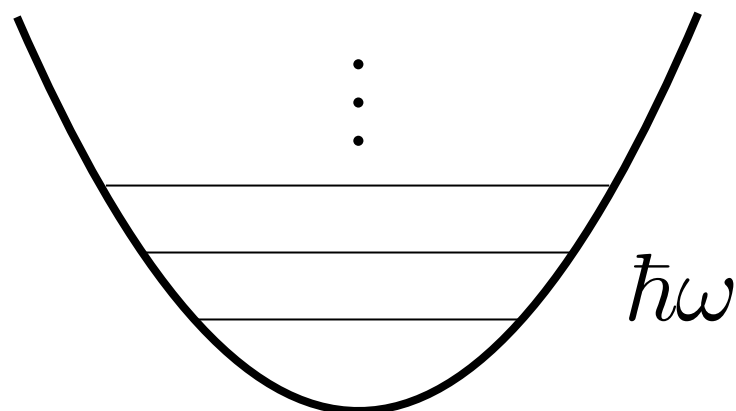
Nuclei can approximately be described classically ...



... when the energy scale of the problem is much larger than the nuclear energy level separations.

$$\lambda = \frac{h}{mv}$$

$$\frac{1}{2}kx^2 \quad \omega = \sqrt{\frac{k}{m}}$$



Numerically solving the dynamics problem



Nuclear dynamics described classically:

Born-Oppenheimer

$$m_a \ddot{\vec{R}}_a = -\vec{\nabla}_a V_0(\mathbf{R}) \quad \left\{ \begin{array}{ll} V_0(\mathbf{R}) = \langle \psi_{0;\mathbf{R}} | \hat{H}_e(\mathbf{R}(t)) | \psi_{0;\mathbf{R}} \rangle & \text{BOMD} \\ V_0(\mathbf{R}) = f(\mathbf{R}) = \sum_{\text{bonds}} + \sum_{\text{angles}} + \sum_{\text{charges}} + \dots & \text{MM (mol. mech.)} \end{array} \right.$$

Non-adiabatic transitions

$$m_a \ddot{\vec{R}}_a = -\vec{\nabla}_a V_x(\mathbf{R}) \quad \left\{ \begin{array}{ll} V_x(\mathbf{R}) = \langle \psi(t) | \hat{H}_e(\mathbf{R}(t)) | \psi(t) \rangle & \text{Ehrenfest} \\ V_x(\mathbf{R}) = V_j(\mathbf{R}) & \text{Surface-Hopping} \end{array} \right.$$

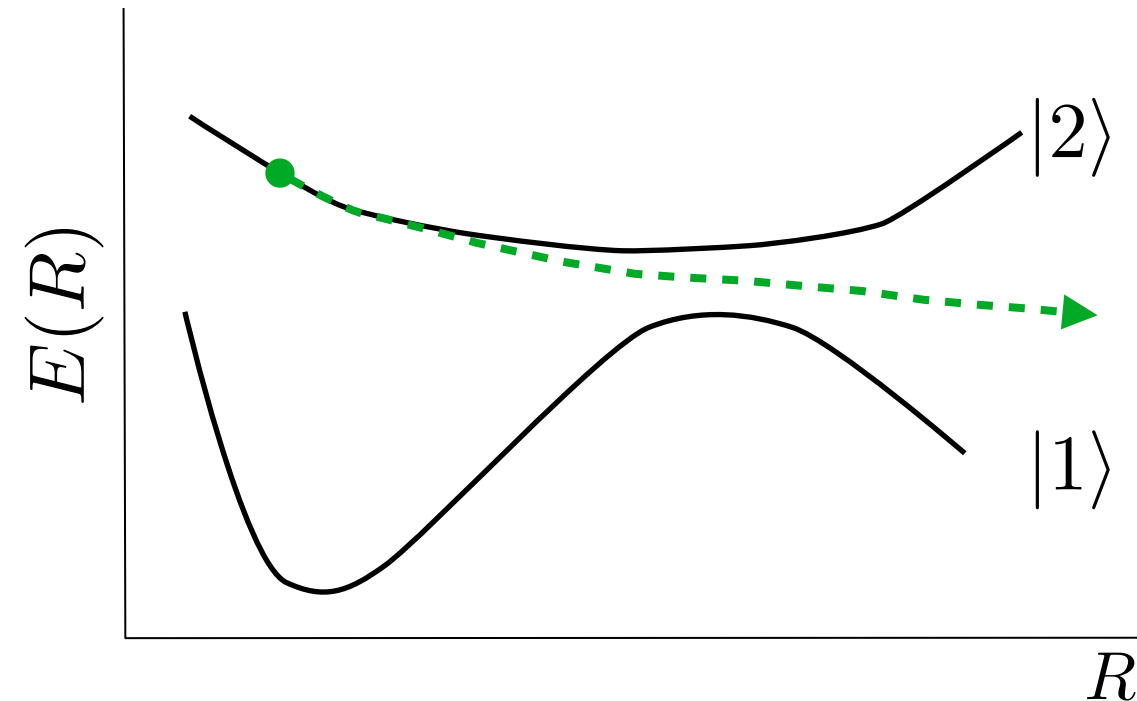
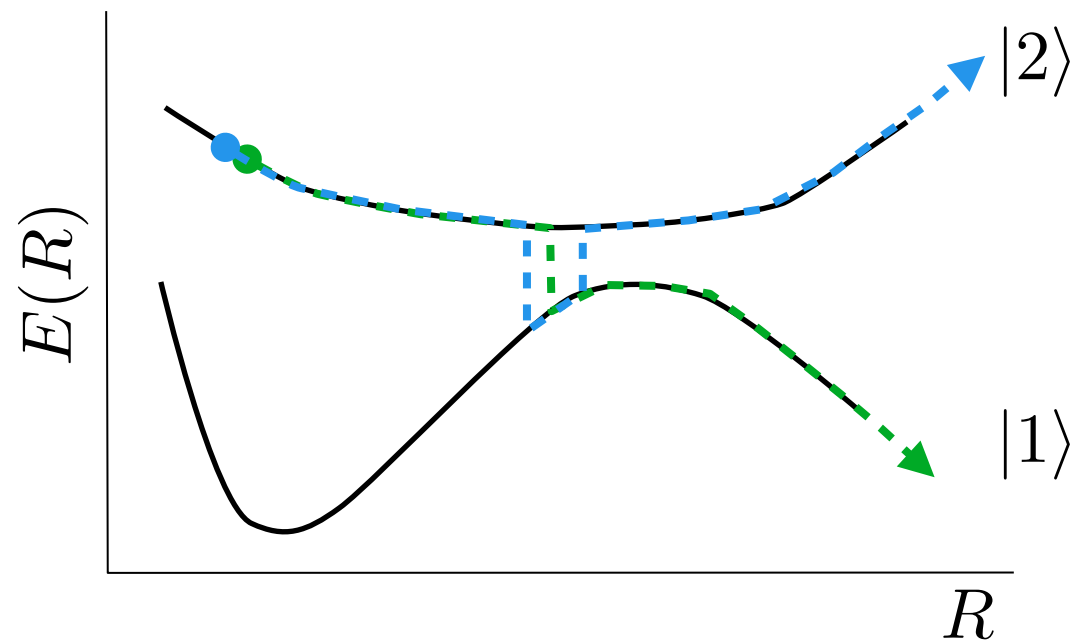
$$i\hbar |\dot{\psi}(t)\rangle = \hat{H}_e(\mathbf{R}(t)) |\psi(t)\rangle$$

$$\hat{H}_e(\mathbf{R}) |\psi_j; \mathbf{R}\rangle = V_j(\mathbf{R}) |\psi_j; \mathbf{R}\rangle$$

J. Tully, *J. Chem. Phys.* **93**, 1061 (1990)

M.D. Hack, D.G. Truhlar, *J. Phys. Chem. A* **104**, 7917, (2000)

Non-adiabatic dynamics with classical trajectories



$$m_a \ddot{\vec{R}}_a = -\vec{\nabla}_a V_j(\mathbf{R})$$

$$i\hbar \dot{c}_j(t) = \sum_i c_i(t) \left(V_i \delta_{ji} - i\hbar \dot{\mathbf{R}} \cdot \mathbf{d}_{ji} \right)$$

$$m_a \ddot{\vec{R}}_a = -\vec{\nabla}_a V_x(\mathbf{R})$$

$$V_x(\mathbf{R}) = \langle \psi(t) | \hat{H}_e(\mathbf{R}(t)) | \psi(t) \rangle$$

$$i\hbar |\dot{\psi}(t)\rangle = \hat{H}_e(\mathbf{R}(t)) |\psi(t)\rangle$$

J. Tully, *J. Chem. Phys.* **93**, 1061 (1990)

M.D. Hack, D.G. Truhlar, *J. Phys. Chem. A* **104**, 7917, (2000)



Nuclei can (sometimes) be approximated as classical particles when...

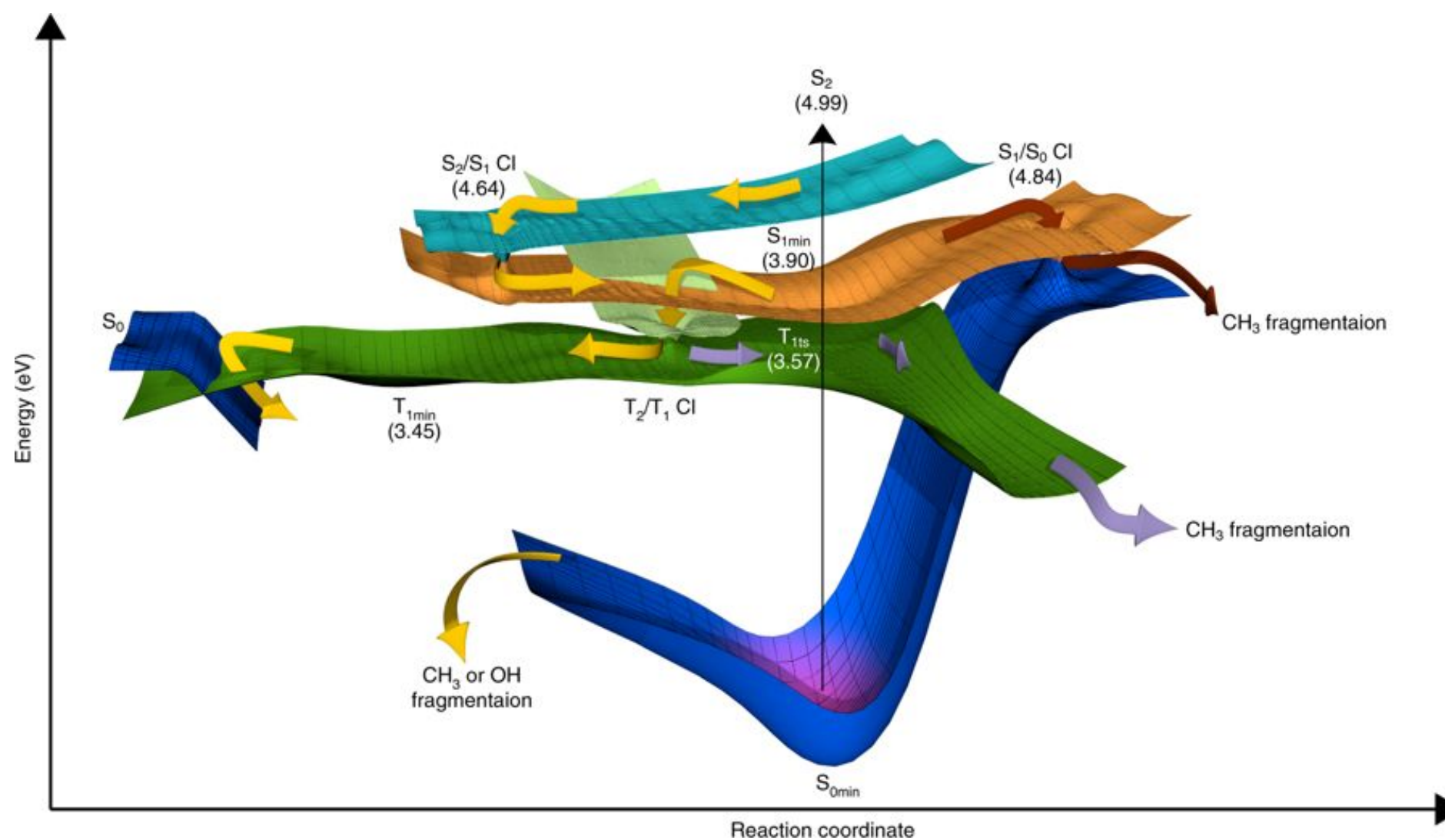
- ...electrons are fast.
- ...electrons are slow.
- ▲ ...nuclei have a large kinetic energy.
- ◆ ...nuclei don't move.



Time-dependent Description of Molecular Spectroscopy

*Nearly all we know about matter is through
its interaction with light.*

— Yoda.



Acetylacetone photodynamics at a seeded free-electron laser
R. J. Squibb et. al.
Nature Communications, **9**, 63 (2018)

Non-perturbative absorption spectrum



$$\sigma(\omega) = \frac{\# \text{ of photons absorbed at freq. } \omega}{\# \text{ of photons per unit area at freq. } \omega} \quad \textcircled{1} \quad \textcircled{2}$$

- ① From the total energy gained (lost) by the system at a given frequency: $N_a(\omega) = \frac{\Delta E(\omega)}{\hbar\omega}$
- ② From classical electromagnetism: $\frac{N(\omega)}{A} = \frac{|\tilde{\epsilon}(\omega)|^2 c}{\hbar\omega}$

$$\sigma(\omega) = \frac{\Delta E(\omega)}{|\tilde{\epsilon}(\omega)|^2 c}$$

$$\text{where } \tilde{\epsilon}(\omega) = \frac{1}{2\pi} \int_{-\infty}^{\infty} \epsilon(t) e^{i\omega t} dt \quad \text{Back FT}$$

So the task is to determine the energy exchanged with the EM field at each frequency.

Non-perturbative absorption spectrum



$$E(t) = \langle \Psi(t) | \hat{H}(t) | \Psi(t) \rangle$$

$$\hat{H} = \hat{H}_0 - \hat{\mu}\epsilon(t)$$

$$\dot{E}(t) = \langle \dot{\Psi}(t) | \hat{H} | \Psi(t) \rangle + \langle \Psi(t) | \hat{H} | \dot{\Psi}(t) \rangle + \langle \Psi(t) | \frac{d\hat{H}}{dt} | \Psi(t) \rangle$$

Note: A red arrow points from the first two terms to a red '0' above the third term, indicating they cancel out.

$$\begin{aligned} \dot{E}(t) &= -\langle \Psi(t) | \hat{\mu} | \Psi(t) \rangle \dot{\epsilon}(t) \\ &= -P(t) \dot{\epsilon}(t) \end{aligned}$$

Total energy increase:

$$\Delta E = \int_{-\infty}^{\infty} \dot{E}(t) dt = \int_{-\infty}^{\infty} -P(t) \dot{\epsilon}(t) dt = -2\pi \int_{-\infty}^{\infty} -i\omega [\tilde{P}(\omega)]^* \tilde{\epsilon}(\omega) d\omega = \int_{-\infty}^{\infty} \Delta E(\omega) d\omega = \Delta E$$

$$\Delta E(\omega) = 2\pi i \omega [\tilde{P}(\omega)]^* \tilde{\epsilon}(\omega)$$

Time-dependent perturbation theory



$$\Psi(t) = \Psi^{(0)}(t) + \lambda \Psi^{(1)}(t) + \lambda^2 \Psi^{(2)}(t) + \dots$$

$$i\hbar \frac{\partial}{\partial t} \Psi^{(0)}(t) = H_0 \Psi^{(0)}(t)$$

$$i\hbar \frac{\partial}{\partial t} \Psi^{(1)}(t) = H_0 \Psi^{(1)}(t) + V(t) \Psi^{(0)}(t)$$

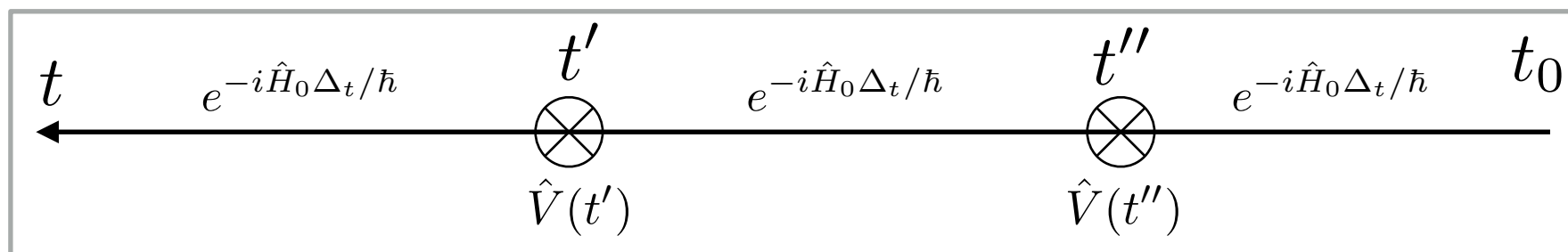
$$i\hbar \frac{\partial}{\partial t} \Psi^{(2)}(t) = H_0 \Psi^{(2)}(t) + V(t) \Psi^{(1)}(t)$$

⋮

$$\Psi^{(0)}(t) = e^{-i\hat{H}_0(t-t_0)/\hbar} \Psi^{(0)}(t_0)$$

$$\Psi^{(1)}(t) = \frac{1}{i\hbar} \int_{t_0}^t dt' e^{-i\hat{H}_0(t-t')/\hbar} \hat{V}(t') e^{-i\hat{H}_0(t'-t_0)/\hbar} \Psi^{(0)}(t_0)$$

$$\Psi^{(2)}(t) = \frac{1}{(i\hbar)^2} \int_{t_0}^t dt' \int_{t_0}^{t'} dt'' e^{-i\hat{H}_0(t-t')/\hbar} \hat{V}(t') e^{-i\hat{H}_0(t'-t'')/\hbar} \hat{V}(t'') e^{-i\hat{H}_0(t''-t_0)/\hbar} \Psi^{(0)}(t_0)$$



The case of a very short pulse



$$\epsilon(t) = a \delta(t) \longrightarrow \tilde{\epsilon}(\omega) = a/2\pi$$

Ideal short pulse.

$$P(t) = \langle \Psi(t) | \hat{\mu} | \Psi(t) \rangle = \left(\langle \Psi^{(0)}(t) | + \langle \Psi^{(1)}(t) | \right) \hat{\mu} \left(| \Psi^{(0)}(t) \rangle + | \Psi^{(1)}(t) \rangle \right)$$

$$| \Psi^{(1)}(t) \rangle = \frac{a}{i\hbar} e^{-i\hat{H}_0 t/\hbar} \hat{\mu} | \Psi^{(0)}(0) \rangle$$

First order TDPT

$$P(t) = 2\Re \left\{ \langle \Psi^{(0)}(t) | \hat{\mu} | \Psi^{(1)}(t) \rangle \right\} = 2\Re \left\{ \frac{1}{i\hbar} e^{iE_0 t/\hbar} \langle \Psi^{(0)}(0) | \hat{\mu} e^{-i\hat{H}_0 t/\hbar} \hat{\mu} | \Psi^{(0)}(0) \rangle \right\} \Theta(0)$$

$$\begin{aligned} \tilde{P}(\omega) &= \frac{1}{2\pi} \int_{-\infty}^{\infty} P(t) e^{i\omega t} dt \\ &= \frac{1}{2\pi i\hbar} \int_{-\infty}^{\infty} \langle \Psi^{(0)}(0) | \hat{\mu} e^{-i\hat{H}_0 t/\hbar} \hat{\mu} | \Psi^{(0)}(0) \rangle e^{i(\omega_0 + \omega)t} dt \end{aligned}$$

$$\sigma(\omega) = \frac{2\pi\omega}{\hbar c} \int_{-\infty}^{\infty} C(t) e^{i(\omega_0 + \omega)t} dt$$

Wavepacket interpretation of 1-photon spectroscopy

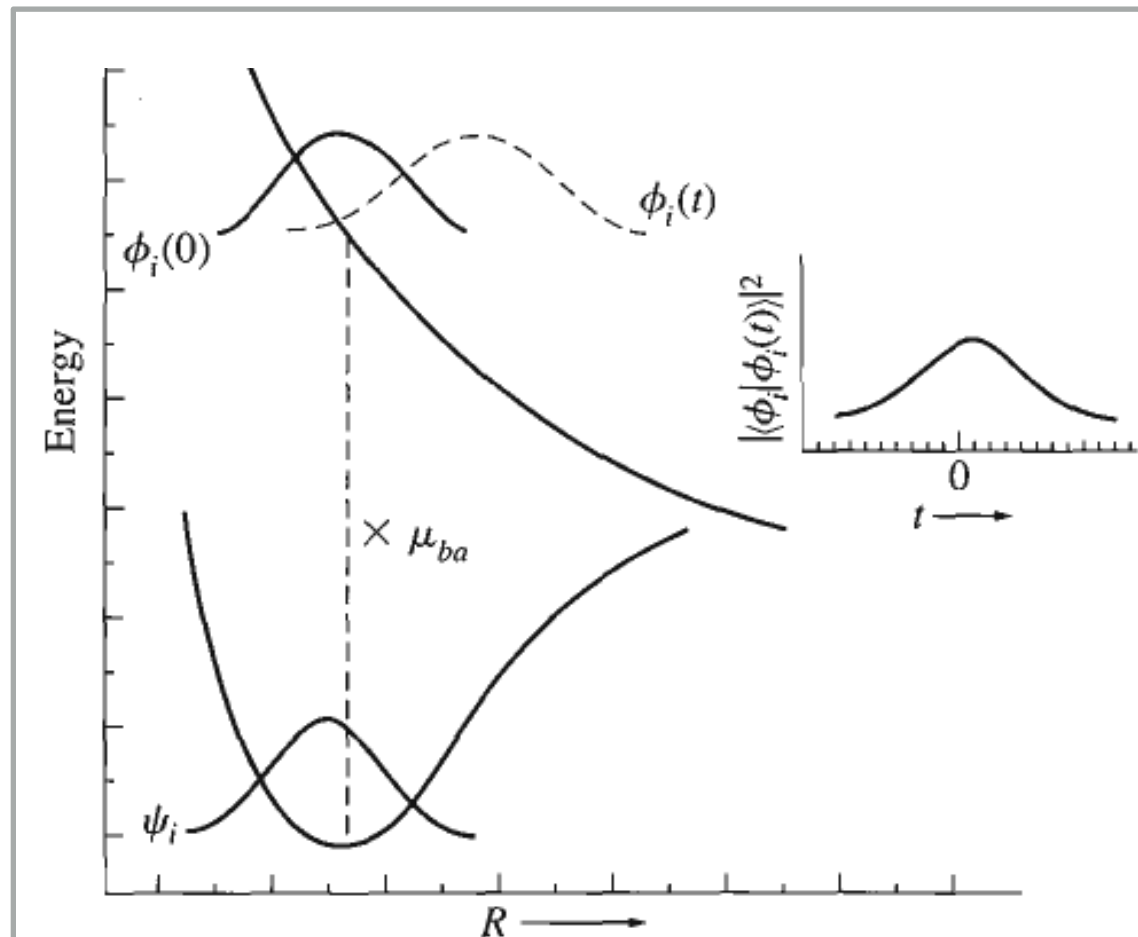
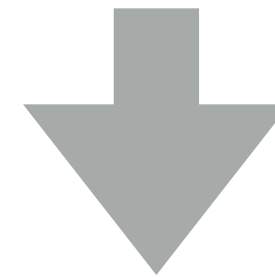


Figure 14.6 Wavepacket interpretation of the formula for the one-photon absorption spectrum, $\sigma(\omega_1)$ (Eq. 14.21). The state $\phi_i = \mu_{ba}\psi_i$ is a *wavepacket* on the excited electronic state potential and begins to move according to the Time-Dependent Schrödinger Equation. The absolute value squared of the autocorrelation function constructed from the overlap of $\phi_i(t)$ with $\phi_i(0)$ is shown in the inset. The Fourier transform of the autocorrelation function at frequency $\tilde{\omega}_1$ determines the absorption spectrum (see Eq. 14.21).

$$\sigma(\omega) \propto \omega \sum_j |\langle \Psi_j | \hat{\mu} | \Psi_0 \rangle| \delta(\omega + \omega_0 - \omega_j)$$



$$\begin{aligned} \sigma(\omega) &\propto \omega \int_{-\infty}^{+\infty} \langle \Psi_0 | \hat{\mu} e^{-\frac{i}{\hbar} \hat{H}_0 t / \hbar} \hat{\mu} | \Psi_0 \rangle e^{i\omega t} dt \\ &\propto \omega 2\Re \left\{ \int_0^{+\infty} \langle \Psi_0 | \hat{\mu} e^{-\frac{i}{\hbar} \hat{H}_0 t / \hbar} \hat{\mu} | \Psi_0 \rangle e^{i\omega t} dt \right\} \end{aligned}$$

$C(t)$

Wavepacket interpretation of 1-photon spectroscopy

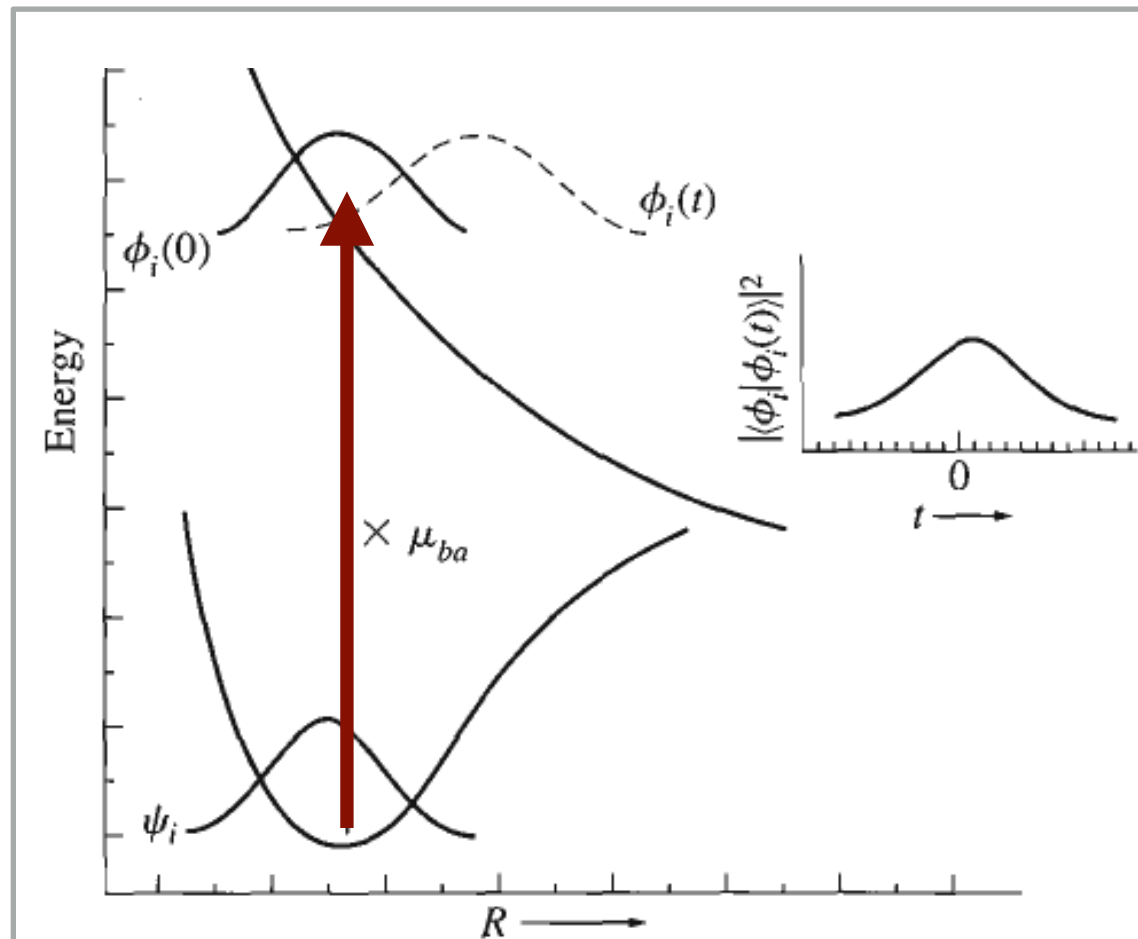
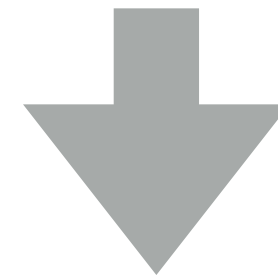


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$$\begin{aligned} \sigma(\omega) &\propto \omega \int_{-\infty}^{+\infty} \langle \Psi_0 | \hat{\mu} e^{-\frac{i}{\hbar} \hat{H}_0 t / \hbar} \hat{\mu} | \Psi_0 \rangle e^{i\omega t} dt \\ &\propto \omega 2\Re \left\{ \int_0^{+\infty} \underbrace{\langle \Psi_0 | \hat{\mu} e^{-\frac{i}{\hbar} \hat{H}_0 t / \hbar} \hat{\mu} | \Psi_0 \rangle}_{C(t)} e^{i\omega t} dt \right\} \end{aligned}$$

Wavepacket interpretation of 1-photon spectroscopy

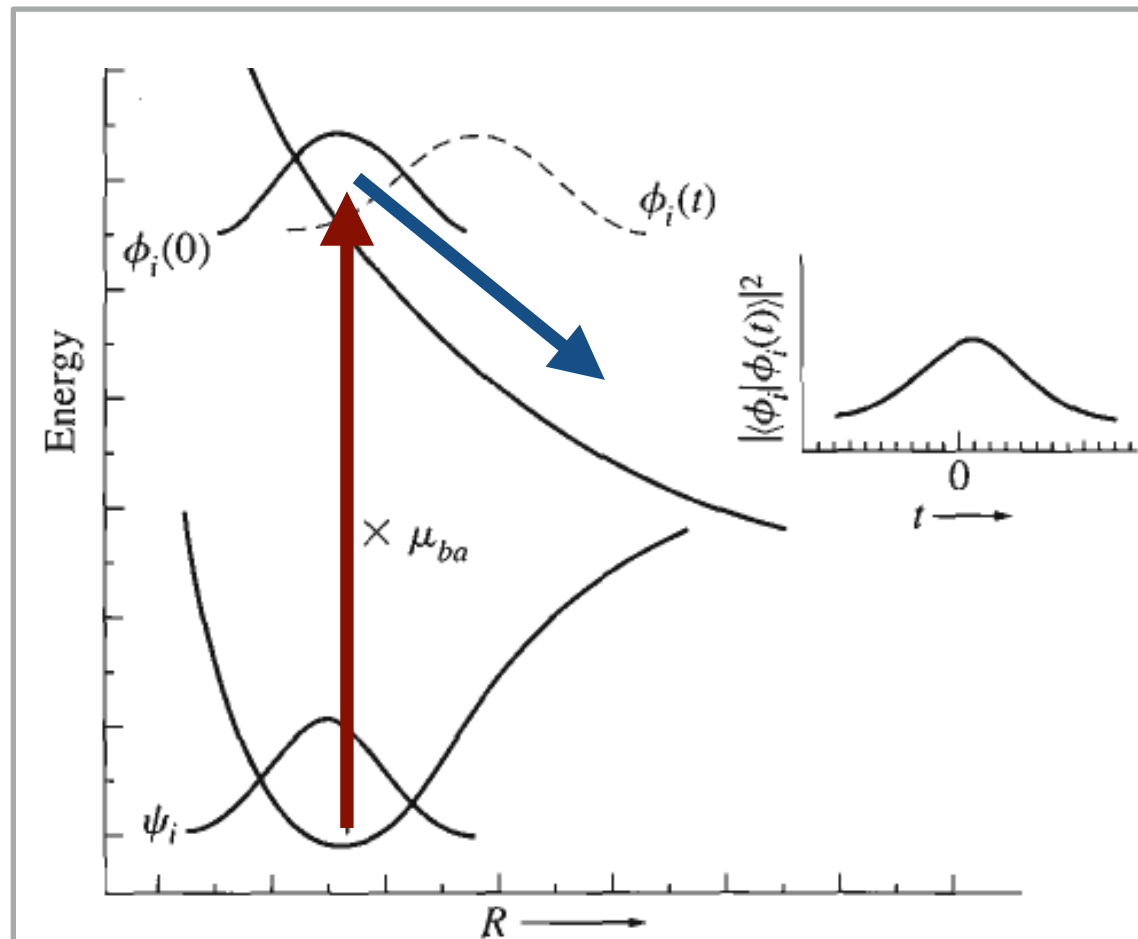
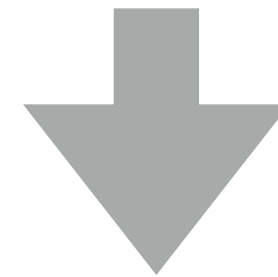


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$$\begin{aligned} \sigma(\omega) &\propto \omega \int_{-\infty}^{+\infty} \langle \Psi_0 | \hat{\mu} e^{-\frac{i}{\hbar} \hat{H}_0 t / \hbar} \hat{\mu} | \Psi_0 \rangle e^{i\omega t} dt \\ &\propto \omega 2\Re \left\{ \int_0^{+\infty} \underbrace{\langle \Psi_0 | \hat{\mu} e^{-\frac{i}{\hbar} \hat{H}_0 t / \hbar} \hat{\mu} | \Psi_0 \rangle}_{C(t)} e^{i\omega t} dt \right\} \end{aligned}$$

Wavepacket picture of 1-photon spectroscopy

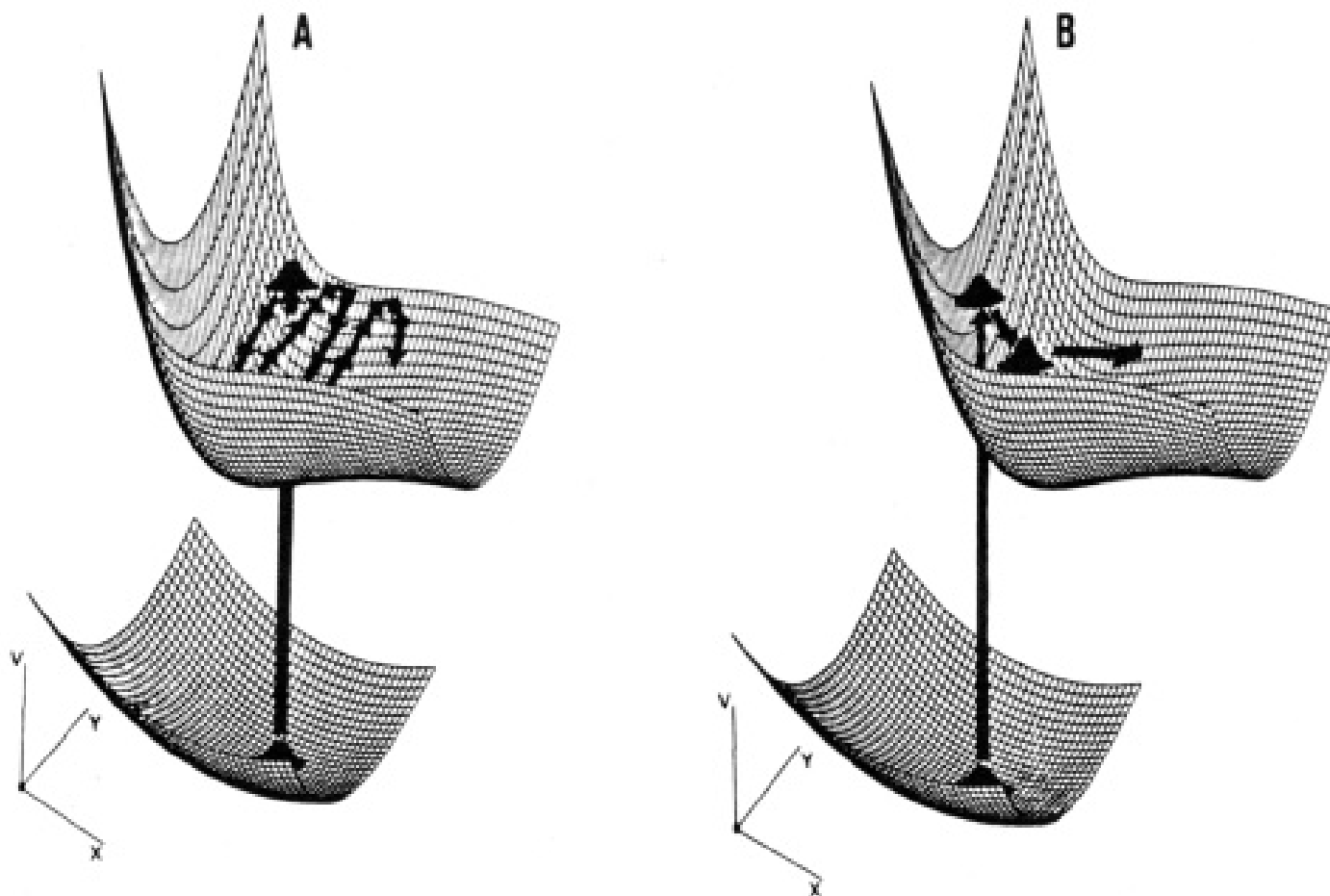


Figure 1. (A) Photoabsorption between two Born–Oppenheimer potential surfaces. The Franck–Condon wave packet $\phi(t)$, arising out of $\phi = \mu\chi$ (χ is shown on the lower surface, ϕ and $\phi(t)$ on the upper), takes a circuitous path in this case, “grazing” $\phi(0)$ several times on the way to dissociation. The result is an absorption band with some low-resolution vibrational structure. (B) Direct dissociation leading to a broad, featureless absorption band.

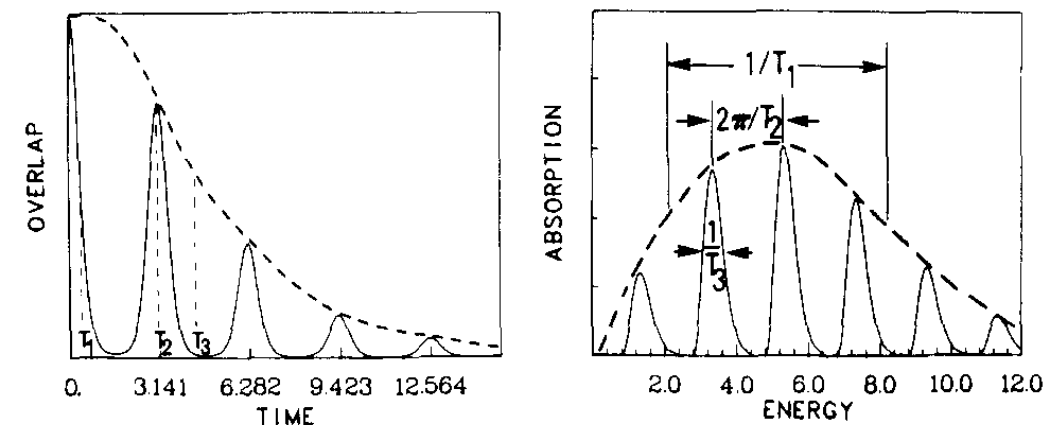


Figure 3. (Left) Absolute value of $\langle \phi | \phi(t) \rangle$ for the case of Figure 1B. Three important time scales, T_1 , T_2 , and T_3 , are shown. (Right) Spectrum arising out of Figure 1A and the autocorrelation function. Note how T_1 , T_2 , and T_3 affect the spectrum.

- Shortest time-scale in $C(t)$ defines total width of spectrum.
- Longest time-scale in $C(t)$, its overall decay time, defines width of highest resolution features.

Example: Photoabsorption spectrum of pyrazine



Molecular dynamics of pyrazine after excitation to the S_2 electronic state using a realistic 24-mode model Hamiltonian

A. Raab, G. A. Worth, H.-D. Meyer, and L. S. Cederbaum
*Theoretische Chemie, Physikalisch-Chemisches Institut, Universität Heidelberg, INF 253,
D-69120 Heidelberg, Federal Republic of Germany*

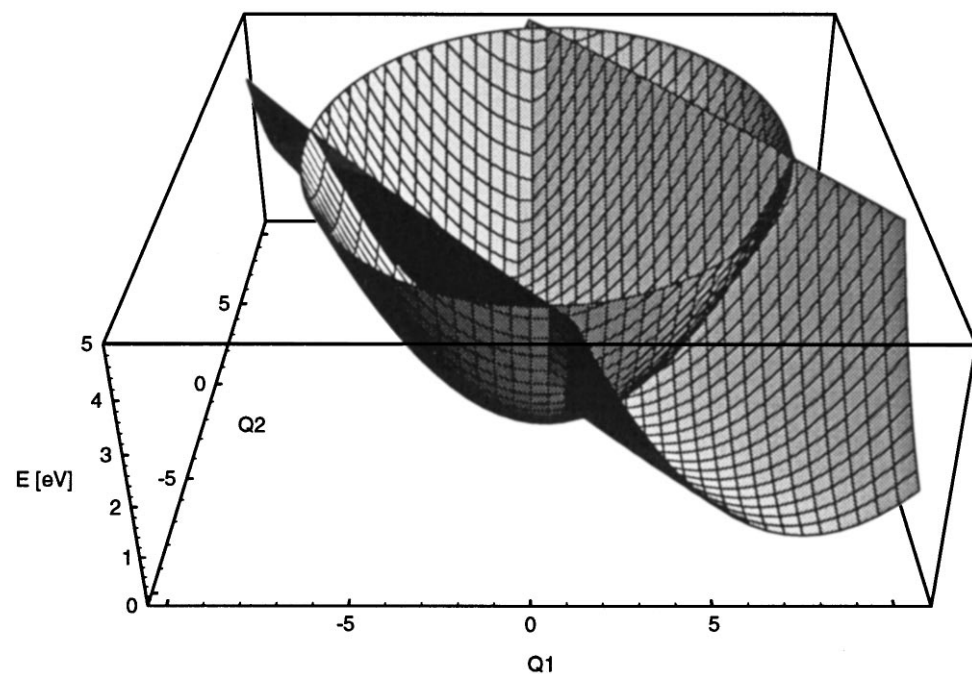
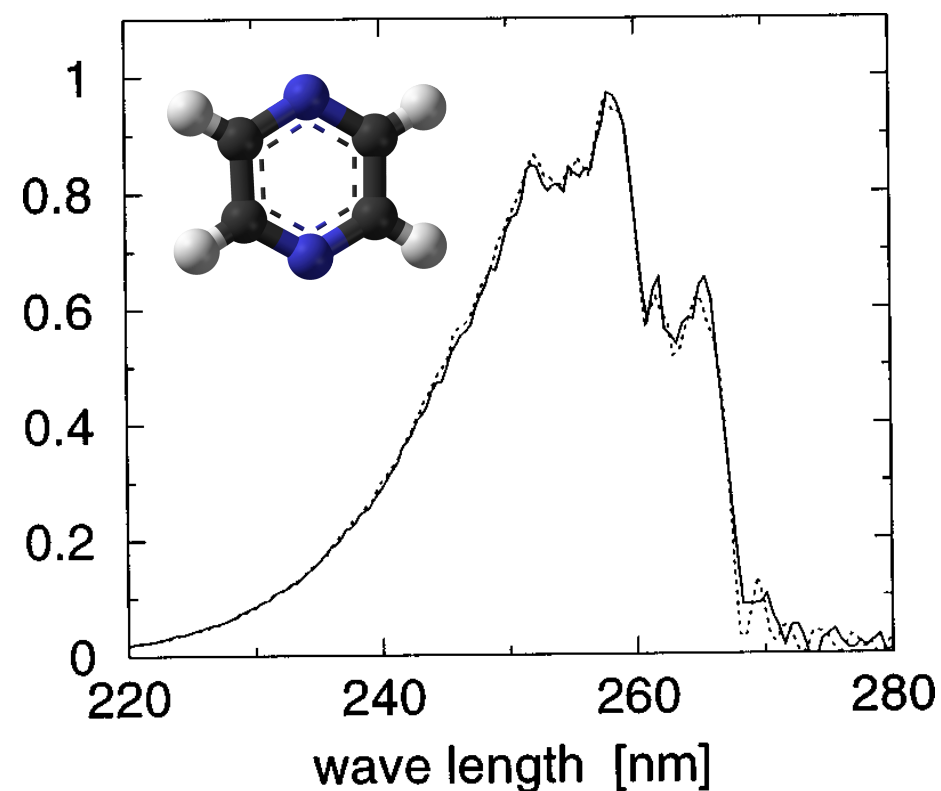
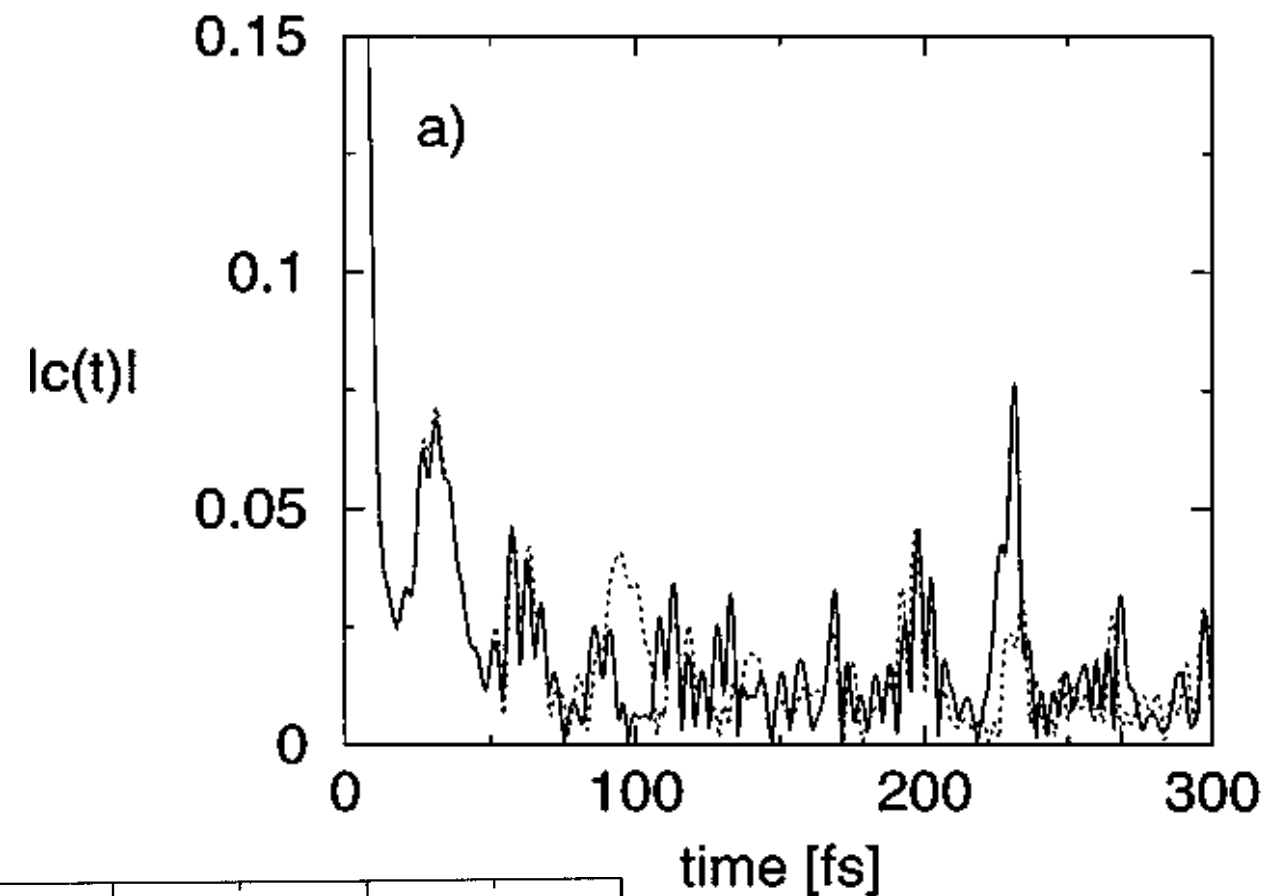


FIG. 1. The PES in dependence of the modes ν_{18b} and ν_{19b} ($Q1 = Q_{18b}$, $Q2 = Q_{19b}$, the remaining coordinates are set to zero): The S_2 PES is rather flat for $Q_{18b} = -Q_{19b}$ (the diabatic picture) leading to a rather interesting conical intersection with the S_1 PES. Note that the mode pair (ν_{18b}, ν_{19b}) belongs to the set G_2 [cf. Eq. (1)] and thus introduces a bilinear coupling.





Broad-band excitation results in wavepacket dynamics

■ ... always.

● ... only if the pulse is sufficiently intense.

▲ ... only if the system has states spanning the given bandwidth.

◆ ... only if there is no observer.

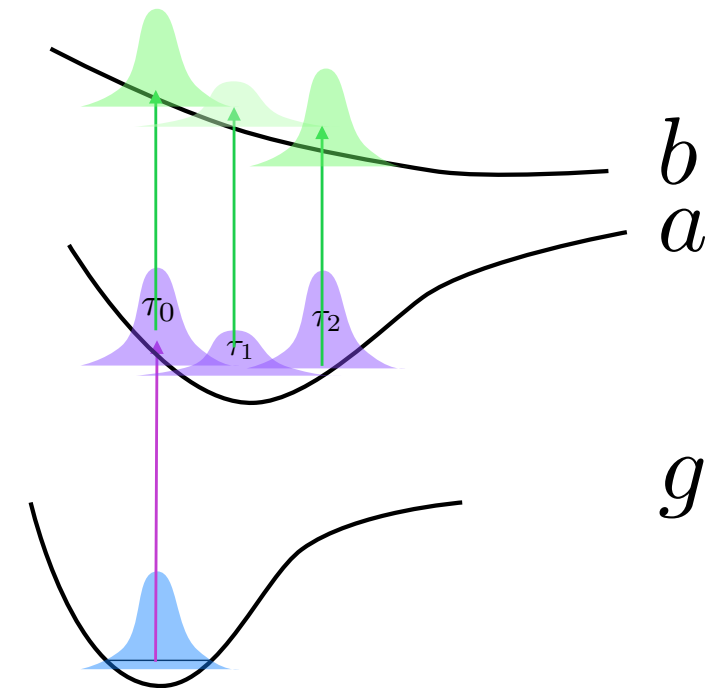
Transient (x-ray) absorption spectrum



Pump and probe pulses do not overlap

Probe pulse is after pump pulse

Probe pulse is short



$$\sigma(\omega, \tau) = \frac{4\pi\omega}{\hbar c} \text{Re} \left[\int_0^\infty dt e^{i\omega t} \langle \Psi_a(\tau) | e^{iH_a t/\hbar} \hat{\mu}_{ab} e^{-iH_b t/\hbar} \hat{\mu}_{ab} | \Psi_a(\tau) \rangle \right]$$

Transient (x-ray) absorption spectrum



Short time approximation:

$$e^{iH_a t/\hbar} \hat{\mu}_{ab} \longrightarrow \hat{\mu}_{ab} e^{iH_a t/\hbar}$$

$$\begin{aligned} e^{iH_a t/\hbar} e^{-iH_b t/\hbar} &= e^{-i(H_b - H_a)t/\hbar} + \mathcal{O}(t^2) + \dots \\ &\approx e^{-i(V_b - V_a)t/\hbar} \end{aligned}$$

$$\sigma(\omega, \tau) = \frac{4\pi\omega}{\hbar c} \text{Re} \left[\int_0^\infty dt e^{i\omega t - \Gamma t/\hbar} \langle \Psi_a(\tau) | \hat{\mu}_{ab} e^{-i(V_b - V_a)t/\hbar} \hat{\mu}_{ab} | \Psi_a(\tau) \rangle \right]$$

$$\sigma(\omega, \tau) = \frac{4\pi\omega}{c} \int d\mathbf{R} |\chi(\mathbf{R}, \tau)|^2 |\mu(\mathbf{R})|^2 \frac{\Gamma}{\Gamma^2 + [V_b(\mathbf{R}) - V_a(\mathbf{R}) - \hbar\omega]^2}$$

(Lorentzian limit)



Tracing chemical dynamics with (ultrafast) x-rays. Examples and applications.



Why x-rays?

- because they are fast.
- because they oscillate a lot.
- ▲ because they have large specificity.
- ◆ because they are intense.



Table 3. Expected properties of sub-femtosecond radiation at several operation points generated in the SASE3 undulator with the XLEAP scheme.

Photon energy (eV)	600	870	1000	1500	2300
Undulator cells	6	7	6/7	8	8
Duration (as, FWHM)	500	500	400/550	300	300
Peak power (GW)	600	500	100/650	800	400
Pulse energy (μ J)	350	300	45/350	250	100

Overview of options for generating high-brightness attosecond x-ray pulses at free-electron lasers and applications at the European XFEL
S Serkez et al., *J. Opt.*, **20**, 024005 (2018)

Ultrafast x-ray light sources



Source	Photon energy	Pulse energy	Repetition rate	Average power	Pulse duration
High-harmonic generation	<1.5 keV (REF. ¹¹⁵); large flux for time-resolved experiments at <450 eV (REFS ^{71,72,116})	• 10 μJ (at 10 Hz, 20 eV) ^{156,157} • Typically <1 nJ at 1 kHz (REFS ^{157–159})	Up to 80 MHz (REF. ¹⁶⁰)	• <10 μW (REFS ^{160,161}) • 1 mW expected ¹⁶²	43 as (at 100 eV) ²⁸
Free-electron laser	<24 keV (0.5 Å) ^{95,96}	Up to 1 mJ	~100 Hz, 27 kHz and 1 MHz planned ⁹⁶	Up to 120 mW (REFS ^{95,96})	• 4.4 fs (REF. ²¹) • Bandwidth supports 200 as (REF. ²²)
Synchrotron (time sliced)	<100 keV (REF. ¹⁶³)	1 nJ (REF. ¹⁶³)	1–10 kHz (REF. ¹⁶³)	10 μ W (REF. ¹⁶³)	100 fs (REF. ¹⁶³)

The ultrafast X-ray spectroscopic revolution in chemical dynamics
 Kraus, Zürich, Cushing, Neumark, Leone, Nat. Rev. Chem., **2**, 82 (2018)

Ultrafast x-ray light sources ("table top")



ARTICLE

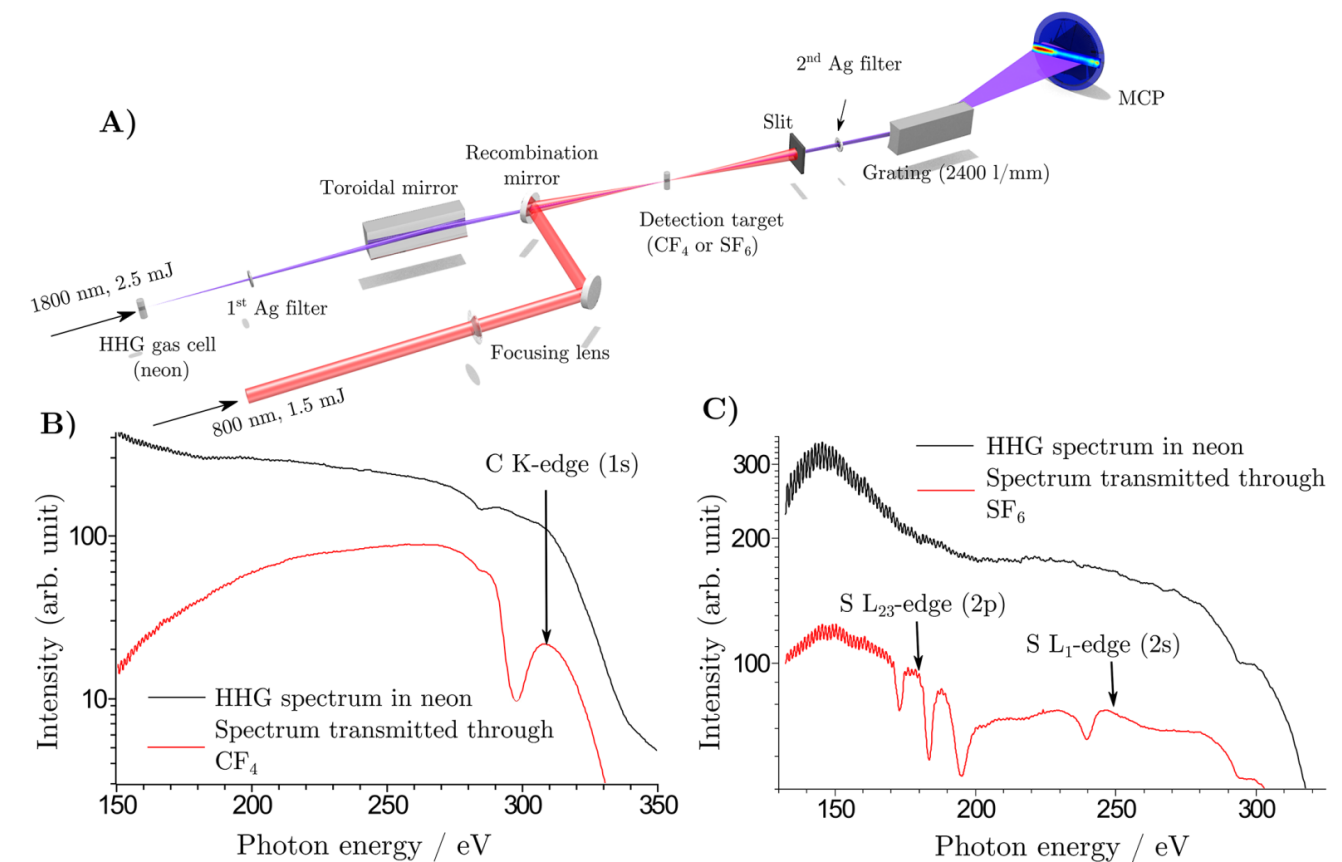
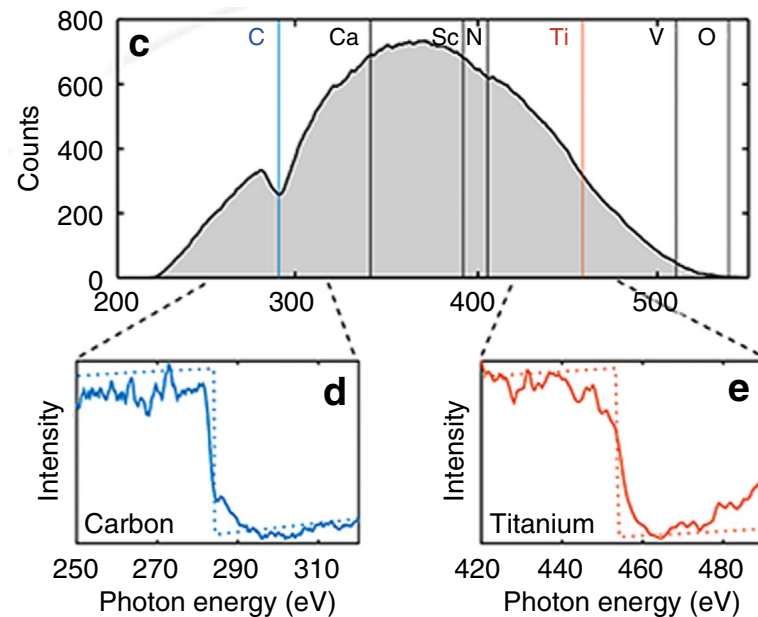
Received 6 Nov 2015 | Accepted 31 Mar 2016 | Published 11 May 2016

DOI: 10.1038/ncomms11493

OPEN

0.5-keV Soft X-ray attosecond continua

S.M. Teichmann¹, F. Silva¹, S.L. Cousin¹, M. Hemmer¹ & J. Biegert^{1,2}



0.5-keV Soft X-ray attosecond continua

SM Teichmann et al. Nat. Commun. 7, 11493 (2016)

Time-resolved x-ray absorption spectroscopy with a water window high-harmonic source

Y. Pertot et al. Science, 355, 264 (2017)

Ultrafast x-ray light sources ("table top")



ARTICLE

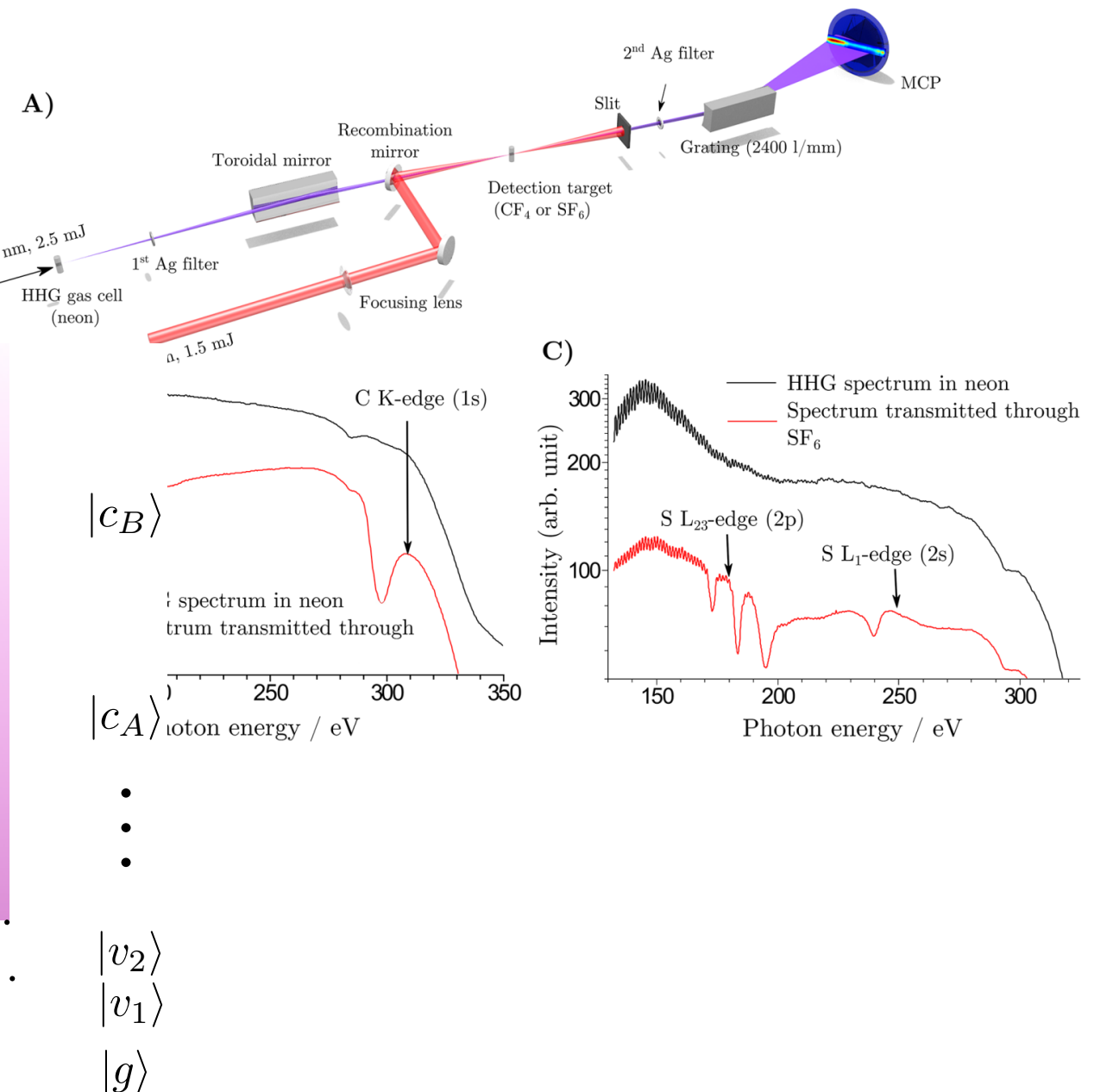
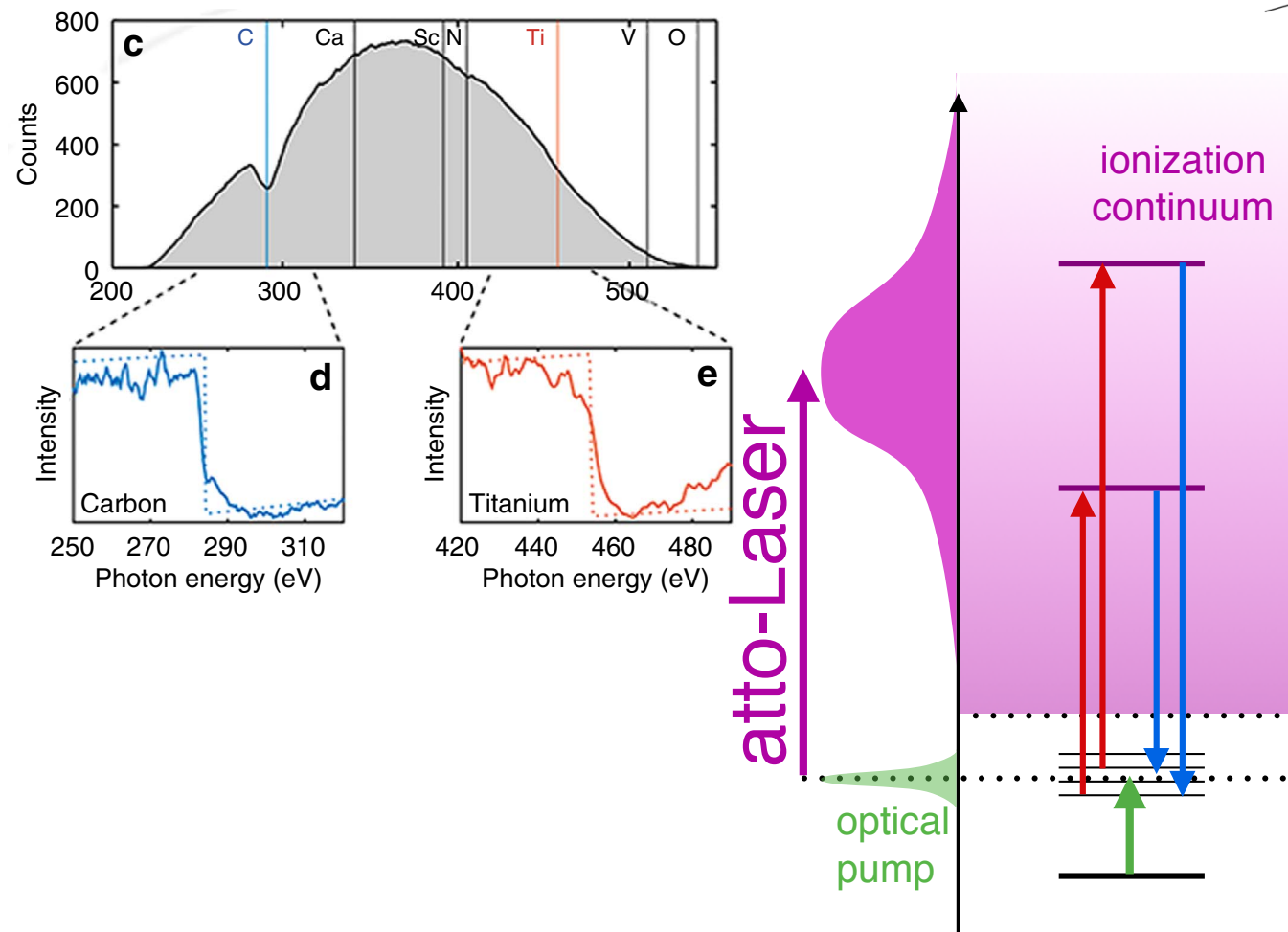
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SM Teichmann et al. Nat. Commun. 7, 11493 (2016)

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Chemical reactions: what do we want to know?



- Nuclear configuration.
- Electronic configuration.
- Electronic state populations.
- Electronic coherences.
- Lifetimes, durations.

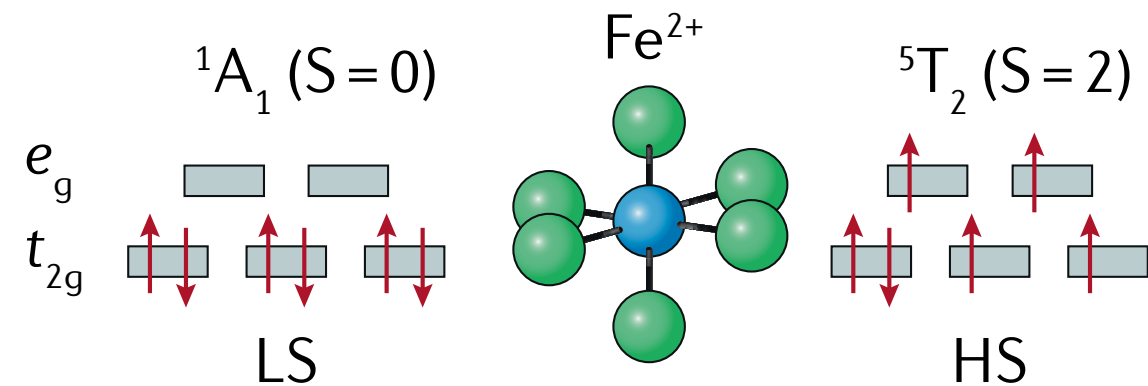
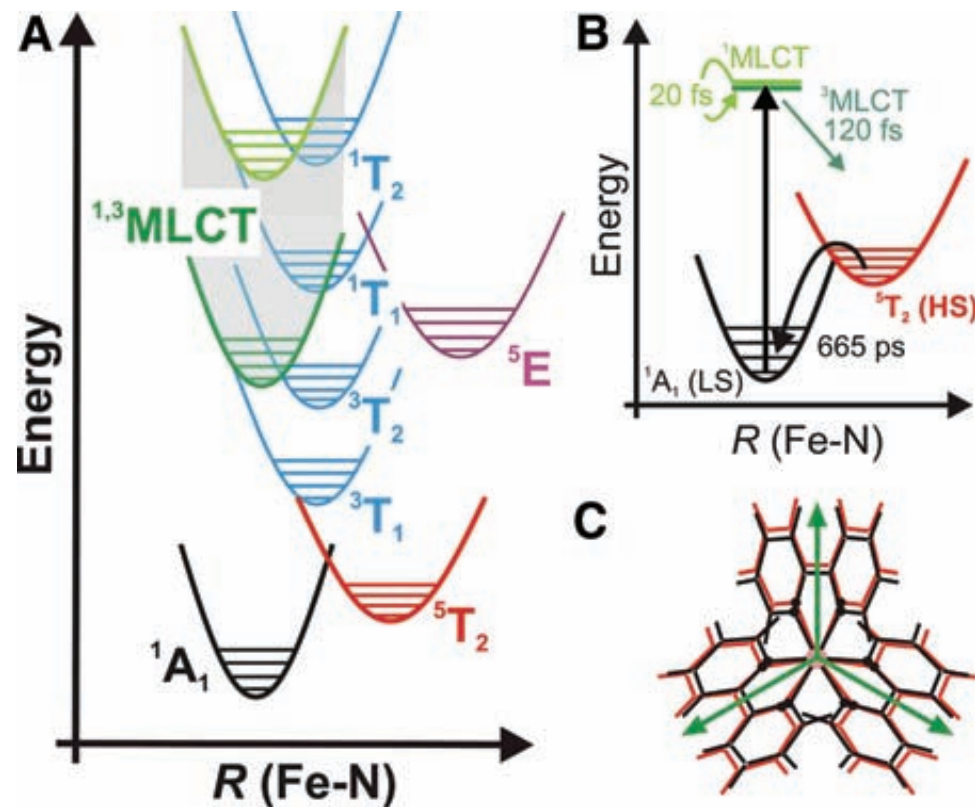
Overview of (TR) x-ray techniques



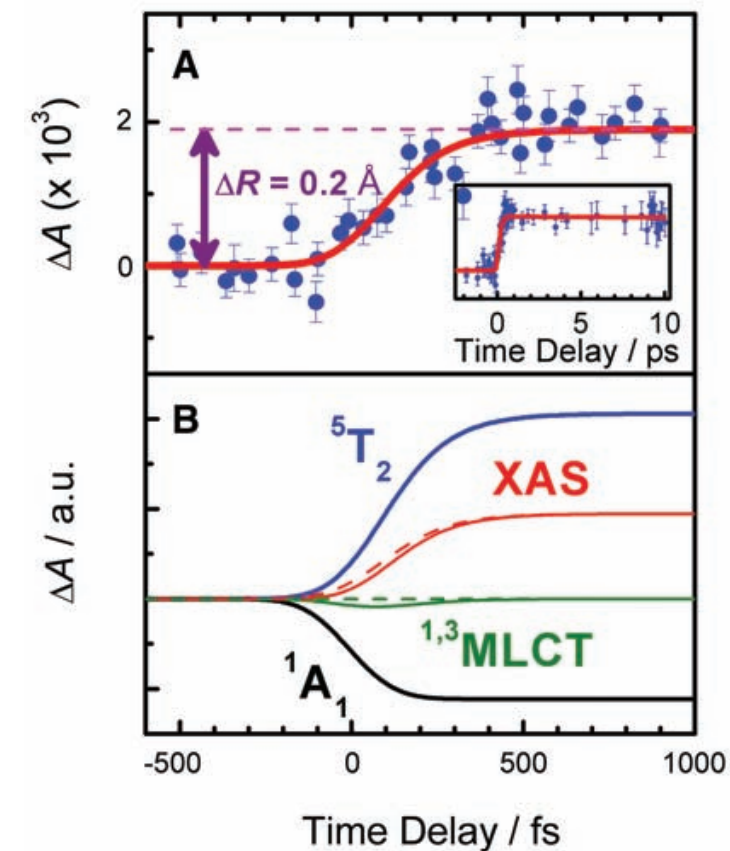
- (TR)-XAS: XANES / EXAFS (Near edge and extended edge absorption)
Core electrons are excited to the valence shell near an edge or several eV. above an edge.
- (TR)-RIXS (Resonant inelastic x-ray scattering)
Core electrons are excited to the valence shell. The core shell is refilled by an electron from another shell and the emitted electron is detected.
- (TR)-PES (Photoelectron spectroscopy)
Core electrons are ionized into the continuum and their energy and/or momentum vector detected.
- Photofragmentation spectroscopy.
A system previously excited by an XUV or x-ray pulse is further ionized by a short probe pulse (IR, XUV,...). Fragments are measured as a function of the time delay.
- (TR)-X-ray diffraction
Diffraction of x-rays at a molecule reveals structural information, possibly time resolved.

The ultrafast X-ray spectroscopic revolution in chemical dynamics
Kraus, Zürich, Cushing, Neumark, Leone, Nat. Rev. Chem., **2**, 82 (2018)

Time-resolved XANES of spin-crossover dynamics

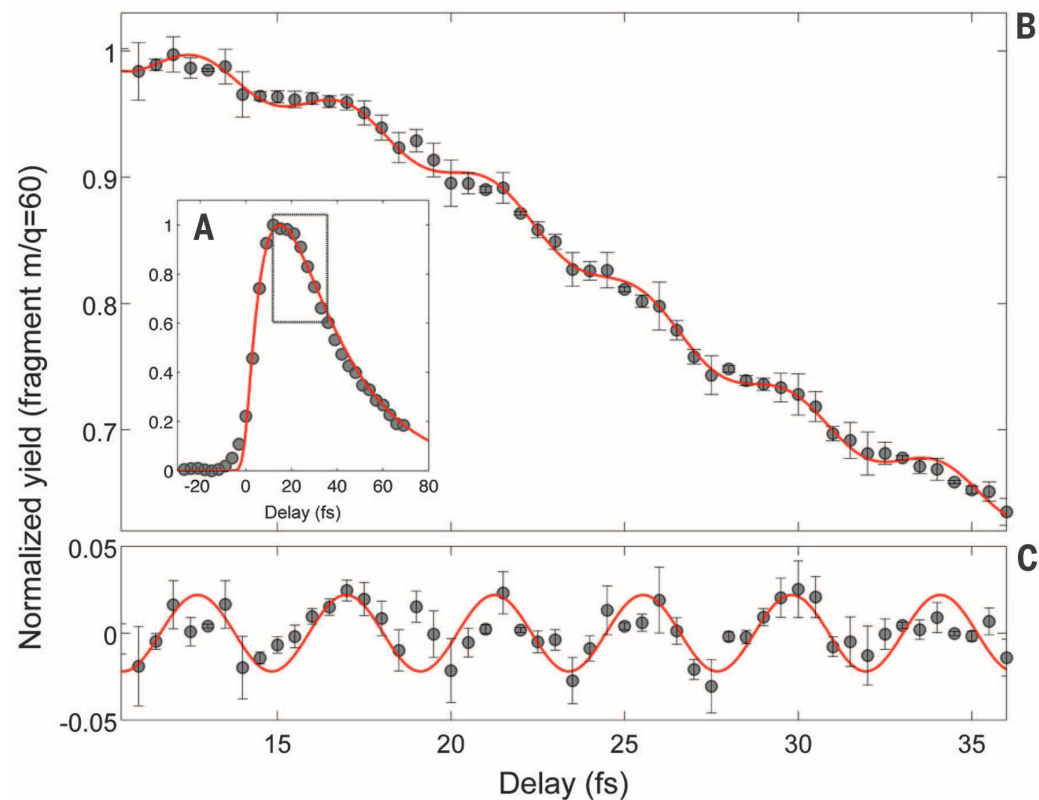
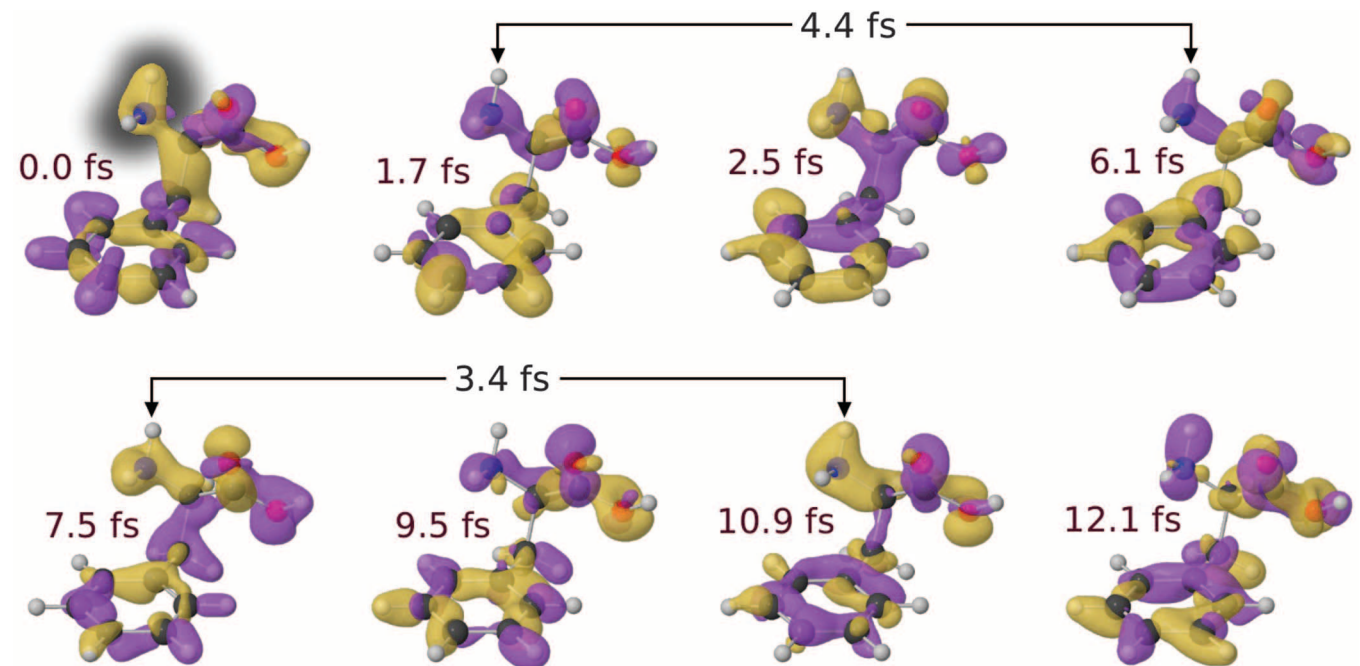
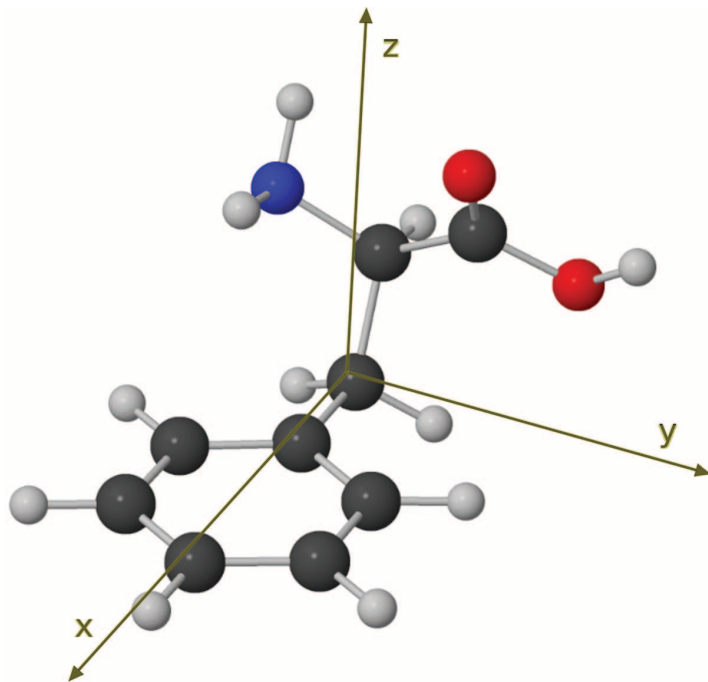


- 400 nm laser excitation followed by Fe K-Edge x-ray probe at about 7.2 keV.
- Conversion from low spin to high spin can be followed in the modulation of the x-ray absorption
- Experiments @ Swiss Light Source.



Femtosecond XANES Study of the Light-Induced Spin Crossover Dynamics in an Iron(II) Complex
Ch. Bressler et al. *Science*, **323**, 489 (2009)

Electronic coherence in atoms and molecules



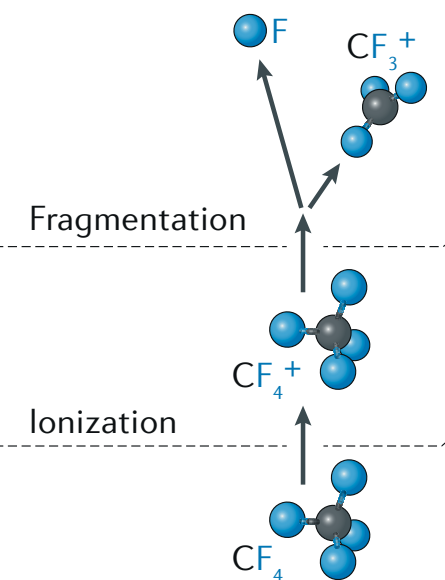
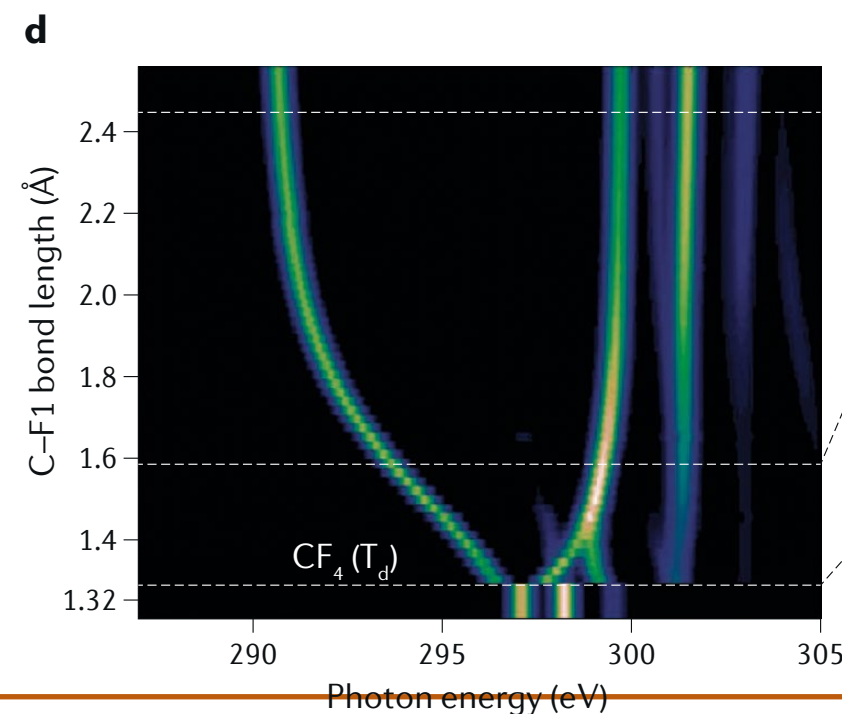
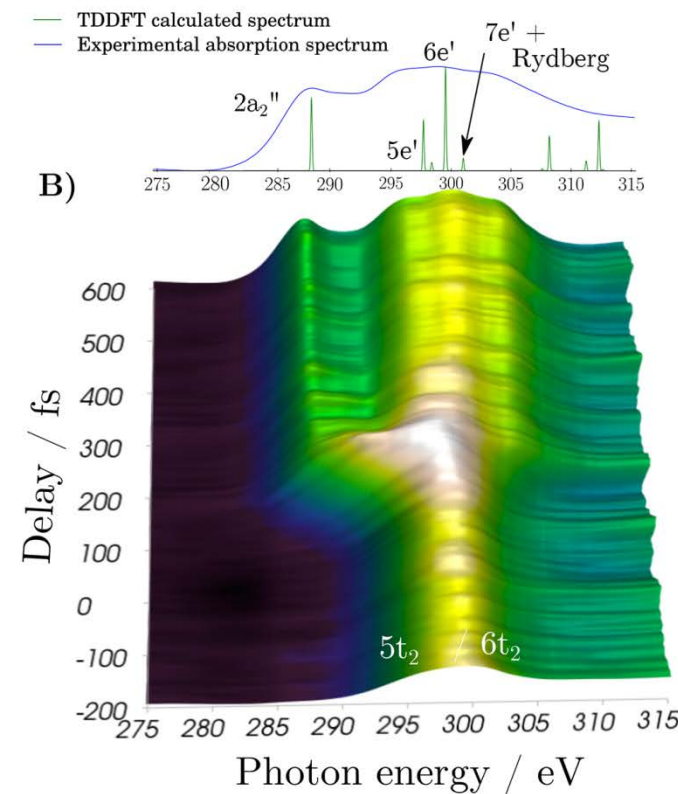
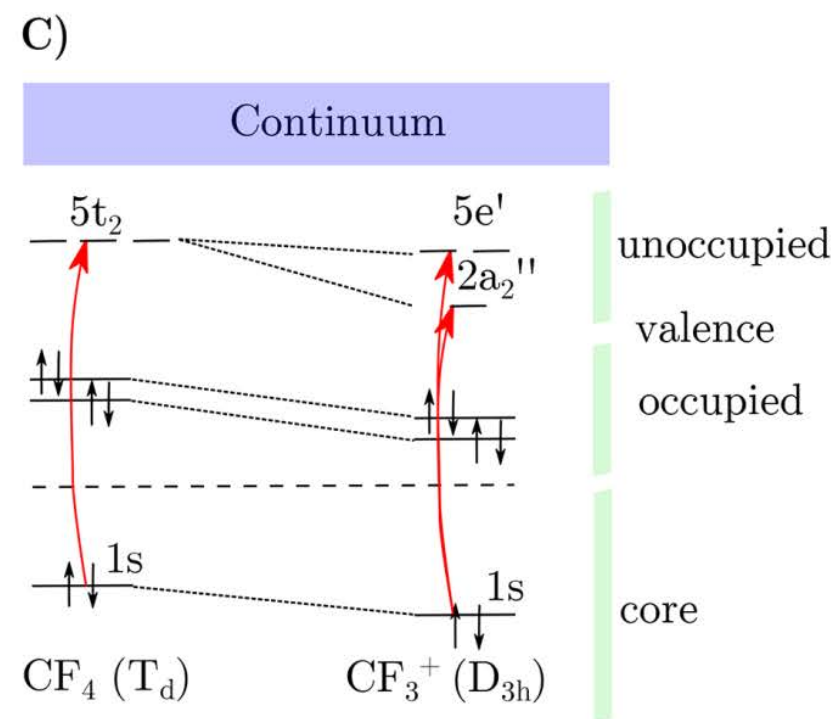
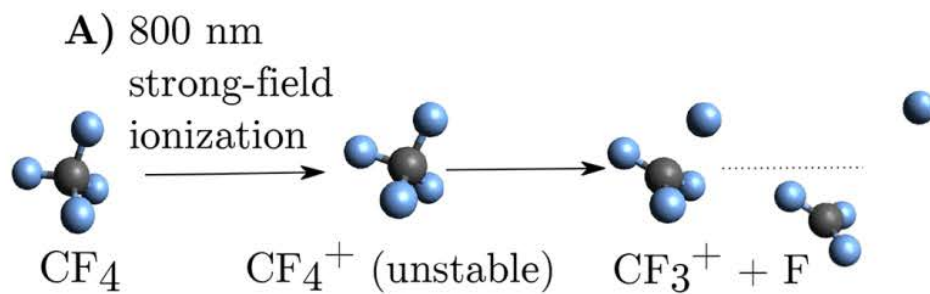
- An attosecond pulse starts electronic and vibrational dynamics in the molecule. Electrons rearrange.
- A delayed IR pulse fragments the molecule. Fragments detected as a function of the pump-probe delay inform about the systems dynamics.

Ultrafast electron dynamics in phenylalanine initiated by attosecond pulses
F. Calegari et al. *Science*, **346**, 336 (2014)

Chemical reactions followed by soft x-ray spectroscopy

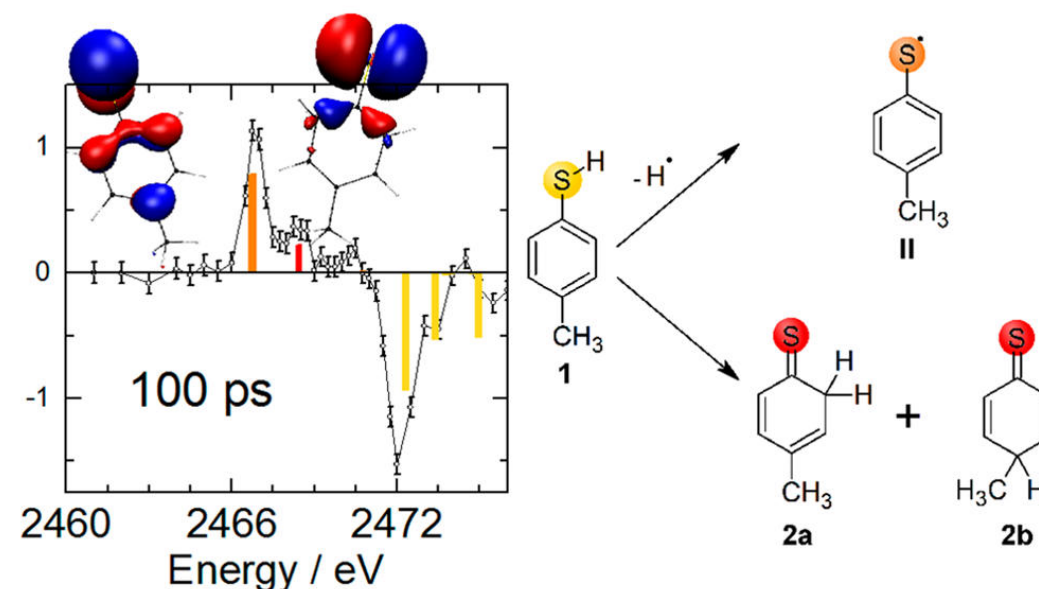
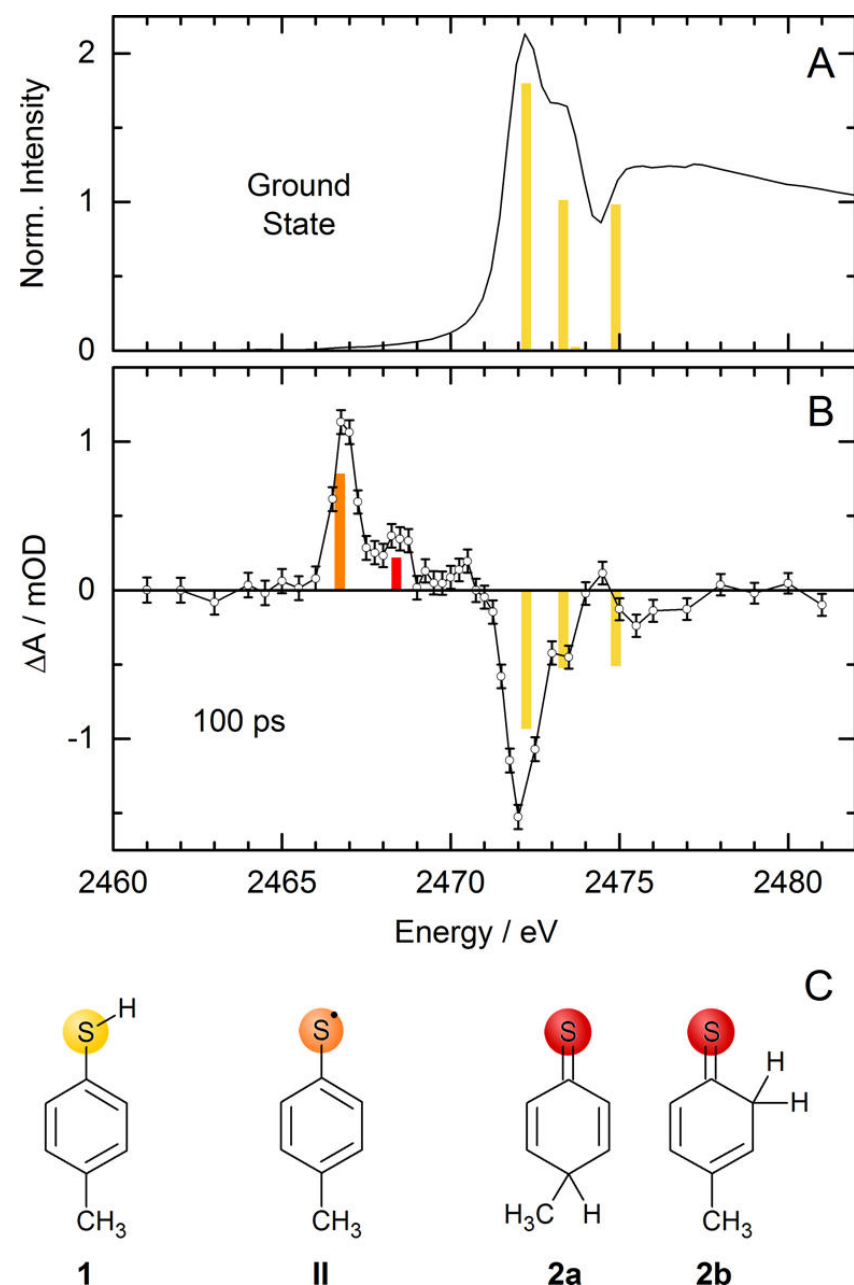


- A mid IR pulse ionizes CF₄ into its cationic ground state. The molecule undergoes ultrafast fragmentation.
- A delayed XUV pulse probes the system at the Carbon K-edge. The transmitted XUV pulse is dispersed and analyzed.



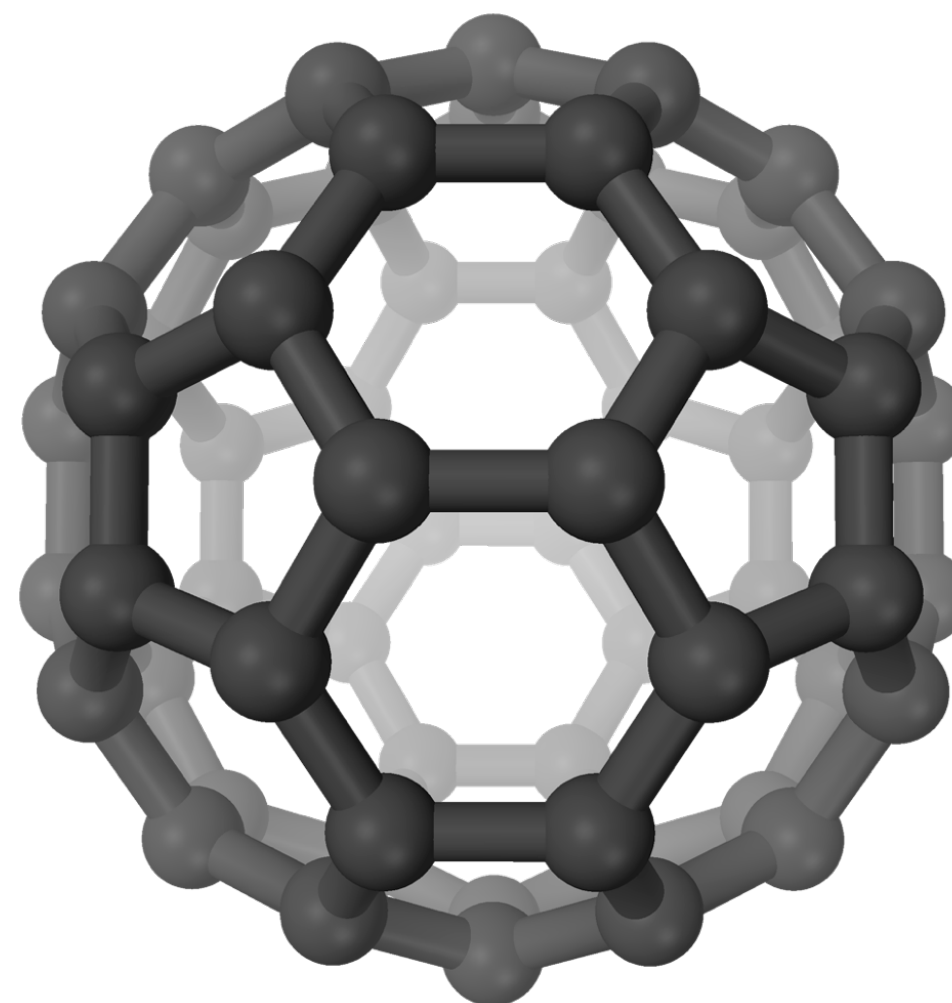
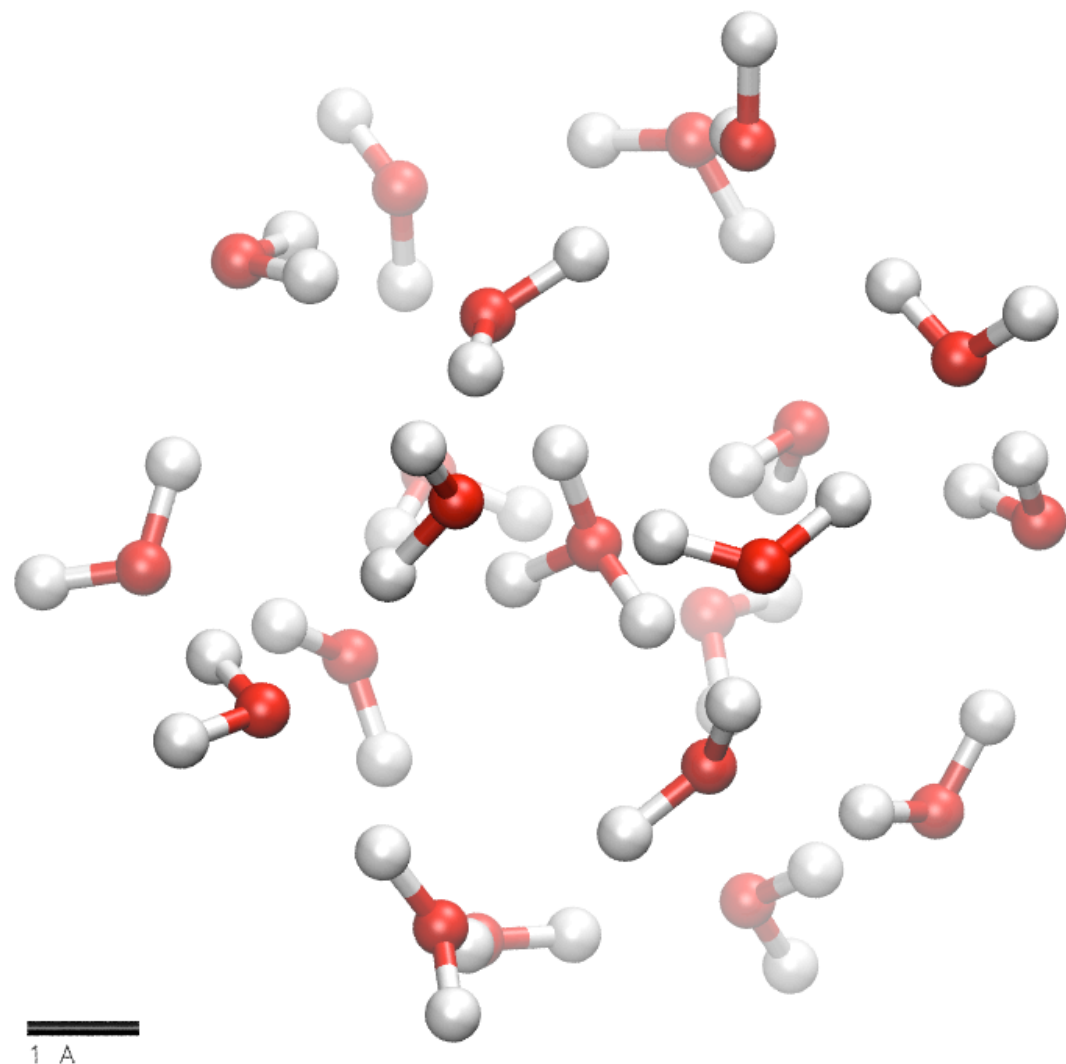
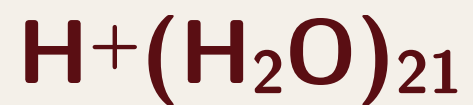
Time-resolved x-ray absorption spectroscopy with a water window high-harmonic source
Y. Pertot et al. *Science*, **355**, 264 (2017)

Photochemistry of sulfurated molecules



- A Thiol molecule is illuminated with laser light at 267 nm. Isomerization and appearance of radical species follow.
- These are probed by time-resolved X-ray absorption at the sulfur K-edge with a resolution of 70 ps.
- Experiments @ Advanced Light Source.

Light-Induced Radical Formation and Isomerization of an Aromatic Thiol in Solution Followed by Time-Resolved X-ray Absorption Spectroscopy at the Sulfur K-Edge
M. Ochmann *et al.*, *J. Am. Chem. Soc.*, 2017, **139**, 4797

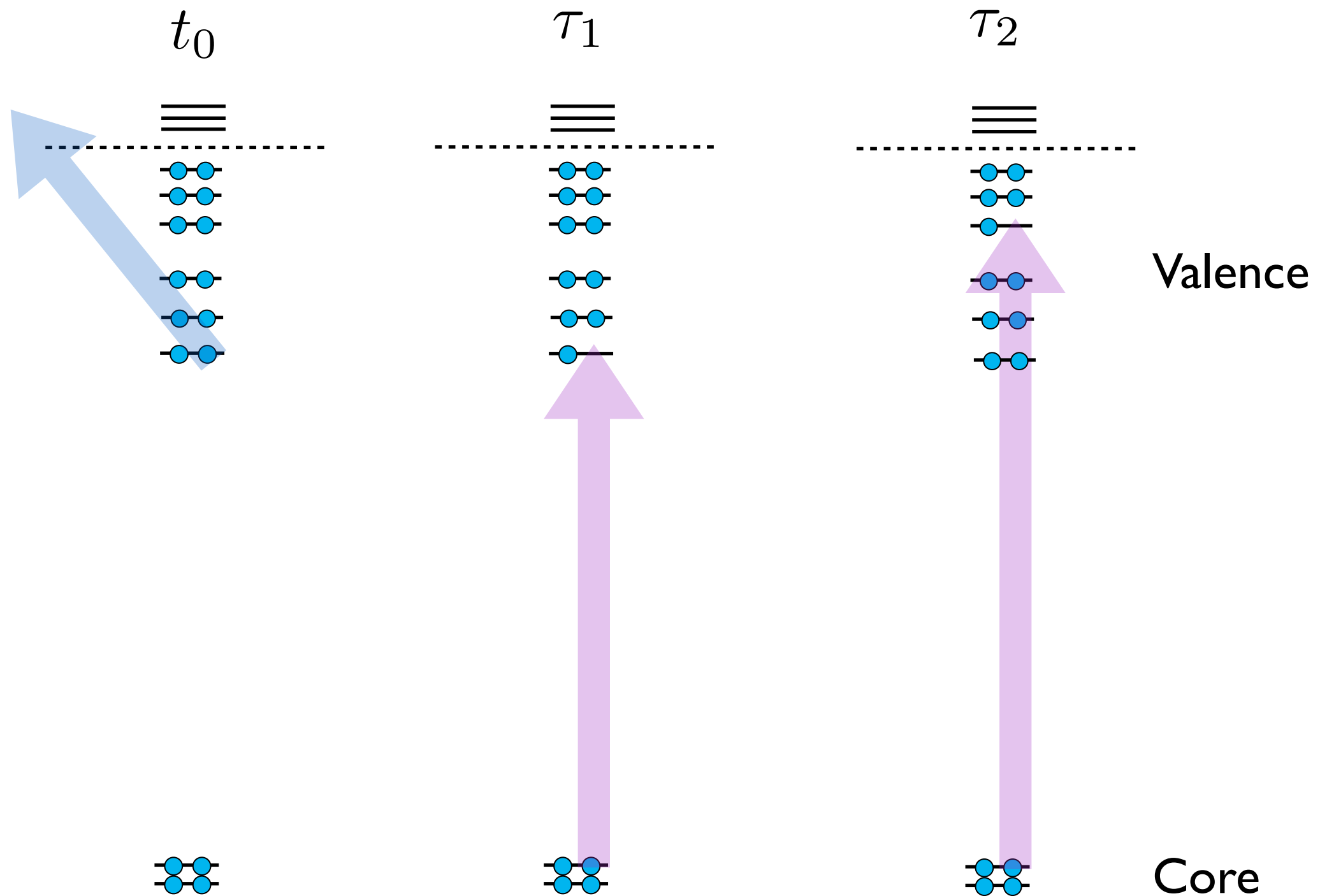


Core-level transient absorption spectroscopy as a probe of electron hole relaxation in photoionized $\text{H}^+(\text{H}_2\text{O})_n$
Z Li, ME Madjet, OV, R Santra, *Farad. Disc.* **171**, 457 (2014)

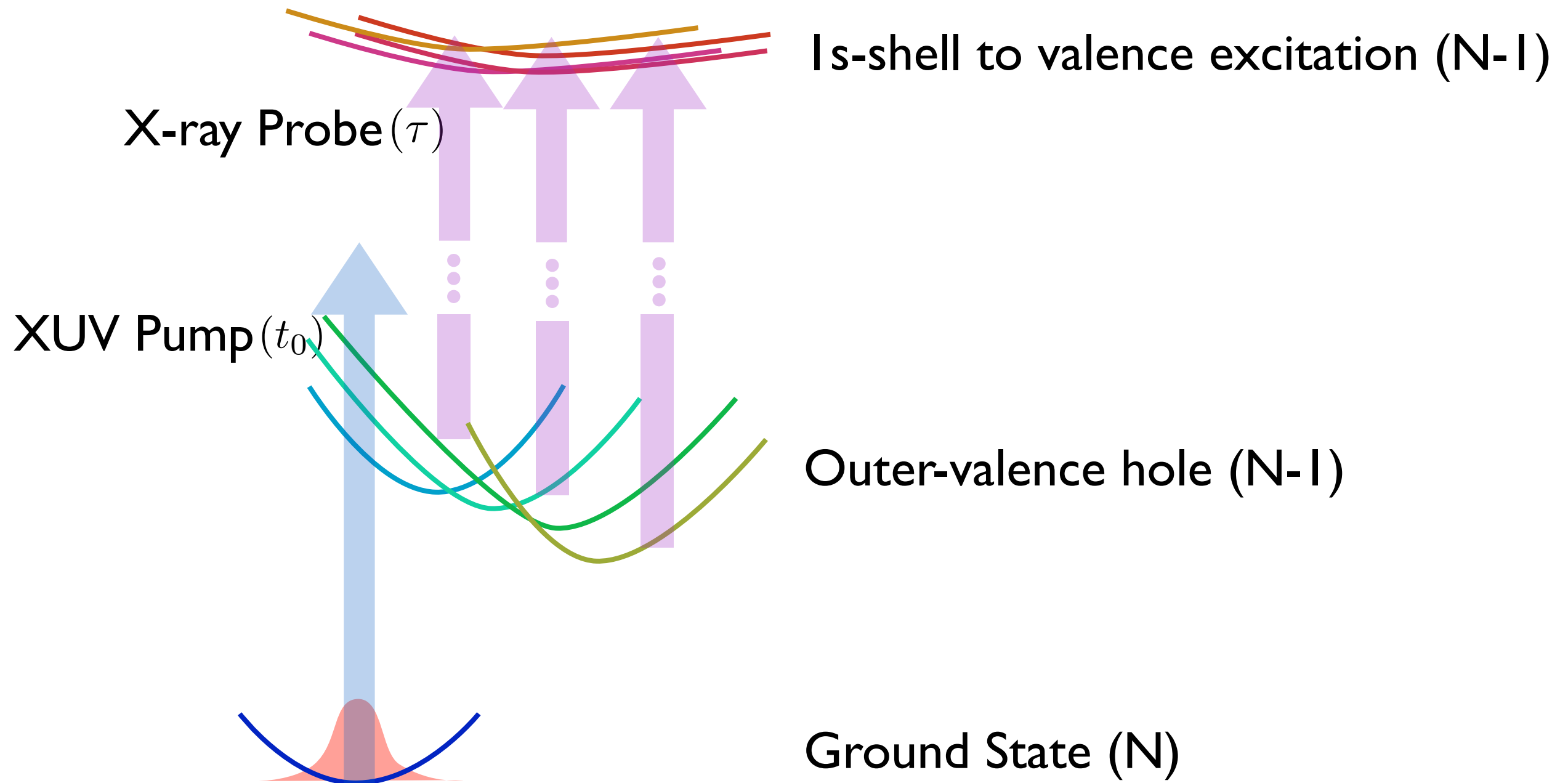
Transient x-ray absorption spectrum



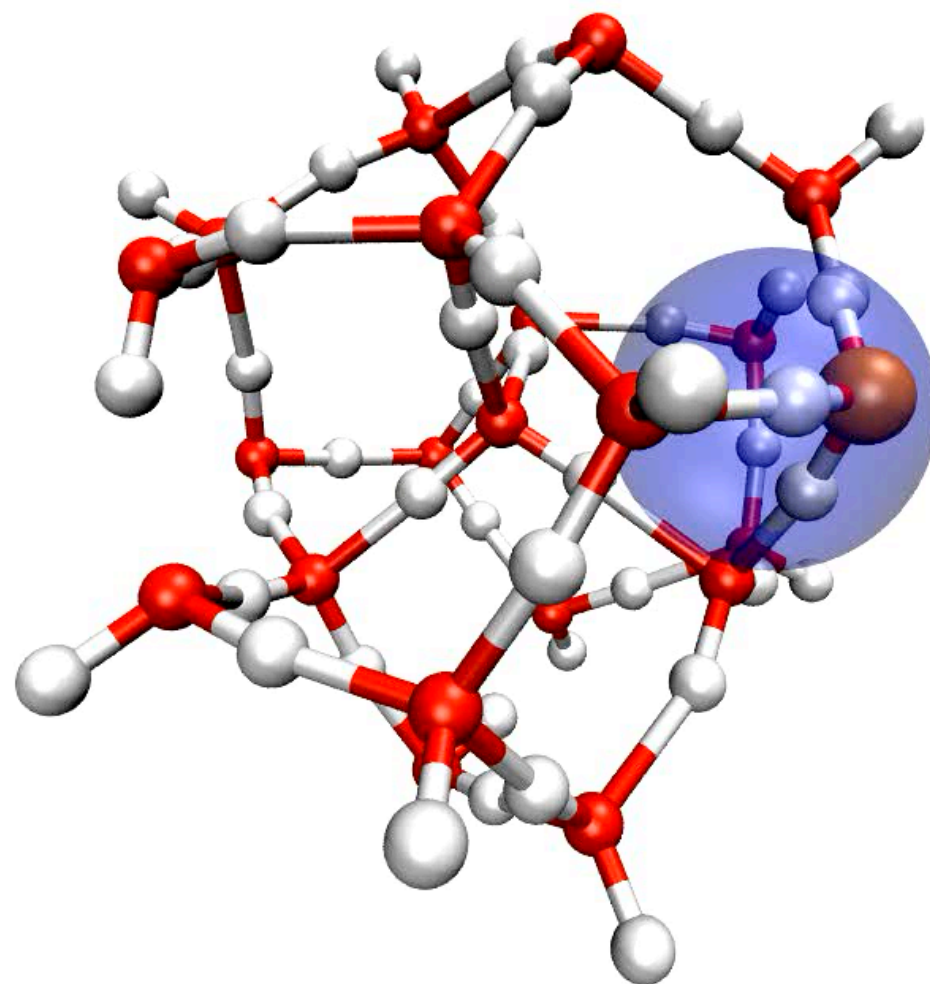
XUV pump - X-ray probe



Transient x-ray absorption spectrum



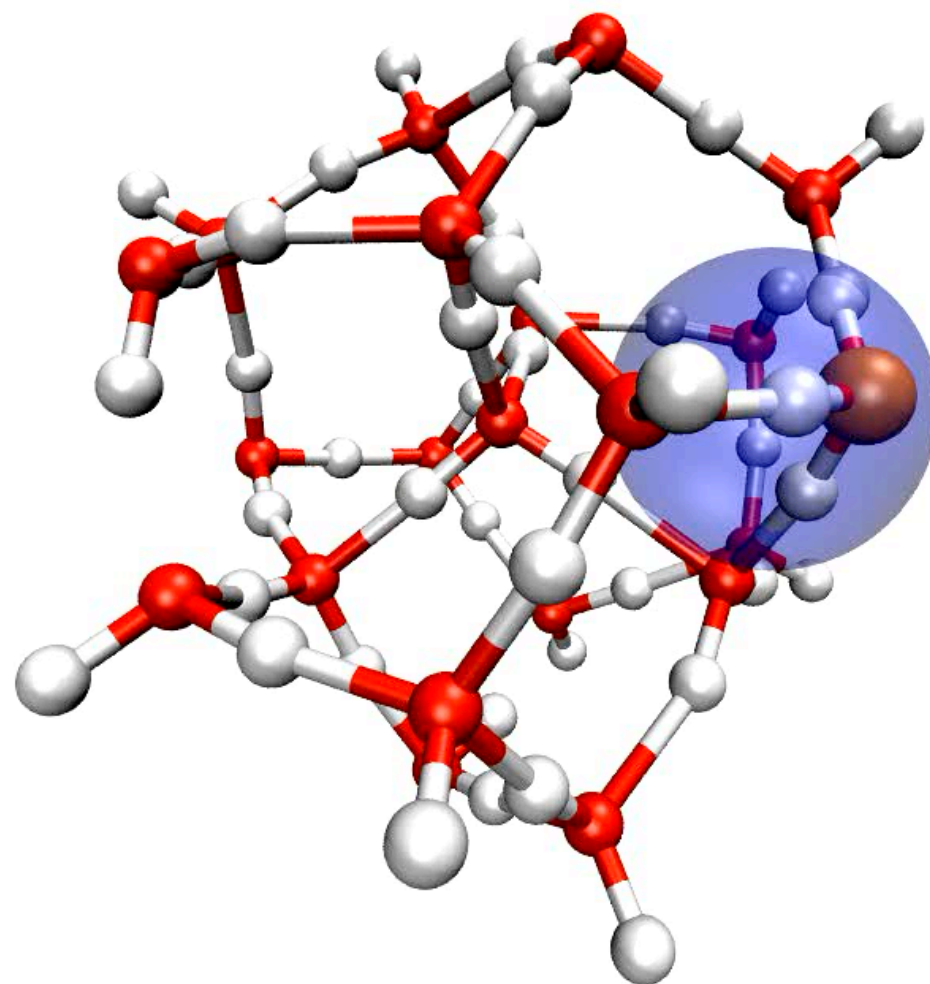
Ultrafast proton-hole motion



Core-level transient absorption spectroscopy as a probe of electron hole relaxation in photoionized $\text{H}^+(\text{H}_2\text{O})_n$

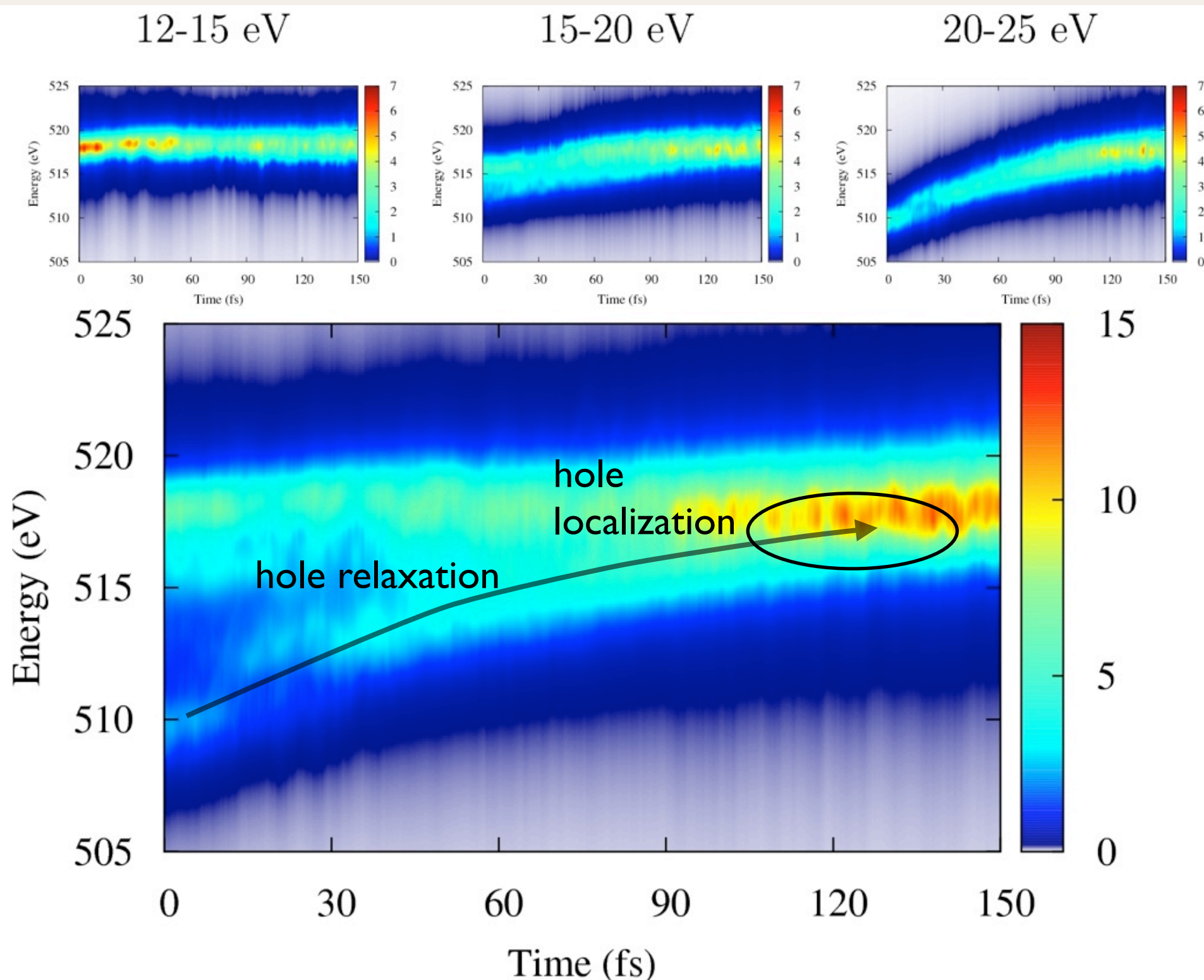
Z Li, ME Madjet, OV, R Santra, *Farad. Disc.* **171**, 457 (2014)

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Core-level transient absorption spectroscopy as a probe of electron hole relaxation in photoionized $\text{H}^+(\text{H}_2\text{O})_n$
Z Li, ME Madjet, OV, R Santra, *Farad. Disc.* **171**, 457 (2014)



J. Phys. Chem. **1995**, 99, 11255–11265

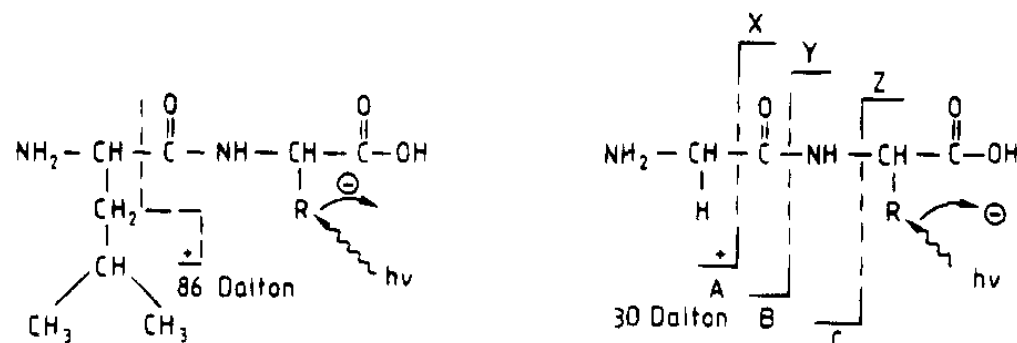
11255

Elementary Processes in Peptides: Electron Mobility and Dissociation in Peptide Cations in the Gas Phase

R. Weinkauf,* P. Schanen, D. Yang,[†] S. Soukara, and E. W. Schlag

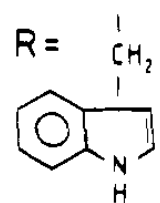
*Institut für Physikalische und Theoretische Chemie, Technische Universität München,
85747 Garching, Lichtenbergstrasse 4, Germany*

Received: January 24, 1995; In Final Form: April 28, 1995[⊗]



c)

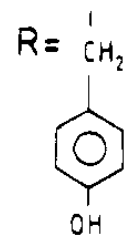
Tryptophan



IP ~ 7.5 eV

$\lambda_{\text{OPT}} = 286.7 \text{ nm}$

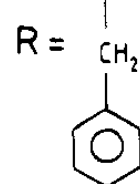
Tyrosine



IP ~ 8.0 eV

$\lambda_{\text{OPT}} = 281.7 \text{ nm}$

Phenylalanine



IP ~ 8.4 eV

$\lambda_{\text{OPT}} = 266.4 \text{ nm}$

- Hole transfer (migration) triggered by two-photon ionization with a nanosecond laser

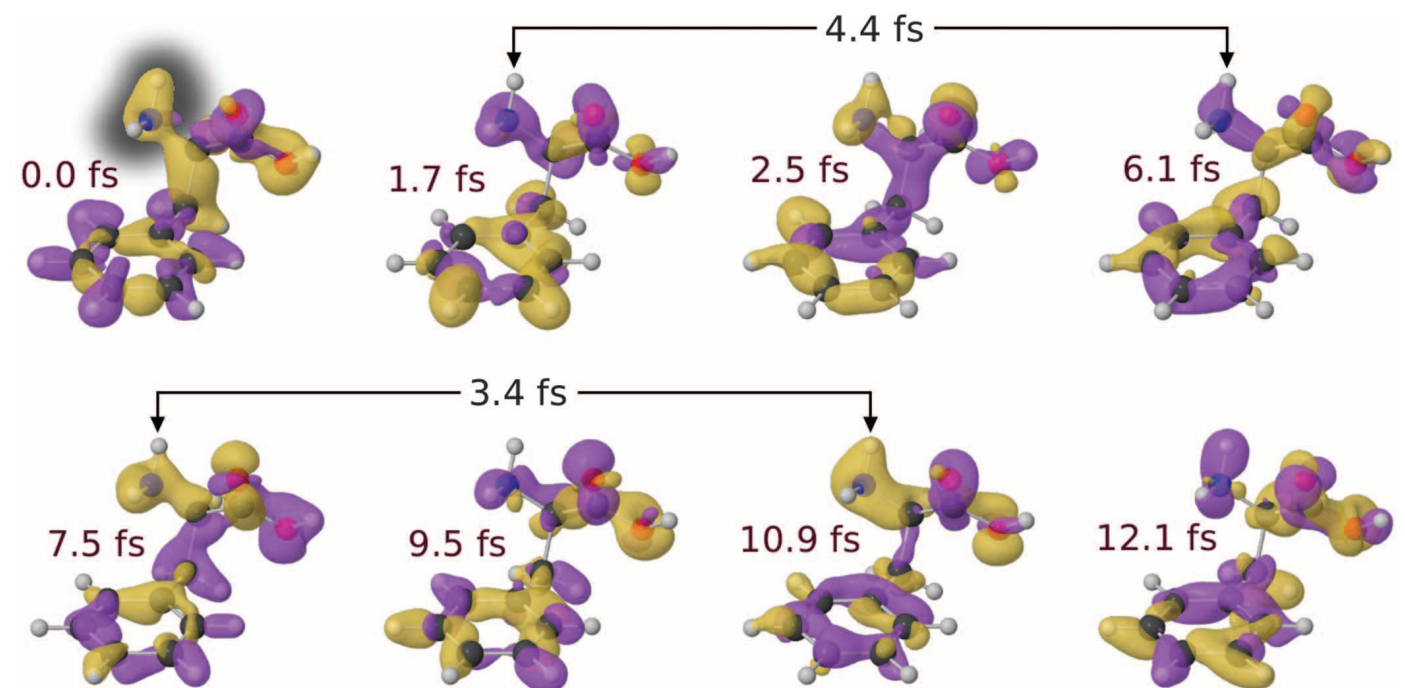
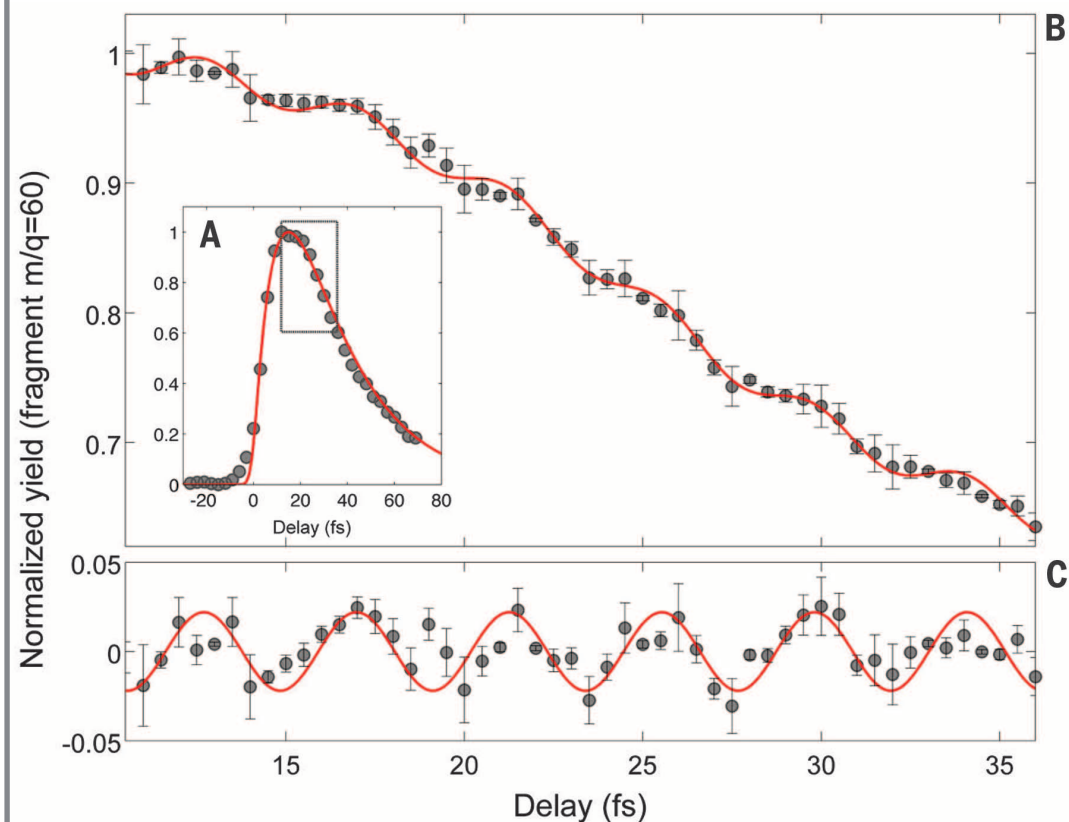


ATTOSECOND DYNAMICS

Ultrafast electron dynamics in phenylalanine initiated by attosecond pulses

F. Calegari,¹ D. Ayuso,² A. Trabattoni,³ L. Belshaw,⁴ S. De Camillis,⁴ S. Anumula,³ F. Frassetto,⁵ L. Poletto,⁵ A. Palacios,² P. Decleva,⁶ J. B. Greenwood,⁴ F. Martín,^{2,7*} M. Nisoli^{1,3*}

- XUV (sub 300 as) pump-VIS/NIR probe
- Oscillations in the yield of some fragments persists for more than 30 fs.

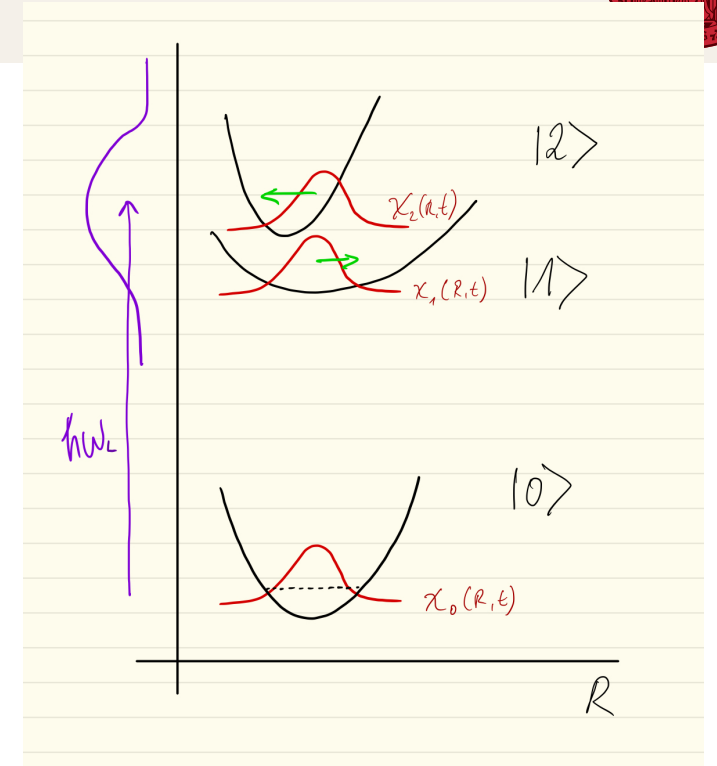


Science **346**, p. 336 (2014)

Electronic decoherence due to nuclear motion



$$\Psi(R, r, t) = \sum_j \chi_j(R, t) \psi(r; R) \quad \text{Born-Huang}$$



Electronic coherences $\Leftrightarrow$ nuclear wave packet overlaps

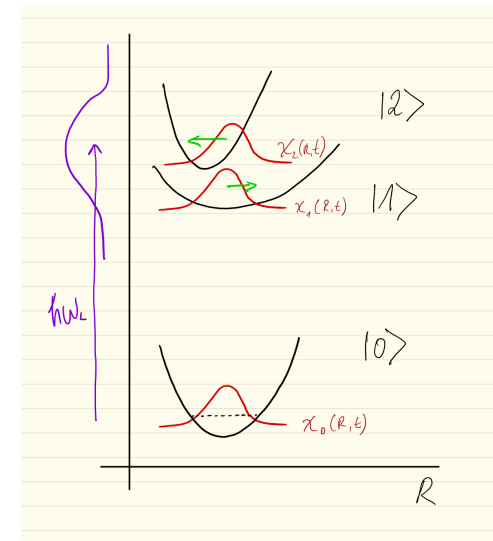
$$\rho_{ji}(t) = \langle \chi_j(R, t) | \chi_i(R, t) \rangle$$

$$\langle \hat{O}(r) \rangle = \sum_j \rho_{jj} O_{jj} + \sum_{j>i} 2\text{Re}\{\rho_{ji}\} O_{ji}$$

Fully quantum description of electronic decoherence



- Full dimensional
- Exact at short times

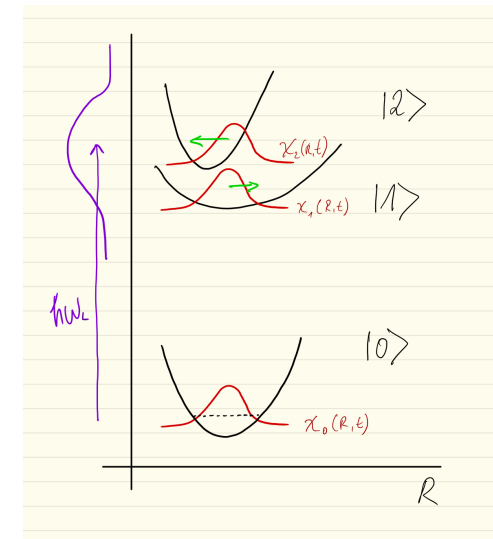


Electronic decoherence following photoionization: Full quantum-dynamical treatment of the influence of nuclear motion
C Arnold, OV, R Santra, *PRA* **95**, 033425 (2017)

Fully quantum description of electronic decoherence

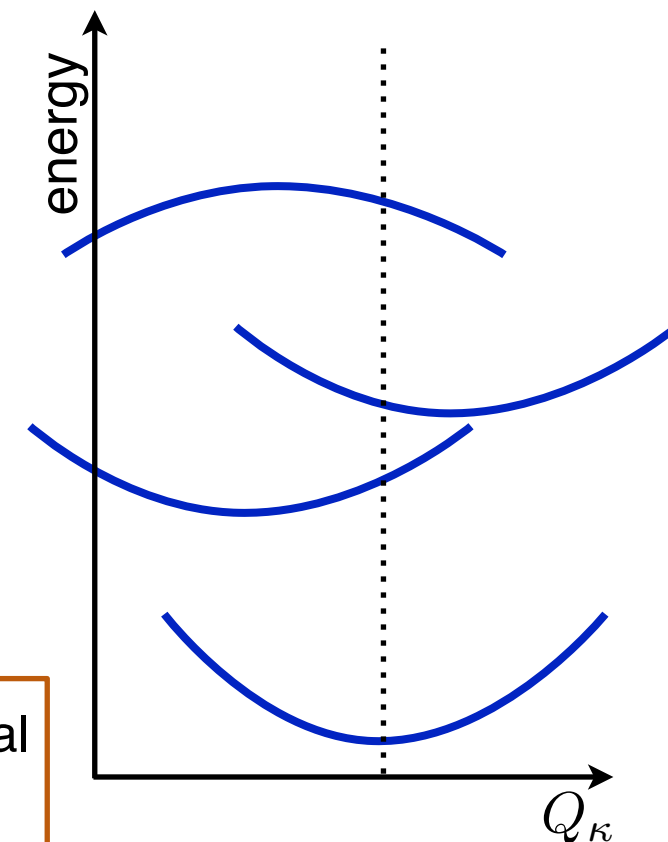


- Full dimensional
- Exact at short times



Hamiltonian: second order expansion of all relevant PES:

$$\hat{H} = \sum_{\kappa=1}^f \frac{\hbar\omega_j}{2} \left(\frac{\partial^2}{\partial Q_{\kappa}^2} + Q_{\kappa}^2 \right) \otimes \mathbf{1} + \sum_j \left(\Delta E^{(j)} + \sum_{\kappa=1}^f g_{\kappa}^{(j)} Q_{\kappa} + \sum_{\kappa\lambda} \gamma_{\kappa\lambda}^{(j)} Q_{\kappa} Q_{\lambda} \right) \otimes |j\rangle\langle j|$$

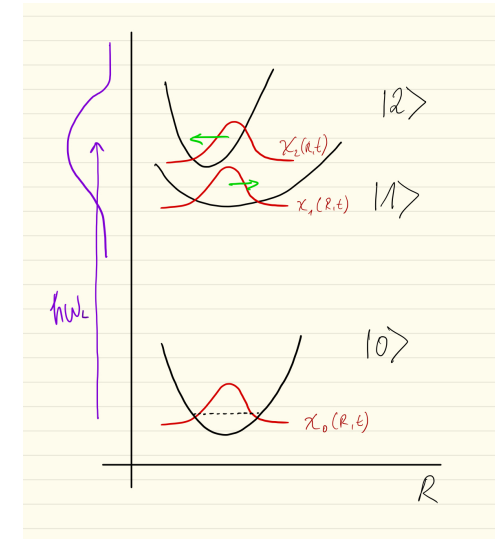


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Fully quantum description of electronic decoherence



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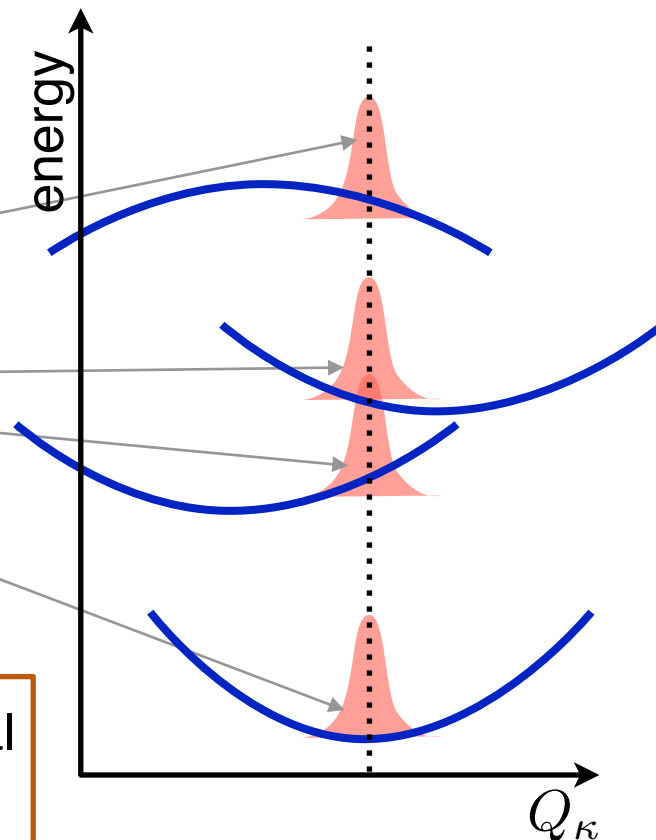


Hamiltonian: second order expansion of all relevant PES:

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Initial wavefunction:

$$\Psi(\vec{Q}, 0) = \sum_j \sqrt{p_j} \left[\prod_{\kappa=1}^f \chi_{\kappa}^{(j)}(Q_{\kappa}, 0) \right] |j\rangle$$



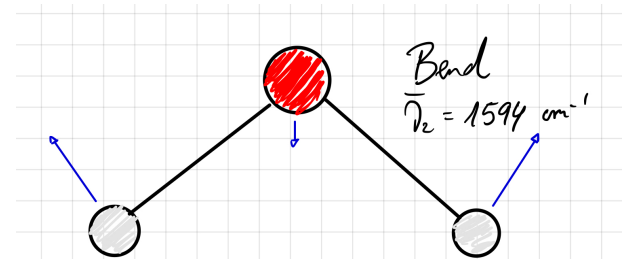
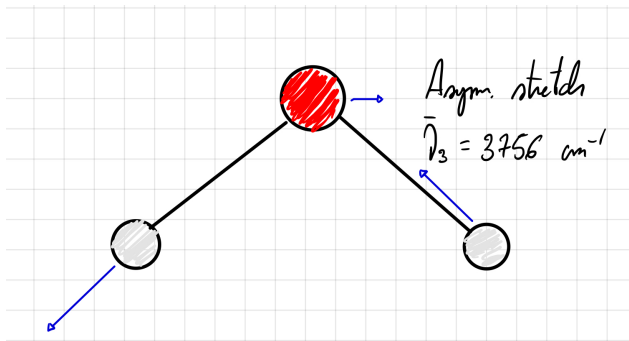
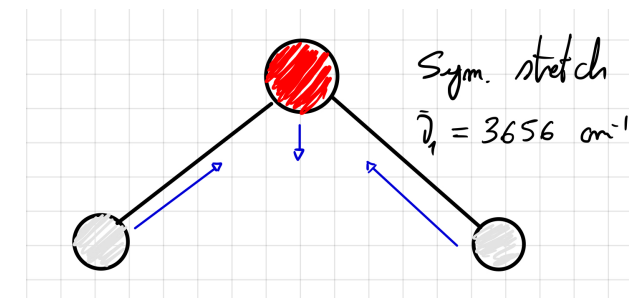
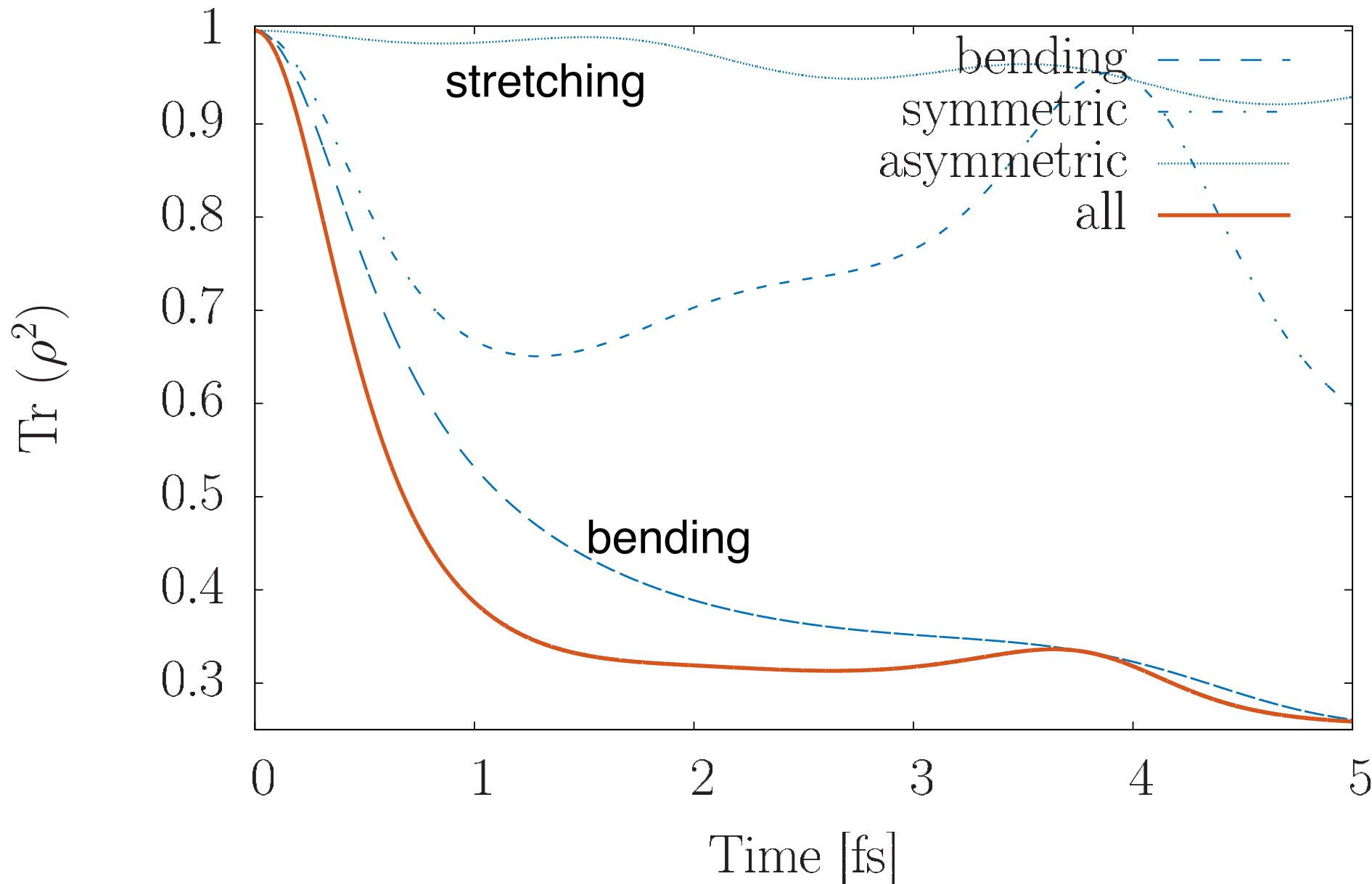
Electronic decoherence following photoionization: Full quantum-dynamical treatment of the influence of nuclear motion
C Arnold, OV, R Santra, *PRA* **95**, 033425 (2017)

Electronic decoherence in water cation



- 4 lowest cationic states equally populated

$$\Psi(\mathbf{Q}, 0) = \chi_1(Q_1, 0)\chi_2(Q_2, 0)\chi_3(Q_3, 0)\frac{1}{\sqrt{4}}(|1\rangle + |2\rangle + |3\rangle + |4\rangle)$$

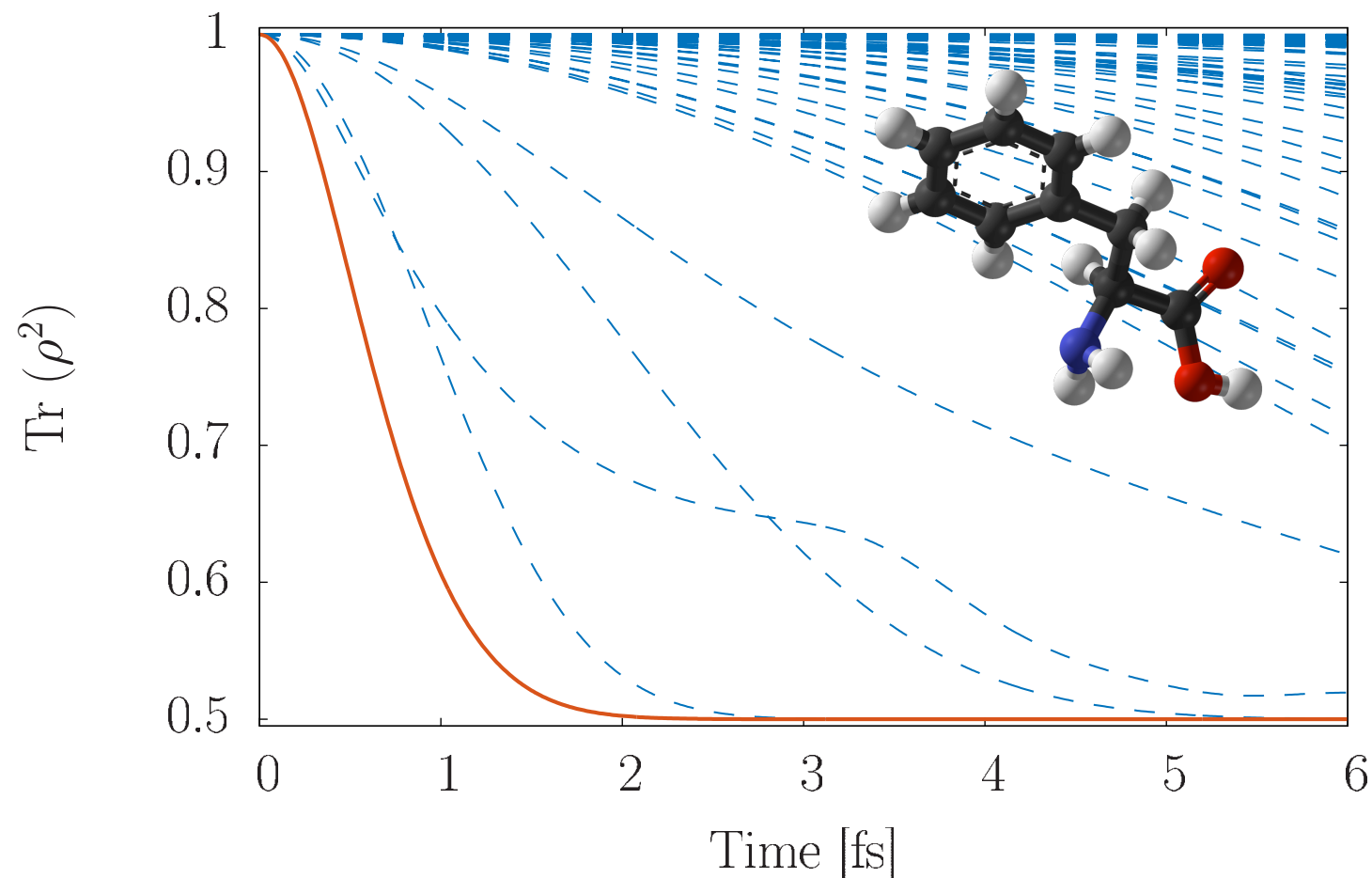


Electronic decoherence following photoionization: Full quantum-dynamical treatment of the influence of nuclear motion
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Electronic decoherence in *phenilananine* cation



- 63 D molecular system
- Linear superposition of X and A cationic states



- Decoherence is not due to the fast modes (C-H)
- Skeletal low frequency modes mostly responsible for electronic decoherence

Electronic decoherence following photoionization: Full quantum-dynamical treatment of the influence of nuclear motion
C Arnold, OV, R Santra, *PRA* **95**, 033425 (2017)



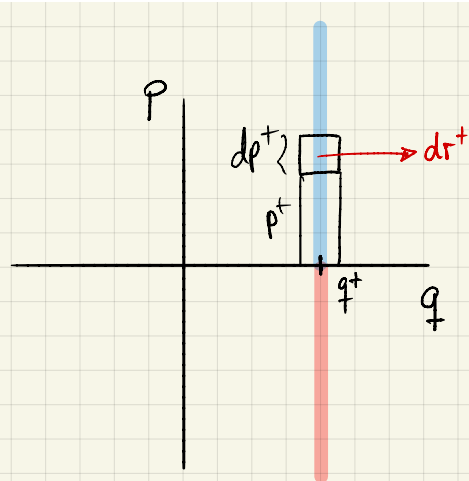
Textbooks:

- Tannor DJ, *Introduction to Quantum Mechanics, A Time-Dependent Perspective*
University Science Books **2007**
- Levine RD, *Molecular Reaction Dynamics*
Cambridge University Press **2005**

Reviews:

- *Molecular Structural Dynamics Probed by Ultrafast X-Ray Absorption Spectroscopy*
Bressler, Chergui, *Annu. Rev. Phys. Chem.* **61**, 263 (2010)
- *Roadmap of ultrafast x-ray atomic and molecular physics*
Young, Ueda et al., *J. Phys. B: At. Mol. Opt. Phys.* **51**, 032003, (2018)
- *The ultrafast X-ray spectroscopic revolution in chemical dynamics.*
Kraus, Zürich, Cushing, Neumark, Leone, *Nat. Rev. Chem.*, **2**, 82, (2018)
- Theme issue ‘*Measurement of ultrafast electronic and structural dynamics with X-rays*’ compiled and edited by Jon Marangos
Philos. Trans. Royal Soc. A **377** (2019)

Microcanonical rate derivation:



①
Our goal is to compute the differential of rate passing through the little box in phase space.

The rate is the density of states per unit length along the reaction coordinate $\times$ the velocity v^+ at the crossing point.

From normalization of a phase space volume $N = \frac{\Delta q \cdot \Delta p}{h}$.

Hence, the number of states in a box of sides

Δq around q^+ and dp^+ is $dN^+ = \frac{\Delta q dp^+}{h}$.

Finally, the differential density is $\rho^+ = \frac{dN^+}{d\Delta q} = \frac{dp^+}{h}$.

The transition state has energy $E - \epsilon_0$, where ϵ_0 is the classical barrier and E the total energy.

②
The energy of the activated complex is divided into translational energy ϵ_T and internal energy on all other DOF, ϵ_I .

$$E - \epsilon_0 = \epsilon_T + \epsilon_I$$

$$dr^+ = \left[\rho_I^+(\epsilon_I) \right] \cdot \left[\rho^+(\epsilon_T) \right] \cdot N^+(\epsilon_T) \\ = \rho_I^+(\epsilon_I) \cdot N^+(\epsilon_T) \frac{dp^+}{h} \rightarrow d\epsilon_T$$

$$dr^+ = \rho_I^+(\epsilon_I) \frac{d\epsilon_T}{h}$$

$$\left[r^+ \right] = \int_0^{E - \epsilon_0} \rho_I^+(\epsilon_I) \frac{d\epsilon_T}{h} = \int_{E - \epsilon_0}^0 \rho_I^+(\epsilon_I) \frac{-d\epsilon_I}{h} \\ = \int_0^{E - \epsilon_0} \frac{1}{h} \rho_I^+(\epsilon_I) d\epsilon_I = \frac{1}{h} N^+(E - \epsilon_0)$$

* State density per unit energy along perp. dir. to trans. state.

* State density per unit length along react. coord.

$$\frac{d\epsilon_T}{dp} = \frac{d}{dp} \left(\frac{p^2}{2m} \right) = \frac{p}{m} = v$$

Calculation of the canonical rate expression



③ We have determined the total microcanonical rate. The rate constant corresponds to the total rate divided by the number of states of reactants at energy $E \rightarrow P(E)$

$$r^+(E) = K(E) P(E)$$

$$K(E) = \frac{N^+(E - \epsilon_0)}{P(E)}; \quad P(E) = \frac{dN(E)}{dE}$$

- The thermal rate constant is now given by a Boltzmann average

$$K(T) = \frac{1}{Q(T)} \int_{\epsilon_0}^{\infty} K(E) e^{-\beta E} dE$$

$$= \frac{k_B T}{h} \frac{Q^+(T)}{Q(T)} e^{-\beta \epsilon_0}$$

Absorption spectrum for a very short pulse (1)



Short (δ) pulse absorption spectrum

①

Start from $\Delta E(\omega) = 2\pi i\omega [\tilde{P}(\omega)]^* \tilde{\Sigma}(\omega)$, the energy gain at frequency ω .

$$\text{For } \Sigma(t) = \delta(t) \xrightarrow{\text{IFT}} \tilde{\Sigma}(\omega) = \frac{1}{2\pi} \int_{-\infty}^{\infty} e^{i\omega t} \delta(t) dt = \frac{1}{2\pi}$$

The TD polarization in first order PT reads

$$P(t) = 2 \operatorname{Re} \left\{ \langle \psi^{(0)}(t) | \hat{\mu} | \psi^{(1)}(t) \rangle \right\} \\ = 2 \operatorname{Re} \left\{ \frac{1}{it} e^{iE_0 t/t} \langle \psi^{(0)}(0) | \hat{\mu} e^{-i\hat{H}_0 t/t} \hat{\mu} | \psi^{(0)}(0) \rangle e^{i\omega t} \right\} \theta(t)$$

where $\theta(t)$ is the Heaviside function since $P(t)$ is causal and therefore different than 0 only after the pulse.

Absorption spectrum for a very short pulse (2)



Now we introduce $\bar{P}(t) = \begin{cases} P(t) & \text{for } t > 0 \\ P(-t)^* & \text{for } t < 0 \end{cases}$ such that $\bar{P}(t) = P(-t)^*$

The IFT can now be taken on $\bar{P}(t)$ without needing to take the real part times 2.

$$\tilde{P}(\omega) = \frac{1}{2\pi i \hbar} \int_{-\infty}^{\infty} e^{i E_0 t / \hbar} \underbrace{\langle \psi^{(\omega)}(0) | \hat{\mu} e^{-i \hat{H}_0 t / \hbar} \hat{\mu} | \psi^{(0)}(0) \rangle}_{C(t)} e^{i \omega t} dt$$

Finally $\Delta E(\omega) = \frac{\omega}{2\pi \hbar} \int_{-\infty}^{\infty} C(t) e^{i(\omega_0 + \omega)t} dt$

and $\nabla(\omega) = \frac{\Delta E(\omega)}{|\tilde{E}(\omega)|^2 \cdot c} = \frac{2\pi \omega}{\hbar \cdot c} \int_{-\infty}^{\infty} C(t) e^{i(\omega_0 + \omega)t} dt$